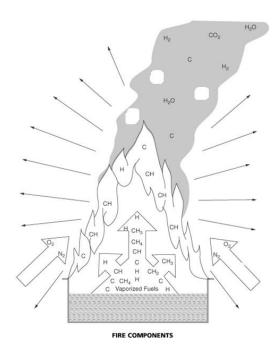
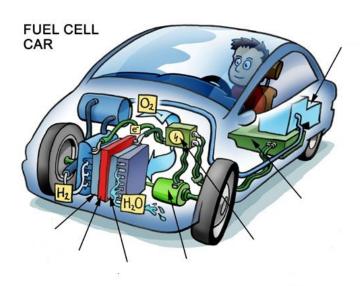
1. Find out?



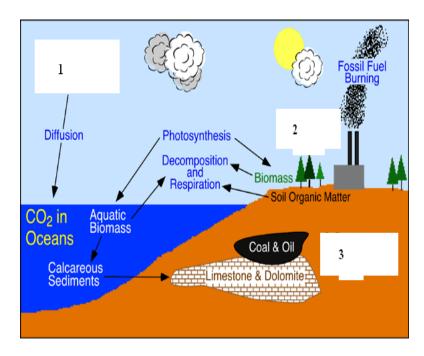
Answer: Air, CO₂, CO.

2.



Answer: Anode, Membrane, Cathode, Electro engine, Power electronics, Batteries, Hydrogen tank.

3. Choose the correct answer



1 - a. CO₂ Lithosphere

2 - b. CO₂ Atmosphere

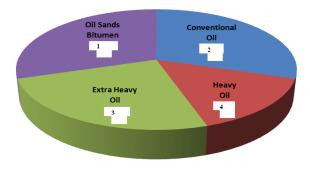
3 - c. CO₂ Biomass

Answer: 1-b, 2-c, 3-a.

4. Find out in order?

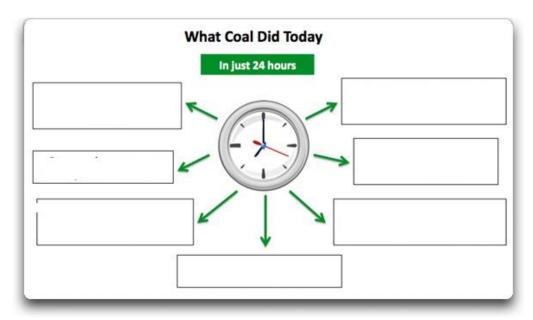
A) 30%, 15%, 25%, 30% B) 30%, 30%, 25%, 15% C) 15%, 25%, 30%, 30%

Total World Oil Reserves

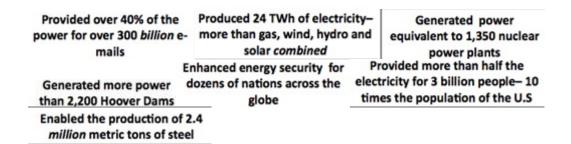


Answer: B

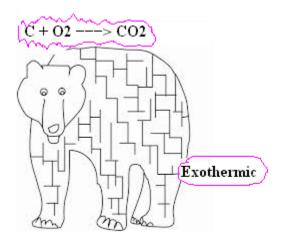
5.



Answer:

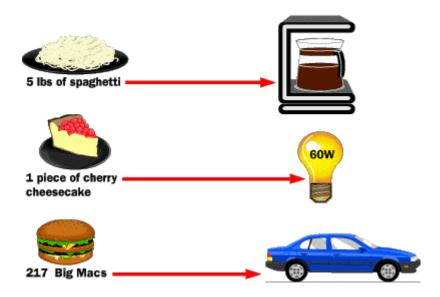


6. Find the way



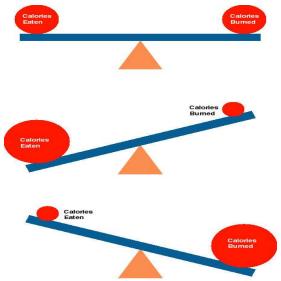
7.

The Calories in these items could:



Answer: Brew a pot of coffee, Light a bulb for 1.5 hours, Drive a car 88 miles.

8. Choose the correct



9. Weight gain, balance weight, weight loss, 2. Weight loss, balance weight, weight gain, 3. Balance weight, weight gain, weight loss

Answer: 3

Fuel

Q	F	M	Q	I	G	C	L	L	Z	В	D	R	C	V
Н	K	F	G	V	J	A	S	E	V	G	E	О	G	N
K	C	P	L	G	В	R	P	U	E	Z	M	Н	A	W
Н	M	J	G	Z	Z	В	E	F	C	В	W	I	K	V
S	A	D	L	P	M	O	Q	P	U	O	K	J	В	M
F	Q	Н	X	W	E	N	K	S	Q	V	E	F	E	J
I	T	Q	A	O	U	D	T	S	L	L	U	C	K	I
N	A	В	A	M	Y	I	E	E	P	X	N	О	K	Z
Z	E	J	S	D	О	O	E	I	R	О	L	A	C	T
V	N	G	O	N	F	X	M	О	K	A	F	Z	O	M
В	F	Z	Y	Z	L	I	I	V	A	Y	K	Н	В	V
D	K	A	C	X	Y	D	G	W	W	D	G	G	U	J
X	U	W	E	P	O	E	J	J	G	W	O	G	J	О
R	O	X	T	Q	F	R	M	L	A	T	X	R	В	Q
R	K	Z	V	O	Е	X	L	R	K	L	Н	W	Y	Z

CALORIE CARBONDIOXIDE COMBUSTION FUEL OXYGEN

11. Match the following

potassium hydroxide – a. CO
 alkaline pyrogallol – b. CO₂
 ammoniacalcuprouschloride – c. O₂

Answer: 1. b, 2. C, 3. A

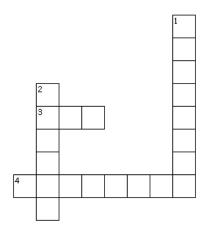
12. The calorific value of a fuel can be calculated if the percentages of the constituent elements are known

Substrate	Calorific value
?	8080
Hydrogen	?
?	2240

Answer: Carbon, 34500 and 2240

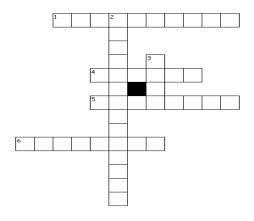
13.

Proximate Analysis



- 1. Moisture
- 2. Volatile
- 3. **Ash**
- 4. Carbon

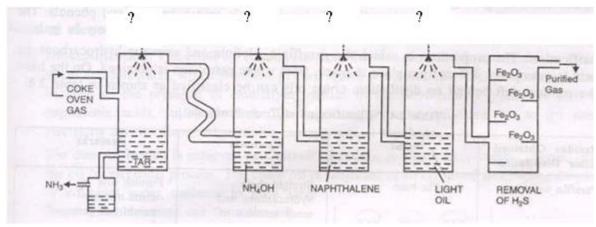
Metallurgical coke



Answer:

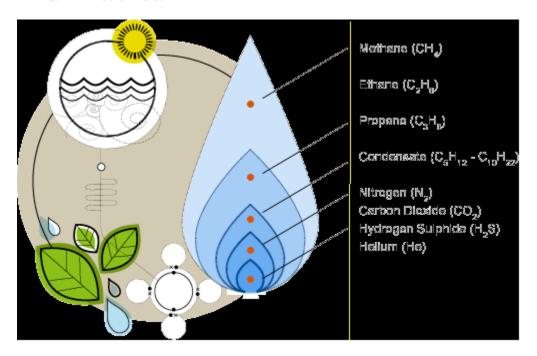
- 1. Porosity
- 2. Strength
- 3. Size
- 4. Combustibility
- 5. Reactivity

15.



Answer: 1. Liqour Ammonia 2. Water 3. Water at low temperature 4. Wash oil

16. True or False



Answer: True

17. What are enzymes involved in Manufacture of Power Alcohol

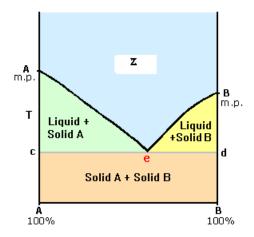
$${}^{C_{12}H_{22}O_{11}+H_{2}O \to C_{5}H_{12}O_{6}+C_{6}H_{12}O_{6}} \\ {}^{C_{6}H_{12}O_{6} \to 2C_{2}H_{5}OH + 2CO_{2}}$$

Answer: 1. Invertase, 2. Zymase

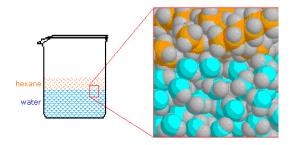
18. It is an allotropic form of carbon and a super conductor identify !!!!!



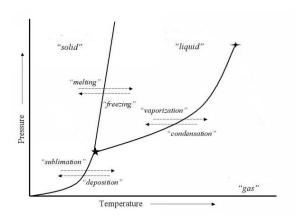
19. Locate the liquid phase and eutectic point in the phase diagram



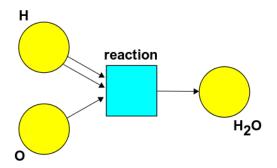
20. Find the polar solvent and identify the phase marked below.



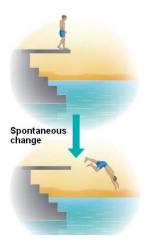
21. Locate the critical point, and triple point in the phase daigram.



22. Whether the reaction reversible or irreversible?



23. What happens to the free energy change when the diver jumps off the platform?



24. Name the polymers

used in making the following materials

1. PARACHUTE



2. COOKWARE



8. DISPOSABLE DIAPERS



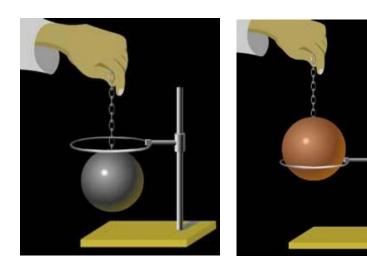
4. TAKE OUT CONTAINERS



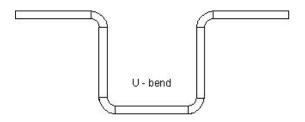




24. Why a ball when heated doesn't pass through the metal string.



25. Water and steam pipes often have a U-bend in them to allow _____



26. Identify RUL detector









27. The concrete cracking in the following figure is due to _____

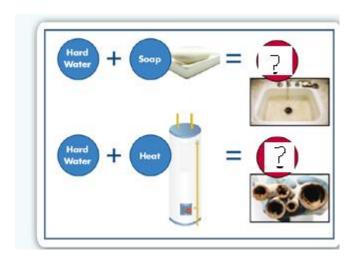


Porosity
$$(P) = \frac{?}{W - S}$$

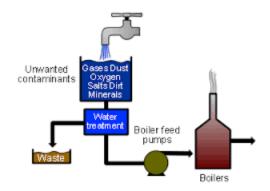
29. Name the following abrasive.



30. Find out



31. Find out input and output

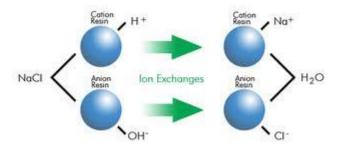


32. Match the following

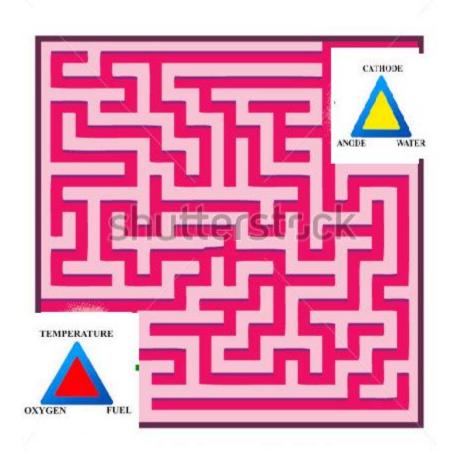


- 1. Water free from cations
- 2. Acid resin
- 3. Basic resin

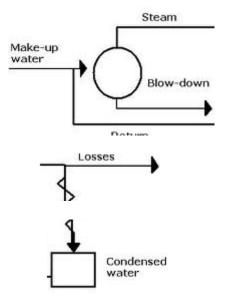
33. True or False



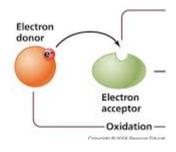
34. Find the way



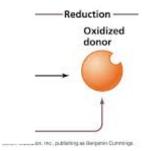
35. Match out



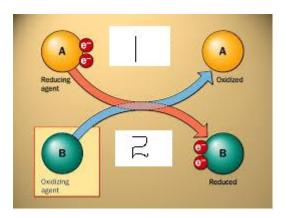
36. Arrange



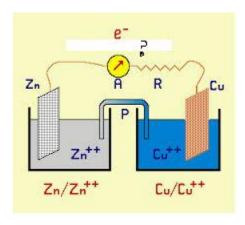




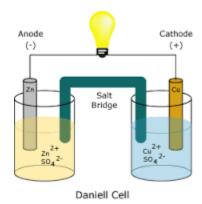
37. Find out loses of electron and gains electron



38. Draw the arrow mark for flow of electron



39. Bulb blow or not



40. Word search

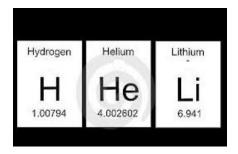
Н	Н	V	S	X	D	M	S	J	S	A	X	S	T	N	Y	Y	L
Y	I	I	X	I	U	O	E	T	V	C	X	F	N	S	K	O	G
X	F	J	U	I	L	L	R	M	M	I	Y	Q	E	J	A	Z	C
Z	J	Q	N	I	U	U	E	Н	Q	D	V	K	M	M	E	E	Н
X	I	A	D	C	C	E	C	E	S	A	В	О	I	S	О	M	E
L	R	U	E	T	L	S	N	N	Y	X	Y	Н	R	A	D	R	M
U	X	L	U	E	X	G	A	C	E	X	Q	О	E	G	W	N	I
U	О	R	M	M	E	J	T	M	R	G	Z	W	P	Q	V	E	S
M	E	E	N	T	О	J	S	Н	О	Y	О	S	X	Н	О	N	T
U	N	Е	J	I	R	G	В	D	Н	T	S	R	E	Н	Z	E	J
T	N	О	В	R	A	C	U	P	Q	M	A	T	D	K	T	R	K
О	X	Y	G	E	N	S	S	L	X	M	C	W	A	Y	Н	G	M
D	N	U	O	P	M	O	C	S	Ο	D	I	U	M	L	Н	Y	R

ACID
CARBON
CRYSTAL
EXPERIMENT
LIQUID
SODIUM
SUBSTANCE

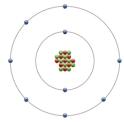
ATOM
CHEMIST
ELEMENT
GAS
MOLECULE
SOLID
URANIUM

BASE
COMPOUND
ENERGY
HYDROGEN
OXYGEN
STRUCTURE

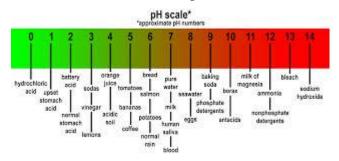
41. What is the first element on the periodic table?



42. What is the centre of an atom called?



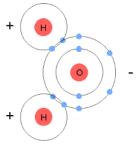
43. True or false? Acids have a pH level below 7.



44. What is the main gas found in the air we breathe?



45. True or false? An electron carries a positive charge.



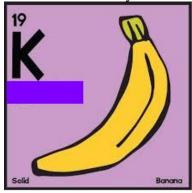
46. Famous New Zealand scientist Ernest Rutherford was awarded a Nobel Prize in which field?



47. What is the chemical symbol for gold?



48. K is the chemical symbol for which element?



49. What is it?



- 50. At room temperature, what is the only metal that is in liquid form?
- 51. True or false? A neutron has no net electric charge.
- 52. A nuclear reaction where the nucleus of an atom splits into smaller parts is known as nuclear fission or nuclear fusion?
- 53. What is H₂0 more commonly known as?



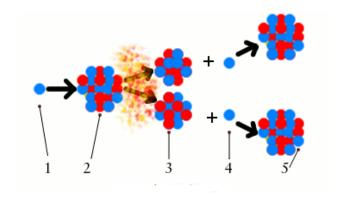
54. What is the third most common gas found in the air we breathe?



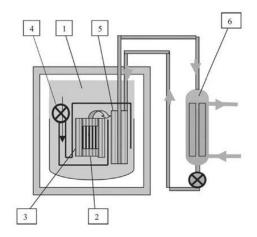
- 55. What is the name given to substances that are initially involved in a chemical reaction?
- 56. True or false? Bases have a pH level below 7.
- 57. Is sodium hydroxide (NaOH) an acid or base?
- 58. Atoms of the same chemical element that have different atomic mass are known as?
- 59. True or false? A proton carries a positive charge.
- 60. What is the fourth most abundant element in the universe in terms of mass?

41. Hydrogen	42. A nucleus	43. True	44. Nitrogen (around 78%)
45. False	46. Chemistry	47. Au	48. Potassium
49. Corrosion	50. Mercury	51. True	52. Nuclear fission
53. Water	54. Argon (around 1%)	55. Reactants	56. False
57. Base	58. Isotopes	59. True	60. Carbon

61. Find out



62. Find out



63. Name the label

