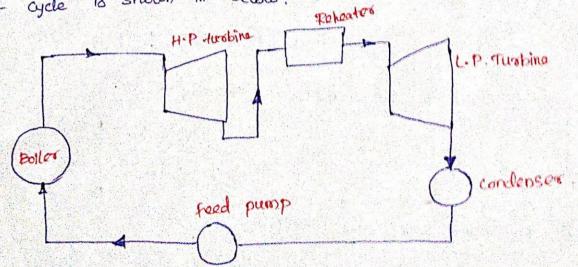




RENEAT RANKINE CYCLE :-

is desimable to increase the average temperature end Supplied to 14 and the steam at which the heat is pressure of the steam as dry as possible at the end of the also to keep pressure increases, the ecopansibo matie in the othe -turbine. TA be increased and the steam becomes wet at will also -turbine the Steam will the end of expansion. Increasing the moisture of the expsion of the tembine blades and also increase the losses.

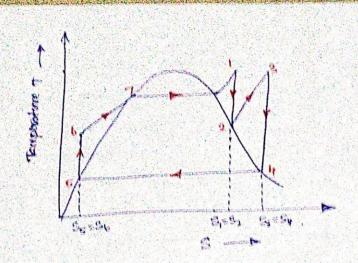
In reheat cycle, the expansion is being carried out in two stages. The steam is initially expanded in the dustaine on some pressure and then it is reheated with the help of flue gases in the boiler. Then the steam is expanded in L.P. turbine to the condenser pressure. The main purpose of reheating is to increase the daynoss fraction of the steam pressing through turbine should never fall below o.B. The thermal efficiency is increased with reheat cycle and also the specific steam consumption is decreased. But, the thermal efficiency of the school cycle may be decreased, if it is used at low pressures. The diagram for reheat cycle is shown in below.







(14.4



The process 1-2 represent itenshopic expansion in high a process turbine, and 3-4 represent isenshopic expansion in low-proseure sturbine. The steam is reheated at constant pressure 2-3. The reheat can be carried out by returning the steam to the boiler and passing it through a heat exchange placed in the boiler at constant prossure.

Heat supplied,
$$\Theta_s = \Theta_{s-\epsilon} - \Theta_{s-s}$$

$$= (h_1 - h_0) + (h_2 - h_0)$$

Work output W= W1-2+W9-11-Wp.

-: The efficiency of the reheat rankina cycle in

-: WA = YA, (P,-P4)

gt the pump work is neglected

where, $h_1 = antealpy$ of Steam Ott Intermediata pressure P_2 .



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Note:

this stage may be super heated, wet & day, stoom at "The comparing SI & S2. it is found duy

has tenthalpy of superheated steam at prosource P3=B2.

by => enthalpy of steam at pressure P4 1.2. condenses pressure he > enthalpy of steam at pump outlet.

$$h_6 = h_5 + hlp$$

Problem: -

power plant operating on an ideal reheat Rankine cycle. The Steam onters the H.P turbine at 30 bors (1) Consider 350°c. After expansion to 5 bors, the steam is relicated to 350°c then expanded the L.P turbine to the condenser pressure 0.075 bas. Determine the thermal efficiency of the cycle and the quality of the steam at the outlet of the L.P turbine

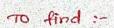
Given :-

P = 30 bar

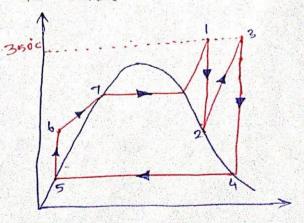
T, = 350°C

P2 = 5 bar.

P3 = 0.075 bay



- 1) Thermal efficiency of cycle.
- 11) Quality of the steam of the outles of . . C.p turbine







Solution :-

(1a)

i) shormal efficiency of cycle:

$$\eta_{RE} = \frac{(h_1 - h_2) + (h_3 - h_4) - Wp}{h_1 - (h_5 + Wp) + (h_5 - h_2)}$$

no find by

To find he

Find the steam condition with the help of S, & S, \$3,532.

at 5 bar. 8 = 80 = 6.819 KI/kg.

.: h2 = hf2 + 22 hfg2.

$$S_1 = S_1 = S_{f_2} + x_2 S_{fg_2}$$
.

@ 5 bar, hez = 640.1 ks/kg.

$$h_2 = 640.1 + (0.98)(2107.4)$$

To find he

The stage 3 is in Superheated Condition.





90 find by At the end of stage 4, the steam will line in wex strain endlinen his = his + as hear. on hind ay S, = S.4. Sa = Sh = Shi + 24 Ston At P = 0.075 bar. Sfor = 0.576 KJ/Hy.k., hfg = 168.65 KJ/Hg. Sfgy = 7.677 KJ/Hg.K , hpgy = 2406.2 KJ/kg Su = Sfy + 2c4 Sfgy . 7.634 = 0.576+ X4 (7.677) x4 = 0.919. his = heat + sea haga. = 168.65 + 0.919 (2406.2) hy = 2379. 94 45/19. hs=ha= 168.65 KJ/kg. WP = Vf4 (P1 - P4) @ P4 = 0.075. Wp = 0.0010075 (30 - 0.075) RE = (h,-h2)+(h3-h4)-Wp. h, - (hga+ Wp) + (hg-h2) = (3117.5 - 2705.35)+(3168.1 - 2379.94) - 3.0149 31175 - (168-65 + 3.6149) + (3168.1 - 2705.35) = 412.15 +788.16 - 3.0149 3117.5 - 171.6649 + 462.75





Cui

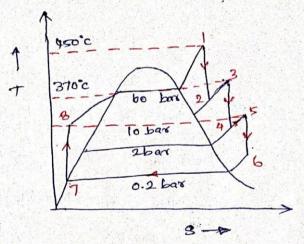
$$D_{RE} = \frac{1197.29}{2483.08} = 0.48.$$

Result)-

- i) Thermal efficiency of the cycle Mer = 48%.
- ii) Quality at the Steam at L.P. turbine 24 = 0.919.
- (2) In the reheat cycle the steam is at 6MN/m² and 450°c. The first reheat is done at 1MN/m² to 370°c. The second reheat is done at 0-2 MN/m² to 320°c. The exhaust pressure is 0-02 MN/m². Determine the thermal efficiency and power developed at a steam safe of 1 kg/s.

Given :-

$$P_1 = 6 \text{ MN/m}^2 = 60 \text{ bar}$$
 $P_2 = 1 \text{ MN/m}^2 = 10 \text{ bar}$
 $P_4 = 0.2 \text{ MN/m}^2 = 2 \text{ bar}$
 $P_6 = 0.02 \text{ MN/m}^2 = 0.2 \text{ bar}$
 $T_1 = 450^{\circ}\text{C}$
 $T_2 = 370^{\circ}\text{C}$
 $T_3 = 320^{\circ}\text{C}$
 $T_{10} = 1 \text{ kg/s}$



To find :-

i) Thermal efficiency (1)

ii) Power developed (P)

Solution :-

At 60 bars and 450°C.

h, = 3301.15 KJ/kg.

S; = 6.714 KJ/kg·k.

To find he

find the State of Steam.

S1 = S2 = 6.714 KJ/kg.





N book, Sy, = 6.583 Ko/Ly (15) 36 S2 > S02 (The Steam is Super Footed) \$ 205'c By wing mollier chart. h2 = 2830 kg/kg. era divid ha :-At 10 box , 73 = 370'c . hg = 3200.86 KU/kg. S3 = 7.3686 KS/kg.k To find ha find the state of steam. Sa = Sy =7.8686 K3/kg.k. At 2 box . Sq = 7.127 KS/kg K. Sy > Sy (The steam is in superhoused condition) Ty = 170°C (By using mollier chart) h = 2800 kJ/kg. To find he At 2 bor, To = 320°c. h= 3112.78 KJ/kg S= 7.962 KJ/kg·K To find he Find the state of Steam. S5 = S6 = 7.962 K3/kg.k At 0.2 bar, Sg6 = 7-909 K5/kg.K. S6> Sg6 (The steam state is superheated) To = 61°c (By using mollier chart) hb = 2600 KJ/kg. hig = 251.5 KJ/kg.





$$P = \frac{(h_1 - h_2) + (h_3 - h_4) + (h_5 - h_6)}{(h_1 - h_{46}) + (h_3 - h_4) + (h_5 - h_4)}$$

$$= \frac{(3301 \cdot 5 - 2838 \cdot 42) + (3200 \cdot 86 - 2809 \cdot 3) + (8112 \cdot 08 - 2613 \cdot 72)}{(3301 \cdot 5 - 2151 \cdot 5) + (3200 \cdot 86 - 2804 \cdot 42) + (8112 \cdot 08 - 2809 \cdot 3)}$$

$$P = \frac{(h_1 - h_2) + (h_3 - h_4) + (h_5 - h_6)}{(3301 \cdot 5 - 2838 \cdot 42) + (3200 \cdot 86 - 2809 \cdot 3) + (3112 \cdot 08 - 2613 \cdot 72)}$$

$$P = \frac{(3301 \cdot 5 - 2838 \cdot 42) + (3200 \cdot 86 - 2809 \cdot 3) + (3112 \cdot 08 - 2613 \cdot 72)}{(3112 \cdot 08 - 2613 \cdot 72)}$$



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$$P = m \left((h_1 - h_2) + (h_3 - h_4) + (h_5 - h_6) \right)$$

$$= 1 \left[(3301.5 - 2838.42) + (3200.86 - 2809.3) + (3112.08 - 2613.72) \right]$$

$$P = 1353 \text{ kW}$$

REGENERATIVE RANKINE CYCLE :-

is less than carnot cycle efficiency is less than carnot cycle for the same temperature times because heating of feed water taken places with a large temperature difference. If the temperature of feed water is raised to the Soturation temperature corresponding to the boiler pressure before it enters into the boilers, the cycle efficiency will be as close as to carnot cycle efficiency.

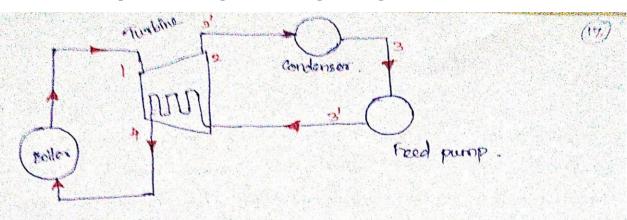
The working fluid bearing the feed pump eixculates around the tuxbine casing where the heat transfer takes places between the incoming steam and the fluid in the liquid state inside the casing.



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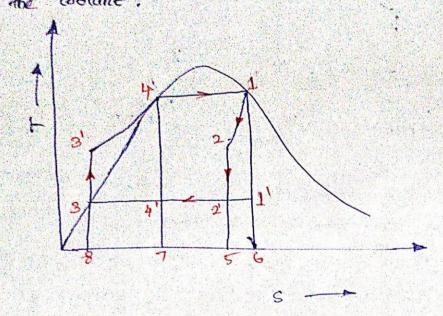


No - heat loss by the steam in the turbine

sy - heat gained by the working fluid flows through the .

throbine caving.

4-1 - heat transfer from the boiler to the working fluid, i's' - amount of heat transferred from the working fluid to the coolant.



of the assea under the curve 1-2 and 3-4 are equal.



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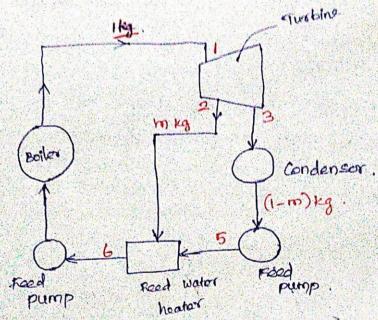


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b) Trackful Regenerative Rankine cycle is

1) Single Stage Regenerative Rusking cycle : - x

The above cycle is cloudy not a practical proposition durbine operaties with wet steam of low dryness, Sho because which will affect the turbine blades severely trowever -Fraction efficiency can be improved upon in practice the rankine cycle by bleeding off some of the stoam at an intermediate pressure during expansion and It is used to hoat the feed water in Separate feed water heater.



Steam expands from condition 1 through the turbine. At the pressure corresponding to point 2, a quantity of Eteam m kg per kg of Steam Supplied from boiler is bled off with heating feed water. The rest of Steam (1-m) to Completes for the expansion and is exhausted to the Conclenser pressure Ps. amount of Steam is then Condensed and pumped to the Same pressure as the bleed steam. The bleed steam of in the and head water are mixed in the feed water under





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Ideal adiabatic conditions, the state of the Condensed mass in hig and the feed water (1-m) kg leaving the heater will be same and is represented by the State 7. The food water is other pumped by the second food pump to the boiler processine where it is heated to State 1 and cycle is repeated.

work done by the turbine per kg of steam

pump work,

$$W_{p} = (1-m)(h_{5}-h_{4}) + (h_{7}-h_{6})$$

$$= (1-m) \times V_{1}(P_{5}-P_{4}) + V_{6}(P_{7}-P_{6})$$

$$W_{p} = (h-m) V_{f_{3}}(P_{2}-P_{3}) + V_{f_{2}}(P_{1}-P_{6})$$

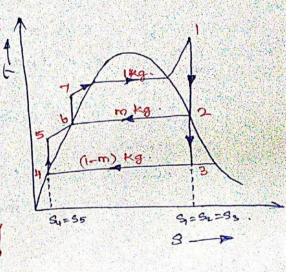
Net work output,

Heat Supplied in the boiler,

Heat rejected by the conditionser,

$$Q_{R} = (1-m)(h_{8}-h_{4})^{2}$$

$$Q_{R} = (1-m)(h_{8}-h_{4})^{2}$$



Thermal efficiency,
$$\frac{W}{Q_3} = \frac{Q_5 - Q_R}{Q_S}.$$

$$\frac{1}{\sqrt{h_1 - h_2} - (h_1 - h_2) - (h_3 - h_{5})}}{\sqrt{h_1 - h_2}}$$

The amount of steam extracted (m) is determined from the heat balance in the feed water heater.



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Hart hast by the steam = Heat gained by the water.

$$m \times (h_0 - h_0) = (1 - m)(h_0 + h_0)$$
 $mh_0 = mh_0 = h_0 - h_0 - mh_0 + mh_0$
 $mh_0 = mh_0 = h_0 + h_0$
 $m(h_0 - h_0) = h_0 - h_0$
 $m = \frac{h_0 - h_0}{h_0 - h_0}$
 $m = \frac{h_0 - h_0}{h_0 - h_0}$
 $m = \frac{h_0 - h_0}{h_0 - h_0}$

A) Two Stage Regenerative Routine cycle:-

. If a regenerative cycle employs two feed water heaters,

ie Steam is bleed from two place of the turbine

Work done per ty of steam supplied to the tembine.

$$\lambda_{1} = (h_{1} - h_{2}) + (i - m_{1})(h_{2} - h_{3}) + (i - m_{2})(h_{2} - h_{4})$$

turbline. $W_{T} = (h_{1} - h_{2}) + (i - m_{1})(h_{2} - h_{3}) + T$ (1-m,-m2)(h3-h4)

Pump work, Wp = Wp + Wp + Wp3

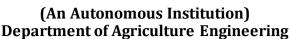
Wp = (1-m1-m2) (P6-P5) + (1-m1) (h8-h-1)+ (h10-ha)

We = (1-m,-m2) Vey (P3-P4) + (1-m) V43 (P2-P3) + V82 (P,-P2)

Heat Supplied by the boiles per leg of Steam generated. Qs = (h,-h10)

Thermal efficiency, $p = \frac{W_1 - W_p}{Q_0} = \frac{W_1 - W_p}{h_1 - h_{10}}$







The amount of steam bleed from turbine in two stages m, and m2 can be determined from energy belance equation

For first heater,

$$m_1h_2 + (i-m_i)h_2 = h_q$$
.
 $m_ih_2 + h_3 - m_1h_3 = h_q$.
 $m_1 (h_2 - h_2) = h_q - h_3$.
 $m_1 = \frac{h_q - h_3}{h_2 - h_2}$.

: ha = hea.

$$m_1 = \frac{h_{f2} - h_{f3}}{h_2 - h_{f3}}$$

for second heater,

$$m_2h_3 + (i-m_1-m_2)h_6 = (i-m_1)h_7$$

 $m_2h_3 + h_6 - h_6m_1 - h_6m_2 = h_7 - m_1h_7$
 $m_2h_3 - m_2h_6 + h_6 = h_7 - m_1h_7 - m_1h_6$
 $m_2(h_3-h_6) + h_6 = h_7 - m_1(h_7-h_6)$
 $m_2 = \frac{h_7 - h_6 - m_1(h_7 - h_6)}{h_3 - h_6}$

$$m_2 = \frac{(i-m_1)(h_7-h_4)}{h_3-h_6}$$





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Problems :-



Deplied to the turbine at 40 bar and 1150°c and the condenser prossure is 6.03 bar. After expansion in the turbine to 3 bar. Some of the Steam 15 extracted from the turbine for heating the feed water from the Condenser in an open heater. The pressure in the boiler is 40 bar and the state of the fluid leaving the heater is Saturated liquid water at 3 bar. Assuming isentropic heat drop, in the turbine of pump.

Given!-

 $P_1 = 40 \text{ bar}$ $T_1 = 450^{\circ} \text{ c}$ $P_2 = 3 \text{ bar}$ $P_3 = 0.03 \text{ bar}$

To find :-

Efficiency of the cycle (Prog)

Solution :-

The State 1 is Superheated Steam

P = 40 bar T = 450'c.

h, = 3330.35 KI/kg.

S, = 6.9363 KJ/kg.k.

The State 2 is in wet steam

S1 = S2 (isentropic process)

S1= S2 = 6.9363 KS/ 19.4.

At 3 bar,

Sfa = 1.672 KJ/kg.K. hfa = 561.5 KJ/kg.

Sfg. = 5.319 K5/kg.k. hfg. = 2163.2 KJ/kg.

Si = Sf2 + X2 Sfg2 .

6-9363 = 1-672 + 32 (5-319)

Tx2 = 0.98



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$$\begin{array}{c} h_{0} = h_{0} + m_{0} h_{1} g_{3} \\ = 86h_{1} + m_{0} h_{1} g_{3} \\ = 86h_{1} + m_{0} h_{2} g_{3} \\ \\ h_{1} = a6h_{1} + m_{0} h_{2} g_{3} \\ \\ h_{2} = a6h_{1} + m_{0} h_{2} h_{2} \\ \\ h_{3} = 6.03 \text{ hav} \\ \\ h_{3} = 8.00 \text{ has } h_{3} h_{2} h_{3} \\ \\ h_{3} = 8.00 \text{ has } h_{3} h_{2} \\ \\ h_{3} = 8.00 \text{ has } h_{3} h_{3} \\ \\ h_{3} = 24h_{1} \cdot h_{2} h_{3} \\ \\ h_{3} = 24h_{1} \cdot h_{2} h_{3} \\ \\ h_{3} = 24h_{1} \cdot h_{2} h_{3} \\ \\ h_{3} = 6.0 \\ \\ h_{3} = 10.0 + (6.8)(24444 \cdot h_{3}) \\ \\ h_{3} = 2056 \cdot 68 \text{ has } h_{3} \\ \\ h_{3} = 10.0 + (6.8)(24444 \cdot h_{3}) \\ \\ h_{3} = 2056 \cdot 68 \text{ has } h_{3} \\ \\ h_{3} = 10.0 \cdot h_{3} \\ \\ h_{5} = 10.0 \cdot h_{5} \\ \\ h_{5} = 10.0 \cdot$$



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Arrount of Stuars bleed,

$$m = \frac{h_{10} - h_{20}}{h_{3} - h_{50}}$$
 $= \frac{160.15}{2620.64}$
 $= \frac{160.15}{160.15}$
 $= \frac{160.$



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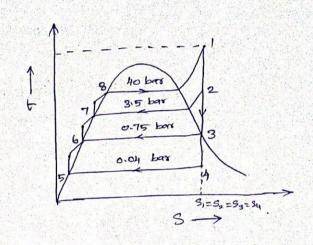
Problem on Two Stage Regenerative Cycle:-



(i). In a regenerative Steam cycle employing two open feed water heater, the Steam 18 Supplied to the turbine at 30 bor and 500°C and is exhausted to the condensor pressure or of the metraction points for two heaters are at 3.5 bor and 0.75 bor respectively. Calculate the thermal efficiency of the plant.

Given :-

$$P_1 = 30 \text{ bar}$$
 $T_1 = 500^{\circ} \text{ C}$
 $P_2 = 3.5 \text{ bar}$
 $P_3 = 0.75 \text{ bar}$
 $P_4 = 0.04 \text{ bar}$



To find :-

Thermal efficiency (PREG)

Solution :-

$$W_{T} = (h_{1}-h_{2}) + (1-m_{1})(h_{2}-h_{3}) + (1-m_{1}-m_{2})(h_{3}-h_{4})$$

To find by

The Steam is Superheated in state (1).

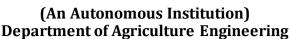
To find he.

Find the State of Steam.

At 3.5 bor, Sg, = 6.939 KJ/kg.k

The steam is superheated







from the mollier chart, through the point P = 3.5 bos , S = 7.235 KJ/g.k. ho = 2850 KS/kg. To find he, Pa = 0.75 bar. Find the state of steam. S, = 53 = 7.235 KJ/19.K At P3 = 0.75 box Sgg = 7.457 KJ/kg.K, Sfg= 1.218 KJ/kg.K, Sig= 6.243 KJ/kg.K $S_{9x} > S_3$. (Steam is Wet). hts = 384.3 kg/kg hay = 2278-6 15/10 Sa = Sf3+ 73 Sf93 . 7.235 = 1.213 + 23 (6.243) x3 = 0.96. ha = hfa + 9 hfga. = 384.8 + (0.96) (2278.6) h3 = 2581.75 K3/Hg. To find ha The steam is Wet condition hy = hf4 + 24 hfg, S1 = S4 = 7.235 KJ/kg.k. Str = 0.483 Kilmin his tolde killing. At 0.04 bar S4 = Sf4 + X4 Sfg4 .. Stan = 8.053 Mayk. high = \$133.4 15/17 7.235 = 0.423 + 74 (8.053) No = 0.001004 m/m 74 = 0.845 hen = hen + the hega = 121.4 + (0.845) (2433-1) h4 = 2419.54 KS/kg



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Pump work (5-6)

$$W_{p} = (1-m_{1}-m_{2}) \left(h_{6}-h_{5}\right)$$

$$W_{p} = (1-m_{1}-m_{2}) \left(V_{fh} \left(P_{3}-P_{4}\right)\right)$$

$$\left(k+m_{6}+m_{4}\right) h_{6}-h_{5} = V_{fh} \left(P_{3}-P_{4}\right)$$

$$h_{6}-h_{5} = 0.001004 \left(75-4\right)$$

$$h_{6}-h_{5} = 0.071284$$

$$h_{6}-121.4 = 0.071284$$

$$h_{6} = 121.53 \text{ KS/kg}$$
Pump work (7-8)

$$W_{p} = (1-m_{1}) V_{f_{3}}(P_{2}-P_{3})$$

$$h_{g}-h_{7} = V_{f_{3}}(P_{2}-P_{3})$$
At 0.75 bor.
$$V_{f_{3}} = 0.001037 m_{1}^{3}/kg.$$

$$h_{g}-h_{7} = 0.001037 (350-75)$$

$$h_{g}-h_{7} = 0.285$$

$$h_{g} = 0.285 + h_{7}$$

$$= 0.285 + h_{7}$$

$$= 0.285 + h_{7}$$

$$= 0.285 + 384.4$$

$$h_{g} = 384.68 \text{ KI fleg}$$

Wp = (1-m) + (hg-hy)

$$W_{p} = h_{10} - hq$$

$$W_{p} = V_{f_{1}} (P_{1} - P_{2})$$
At 20 bors
$$V_{f_{1}} = 0.001079 \text{ m}^{2}/kg$$

$$h_{10} - h_{9} = V_{f_{1}} (P_{1} - P_{2})$$

$$h_{10} - h_{9} = 0.001079 (3000 - 350)$$

$$h_{10} - h_{9} = 2.859$$

$$h_{60} = 2.859 + h_{9}$$

Pump work (9-10)





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$$h_{16} = 2.869 + h_{1/2} \qquad \text{:}h_{1} = h_{1/2}$$

$$= 2.869 + t98h \cdot 33 \qquad h_{1/2} = 58h \cdot 33 + t3/h_{1/2}$$

$$h_{10} = 587 \cdot 19 \quad k5/h_{1/2}$$

$$m_{1} = \frac{h_{1} - h_{1/2}}{h_{2} - h_{1/2}} = \frac{h_{1/2} - h_{1/2}}{h_{2} - h_{1/2}} = \frac{58h \cdot 33 - 28h \cdot 6}{2850 - 28h \cdot 6}$$

$$m_{1} = 0.081 k_{1/2}$$

$$m_{2} = \frac{(i - m_{1})(h_{2} - h_{1/2})}{h_{2} - h_{1/2}} = \frac{(i - m_{1})(h_{1/2} - h_{1/2})}{h_{3} - h_{1/2}}$$

$$= \frac{(i - m_{1})(h_{1/2} - h_{1/2})}{h_{3} - h_{1/2}} = \frac{(i - m_{1})(h_{1/2} - h_{1/2})}{h_{3} - h_{1/2}}$$

$$= \frac{(0.918)(502.8)}{24k0.22}$$

$$m_{2} = 0.098 k_{1/2}$$

$$= \frac{(0.918)(502.8)}{24k0.22}$$

$$m_{2} = 0.098 k_{1/2}$$

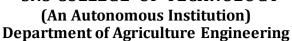
$$= \frac{(0.918)(502.8)}{24k0.22}$$

$$m_{2} = 0.098 k_{1/2}$$

$$= \frac{(0.918)(502.8)}{24k0.22}$$

$$= \frac{(0.$$







(2). A reheat cycle operating between 30 box and 0.04 box has a superheat and nehout temperature of 150°C. The first expansion take place till the Steam is dry saturated and then reheat is glinen. Neglecting feed pump work determine the ideal cycle efficiency.

Given :-

$$P_1 = 30 \text{ bar}$$
.
 $P_4 = 0.04 \text{ bar}$.
 $T_1 = 450^{\circ} \text{c}$.
 $T_3 = 450^{\circ} \text{c}$.
 $x_1 = 1$

To find: -

Thermal efficiency of cycle . (Pcycle) =
$$\frac{(h_1-h_2)+(h_3-h_4)}{(h_1-h_4)+(h_3-h_2)}$$

Solution :-

The State 1 is superheat steam.

At 0.04 bar,

$$h_{4} = 121.4 \text{ KJ/kg}$$
. $S_{4} = 0.423 \text{ KJ/kg·k}$. $h_{5} = 2433.1 \text{ KJ/kg}$. $S_{5} = 8.053 \text{ KJ/kg·k}$.

In day saturated steam condition.

$$S_2 = S_g$$
, \Rightarrow somplarity $h_2 = h_g$.

(from Steam table) P2 = 2.3 barr.

AF 2.3 bar.
$$h_2 = 2712.6 \text{ kS/kg}$$
.

At 2.3 bars and 450°C.

Process 3-4

$$S_3 = S_4 = 8.3061 \text{ KS/kg/k}$$

 $S_4 = S_{54} + 2_4 S_{594}$





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at .o. 01 bar
$$S_{R_1} = 0.423 \text{ kJ/kg·k}$$
. $h_{R_1} = 121.4 \text{ kJ/kg}$. $S_{R_3} = 8.052 \text{ kJ/kg·k}$, $h_{R_3} = 44.93.1 \text{ kS/kg}$. $8.2061 = 0.123 + 9c4 (6.052)$

$$\boxed{X_1 = 0.98}.$$

$$h_4 = h_{R_4} + X_4 \text{ hfgg}.$$

$$= 121.4 + (6.98 \times 24.93-1)$$

$$h_4 = 2505.84 \text{ kJ/kg}.$$

$$H_6 = cycle \quad \text{efficiency} \quad P = \frac{(h_1 - h_2) + (h_3 - h_4) - WP}{(h_1 - h_2) + (h_3 - h_4)}$$

$$H_7 = \frac{(h_1 - h_2) + (h_3 - h_4)}{(h_1 - h_2) + (h_3 - h_4)}$$

$$H_7 = \frac{(33.14.35 - 2712.6) + (32.81.46 - 27505.84)}{(33.44.35 - 121.4) + (32.81.46 - 2712.6)}$$

$$P = 6.3873.$$

$$P = 38.73 \text{ V.}$$



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Problem .-

UNIT - 4

In a Robbine cycle, the steam at lolet to turbine is saturated at a prossure of 20 bar superheated at 300°C. The exhaust pressure is over born and expansion take place isentropically. Determine, i) pumpuak ii) Work turbine iii) Rankine esticiency iv) Work done v) specific steam consumption (Steam Rate)

Given ! -

$$P_1 = 20 \text{ bar}$$
.
 $T_1 = 300^{\circ}\text{C}$.
 $P_2 = 0.07 \text{ bar}$.

70 find :-

- i) pump work.
- ii) Turbine work.
- iii) Rankine efficiency
- iv) work done .
- v) specific Steam Consumption (SSC)

Solution :-

At
$$P_i = 20$$
 bay $T_i = 300^{\circ}C$.
 $h_i = 3025.0$ KJ/kg.
 $S_i = 6.770$ KJ/kg.K.

At . 0.07 bar.

$$S_1 = S_2 = 6.770 \text{ kS/kg·k}$$
 $S_1 = S_2 = 6.770 \text{ kS/kg·k}$
 $S_2 = \frac{0.559}{1.718} \text{ kJ/kg·k}$
 $S_3 = \frac{1.718}{1.718} \text{ kJ/kg·k}$
 $S_4 = \frac{1.718}{1.718} \text{ kJ/kg·k}$
 $S_5 = \frac{1.718}{1.718} \text{ kJ/kg·k}$
 $S_5 = \frac{1.718}{1.718} \text{ kJ/kg·k}$

$$g_2 = S_{f_2} + x_2 S_{fg_2}$$
.
 $6.770 = 0.559 + x_2 (7.718)$
 $x_2 = \frac{6.211}{7.718}$
 $x_2 = 0.80$
 $h_2 = h_{f_2} + x_2 h_{fg_2}$.
 $= 163.4 + (0.80)(0409.2)$

he = 2090.76 K3/kg.

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