



SNS COLLEGE OF TECHNOLOGY

Coimbatore-35 An Autonomous Institution

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DEPARTMENT OF FOOD TECHNOLOGY

UNIT 1 - CONDUCTION

Topic 3: 1D SS Conduction equation for wall

Conduction Heat transfer 1) I objective is to determine the temp field. (In) in a body (how temp varies with post within the body) -) Objectives

2) Ten) depends on boundary condition, initial, ma en'al ppts (8, k, (p and geometry) 3) Why we need Ten)

3) Why we need Tin)

a) compute heat flow at any point b) compute thermal strent, enpantion, deflection

due to temp e) Design polts in applications such as insula theren, Heat treatment of metals.

Bounday & Initial Condition.

1) Heat equation is second order on spatial coordinates, hence à boundary cond'néeded for each copedinate

The initial conditions describe the temperature distribution in a medium at the initial moment of time and there are needed only for time dependent pblons.

t>0, T=7(2,4x8)

one dimensional steady state heat conduction.

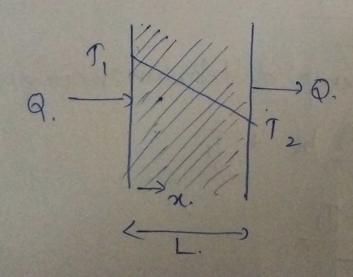
without heat generation

- Lonsider a plane wall of a material of uniform thermal conductivity, k, which is a summed to be extending to infinity in y and z directions.

walls of room may be considered as a plane, if the energy lox through the edger is negligible

$$\frac{\partial^2 T}{\partial x^2} + \frac{\partial^2 T}{\partial y^2} + \frac{\partial^2 T}{\partial z^2} + \frac{9}{1c} = \frac{1}{2c} \frac{\partial T}{\partial t}.$$

$$\frac{\partial^2 T}{\partial n^2} \pm \frac{\partial^2 T}{\partial y^2} = 0$$
 (one dinensional)



Conduction equation simplifies to $\frac{\partial^2 T}{\partial n^2} = 0 \quad \text{on} \quad \frac{d^2 T}{dn^2} = 0 \quad - \quad \boxed{0}$ This is a 2nd order differential eq requining 2 BC $T=T_1$ at x=0T=T2 at N2L entegrating 1 twice we get T = 9x+ G Where G and G can be determined from bounda n=0, T=T, so that C=T, n=L, T=T2 & that T2=GL+T, 9=72-1 so the eq" for temp distribution becomes T=(T2-T1) n + T, $\frac{d\Gamma}{dx} = \frac{T_2 - T_1}{L}$

$$Q = -KA \frac{dT}{dr} = -KA \left(T_2 - T_1\right) = KA \left(T_1 - T_2\right)$$

The thermal gly of heat supplied to the left face of wall to manistain a temperature difference T, -T, across it.

$$R_{th} = \frac{L}{KA}$$

$$Q = -KA \frac{\partial T}{\partial x}$$

$$\int Q dx = -KA \int^{2} dT$$

$$Q = -KA \left(T_{2}-T_{1}\right)$$

$$Q = -\frac{KA(T_2-T_1)}{L} = \frac{KA(T_1-T_2)}{L} = \frac{2}{L}$$

Plane wall with convective boundary conditions

The differential eqn is
$$\frac{d}{dn}(k \cdot \frac{dr}{dn}) = 0 - 0$$

Bounday conditions are, a) n=0, $T=T_{S_1}$, $T(s) = T_{S_1}$ — ② b) n=L, $T=T_{S_2}$ $T(L) = C_1 L + T_{S_1}$ -: C1 = T(L) - Ts1 -: T(x)=(T(L))-Ts)x+Ts1-4 Heat flux acron the wall is given by $Q = -KA \frac{d\tau}{dx} = \frac{KA}{L} \left(T_{S_1} - T_{S_2}\right) = \frac{T_{S_1} - T_{S_2}}{(H_{KA})} - 3$ Thermal Resistance concept for convection 9=hA(To -Tsi) = , 9 = To, -Ti) Ina Applying thermal tenstame concept for plane we with convertion boundary conditions. Tool Tri JUF 1 T12
HF F002

Tolon Til To2 9 1/h, A 9 4/14 9 1/h, A

teat transfer late may be determined by conviduing each element of the resistance network as.

$$2 = T_{\infty_1} - T_{S1} = T_{S_1} - T_{S2} = T_{S_2} - T_{\infty_2} - T_{\infty_2}$$

$$\frac{1}{\ln_1 A} = \frac{1}{\ln_2 A}$$

since revistances are in seues.

$$R_{total} = \sum_{i=1}^{r} f_{i} = \int_{n_{i}A}^{r} f_{i} + \int_{n_{i}A}^$$





1) Heat transfer deals with the rate of

- a. work transfer
- **b.** temperature transfer
- c. energy transfer
- d. none of the above

Answer:

c. energy transfer





2. The amount of heat required to raise the temperature of a substance by 1°C is called:

A. work capacity

B. heat capacity

C. Energy capacity

D. none of the above

Ans: B



MCQ



- 3 Heat bring.....change
- A. Physical
- B. chemical
- C. reversible
- D. periodic

Ans: B Heat bring chemical change





- 4. The process of transfer of heat in liquids & gases is called:
- A. Conduction
- B. Radiation
- C. Convection
- D. Absorption

Ans: C It is the process of transfer of heat in liquids & gases



MCQ

5) Solids are not heated by convection because:

solid are not free to move from one place to another

A. molecules only vibrate about a fixed position

B. both A and B

C. none of the above

Ans: C Solids are not heated by convection because the molecules of a solid are not free to move from one place to another; they can only vibrate about a fixed position