



UNIT – 3

BIOPOTENTIAL ELECTRODES & CONFIGURATION

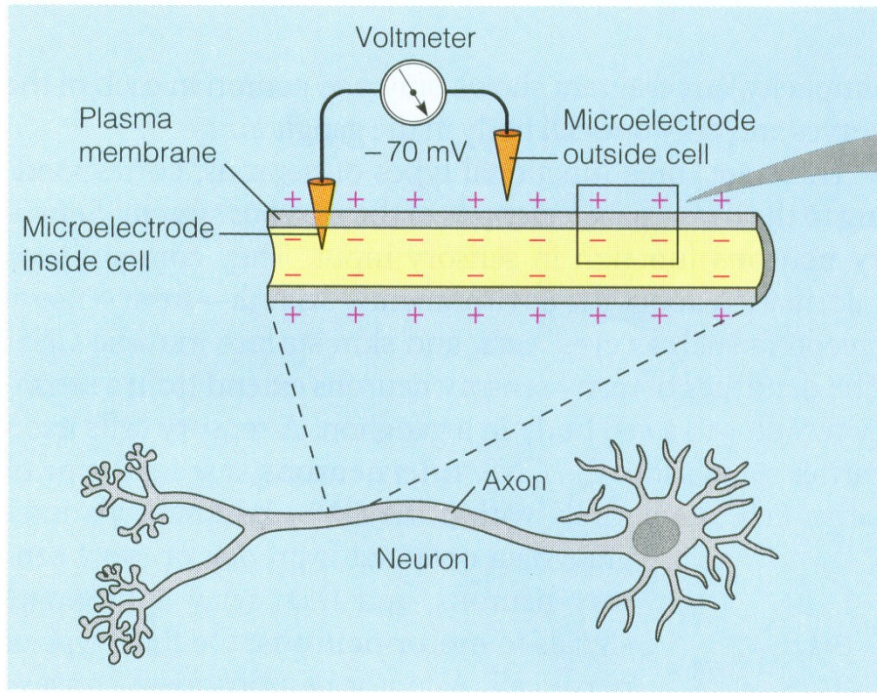
Goldman-Hodgkin-Katz Eq.

Bioelectricity

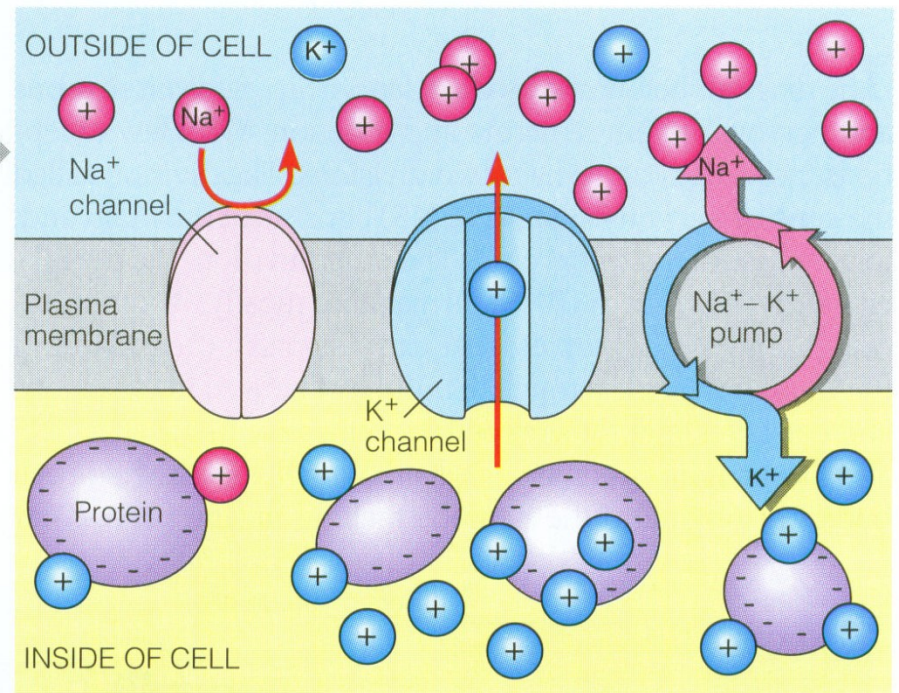
- Provides basis for “irritability” or “excitability”
- Fundamental property of all living cells
- Related to minute differences in the electrical potential across a cell

‘resting potential’

- Resting potential values can range from 20 to 100 mv, with inside of membrane **negative** to outside
- Any stimulus which evokes a response in a cell is associated with a change in this potential



A. Measuring a neuron's resting potential



B. How the resting potential is generated

Membrane Channels:

Leak Channels

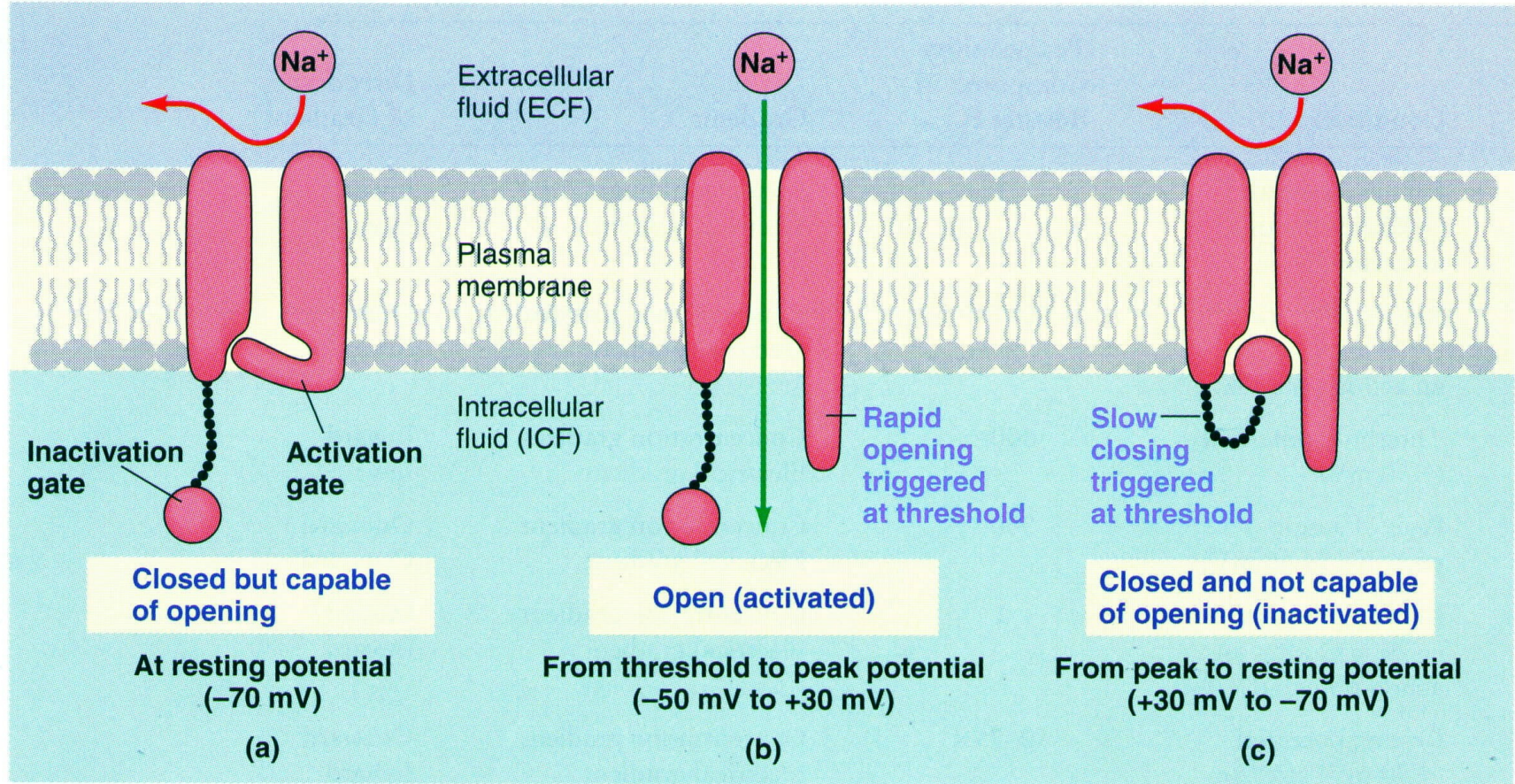
Gated Channels

Chemical (Ligand)

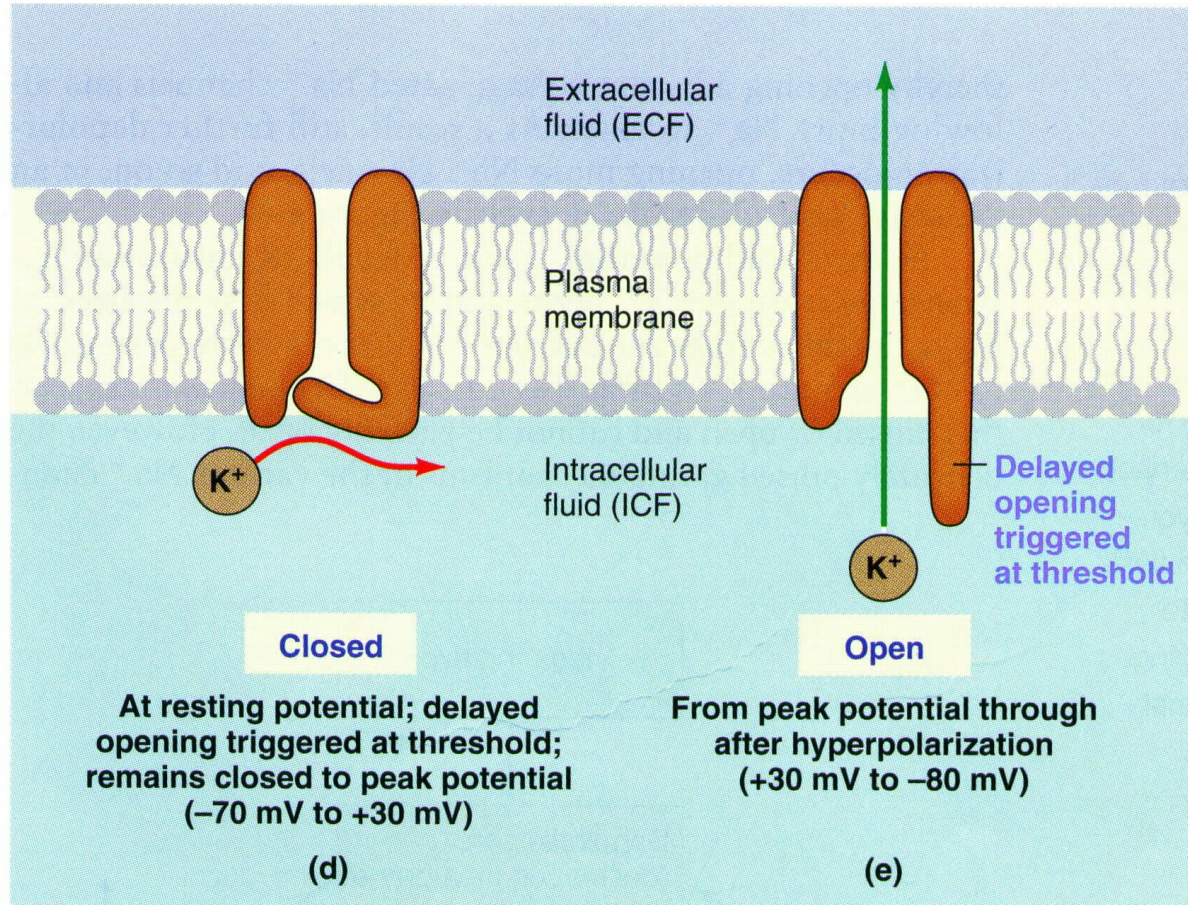
Voltage

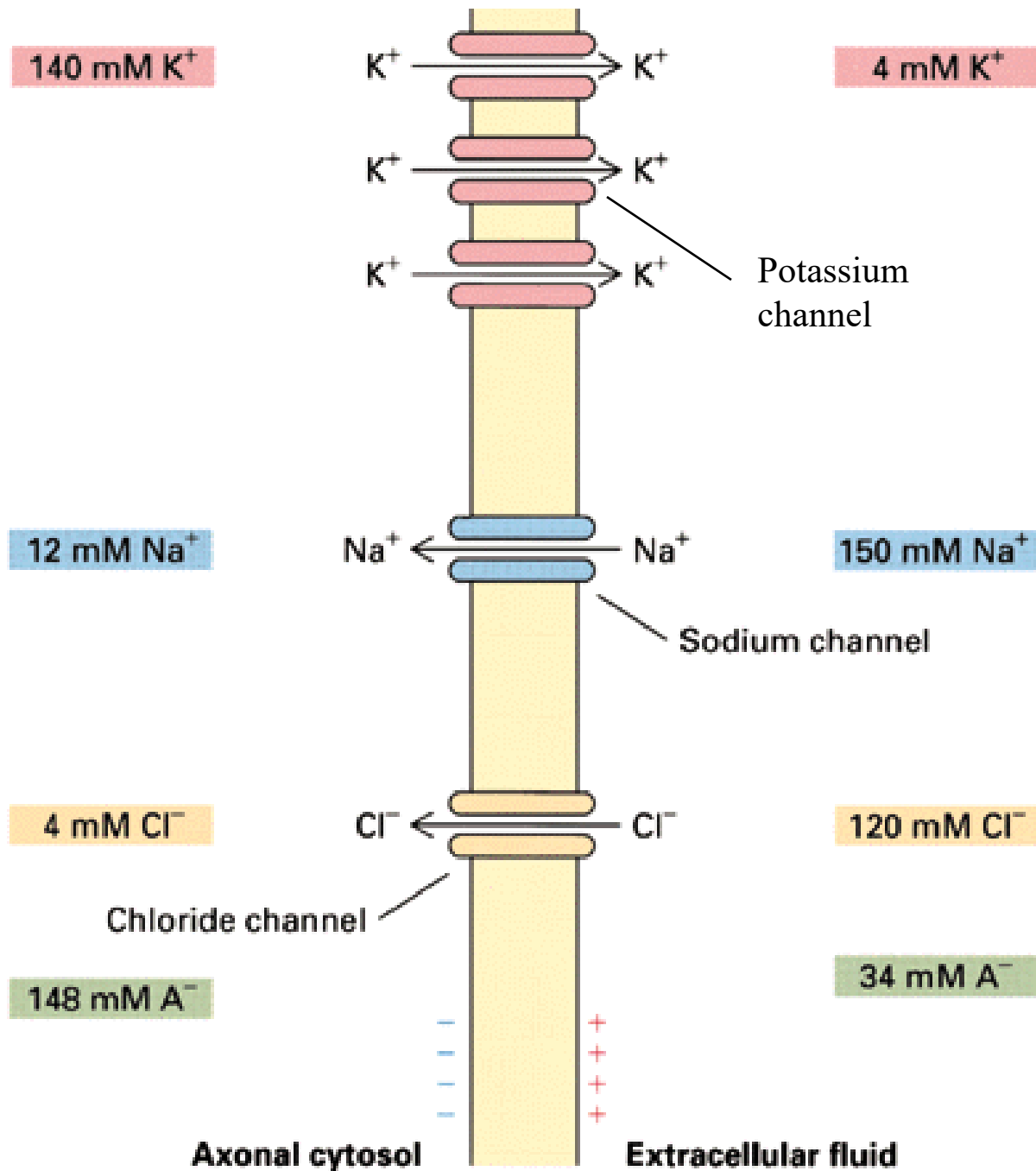
Mechanical

Voltage-Gated Sodium Channel



Voltage-Gated Potassium Channel





Look at squid giant axon:

$$V_m = -70 \text{ mV}$$

$$[\text{Pr}^-] = 415 \text{ mM}$$

$$[\text{Na}^+] = 50 \text{ mM}$$

$$[\text{K}^+] = 400 \text{ mM}$$

$$[\text{Cl}^-] = 35 \text{ mM}$$

$$[\text{Na}^+] = 440 \text{ mM}$$

$$[\text{K}^+] = 20 \text{ mM}$$

$$[\text{Cl}^-] = 560 \text{ mM}$$

$$E_{\text{Na}} = \frac{+25\text{mV}}{+1} \ln \frac{440}{50} = +54.4 \text{ mV}$$

$$E_{\text{K}} = -74.9 \text{ mV}$$

$$E_{\text{Cl}} = -69.3 \text{ mV}$$

What determines the value of the membrane voltage?

Goldman-Hodgkin-Katz Eq.

$$V_m = + 25 \text{ mV} \cdot \ln \frac{P_{\text{Na}} [\text{Na}^+]_o + P_{\text{K}} [\text{K}^+]_o + P_{\text{Cl}} [\text{Cl}^-]_i}{P_{\text{Na}} [\text{Na}^+]_i + P_{\text{K}} [\text{K}^+]_i + P_{\text{Cl}} [\text{Cl}^-]_o}$$

If the membrane were only permeable to Na:

$$V_m = + 25 \text{ mV} \cdot \ln \frac{\cancel{P}_{\text{Na}} [\text{Na}^+]_o}{\cancel{P}_{\text{Na}} [\text{Na}^+]_i} = E_{\text{Na}} = + 54.4 \text{ mV}$$

Goldman-Hodgkin-Katz Eq.

$$V_m = + 25 \text{ mV} \cdot \ln \frac{P_{\text{Na}} [\text{Na}^+]_o + P_{\text{K}} [\text{K}^+]_o + P_{\text{Cl}} [\text{Cl}^-]_i}{P_{\text{Na}} [\text{Na}^+]_i + P_{\text{K}} [\text{K}^+]_i + P_{\text{Cl}} [\text{Cl}^-]_o}$$

If the membrane were only permeable to K:

$$V_m = + 25 \text{ mV} \cdot \ln \frac{\cancel{P_{\text{K}}} [\text{K}^+]_o}{\cancel{P_{\text{K}}} [\text{K}^+]_i} = E_{\text{K}} = - 74.9 \text{ mV}$$

If the membrane were only permeable to Cl:

$$V_m = + 25 \text{ mV} \cdot \ln \frac{P_{\text{Cl}} [\text{Cl}^-]_i}{P_{\text{Cl}} [\text{Cl}^-]_o} = E_{\text{Cl}} = - 69.3 \text{ mV}$$

Thus, the GHK eq. is a sum of the Nernst eqs. weighted by permeability!

In the resting neuron, $P_K \& P_{\text{Cl}} \gg P_{\text{Na}}$

(~ 20-50 x)

Therefore, V_m is close to $E_K \& E_{\text{Cl}}$

with very little contribution from E_{Na}

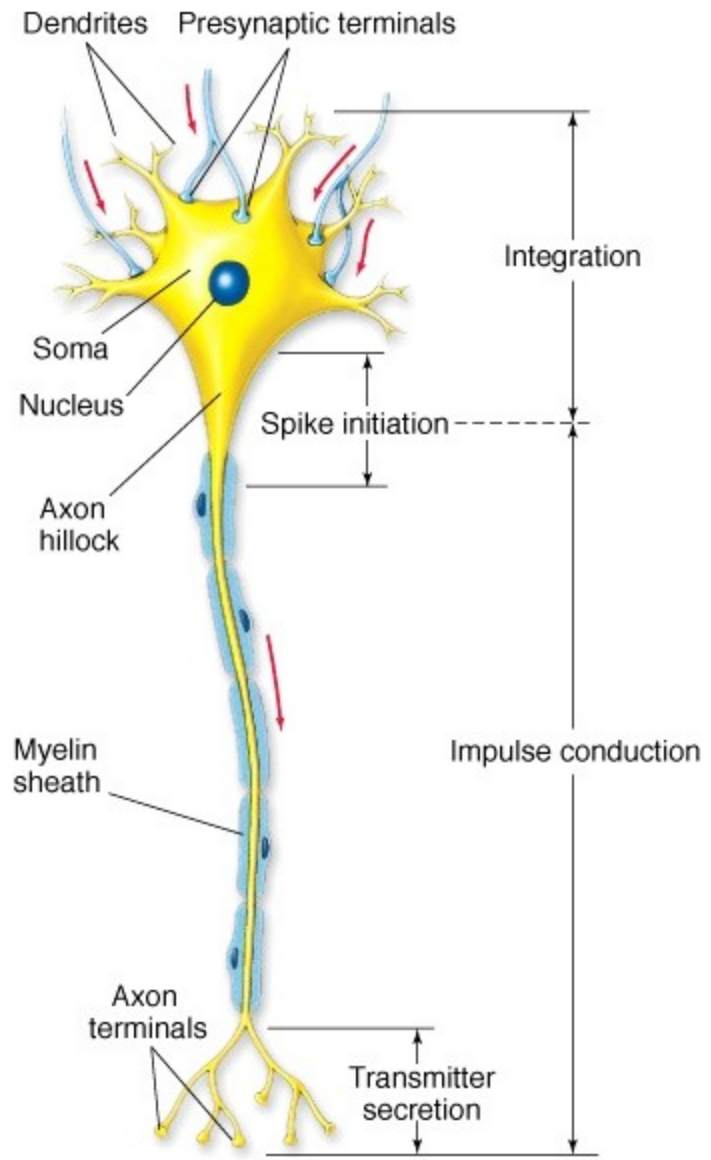
Neuronal Physiology

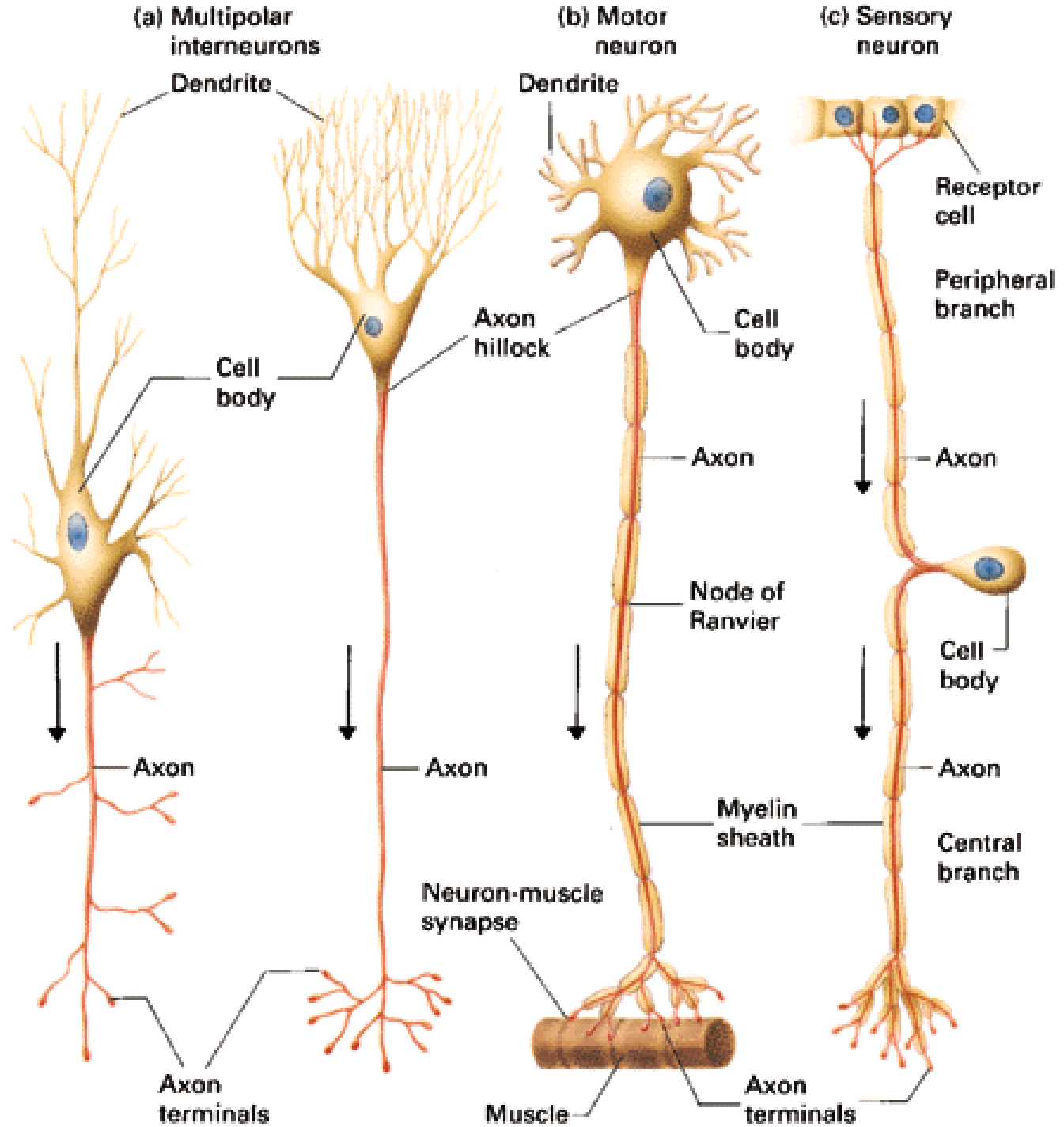
Neurons = nerve cells; fundamental unit of nervous system

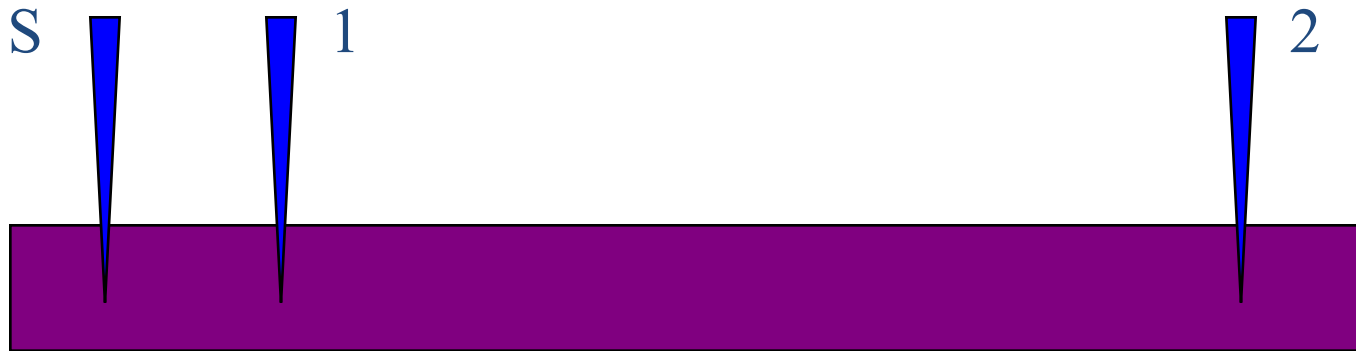
Surrounded by plasma membrane that possesses an electrical potential (resting potential = -70 mv)

Membrane potential due to uneven distribution of ions on either side of membrane

Nerve (and muscle) cells can make special use of this potential (i.e., excitable tissues) – use changes in the potential to create signals and hence transmit information or bring about contractions







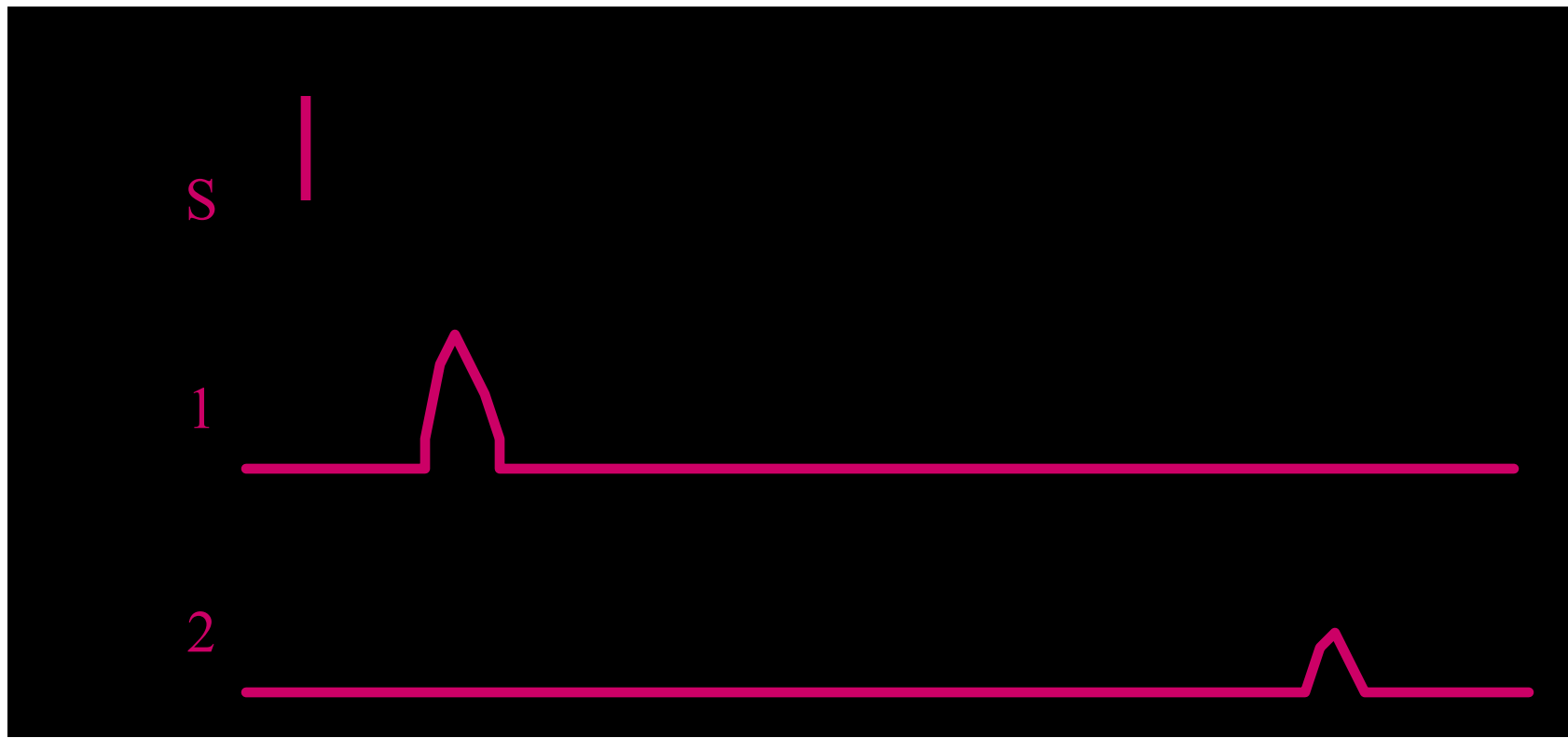
Graded Potential



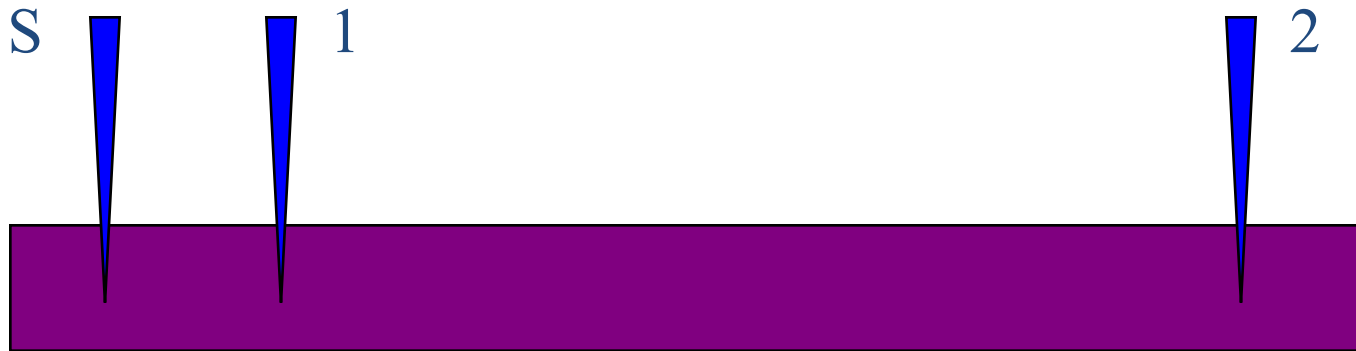
Time (ms)



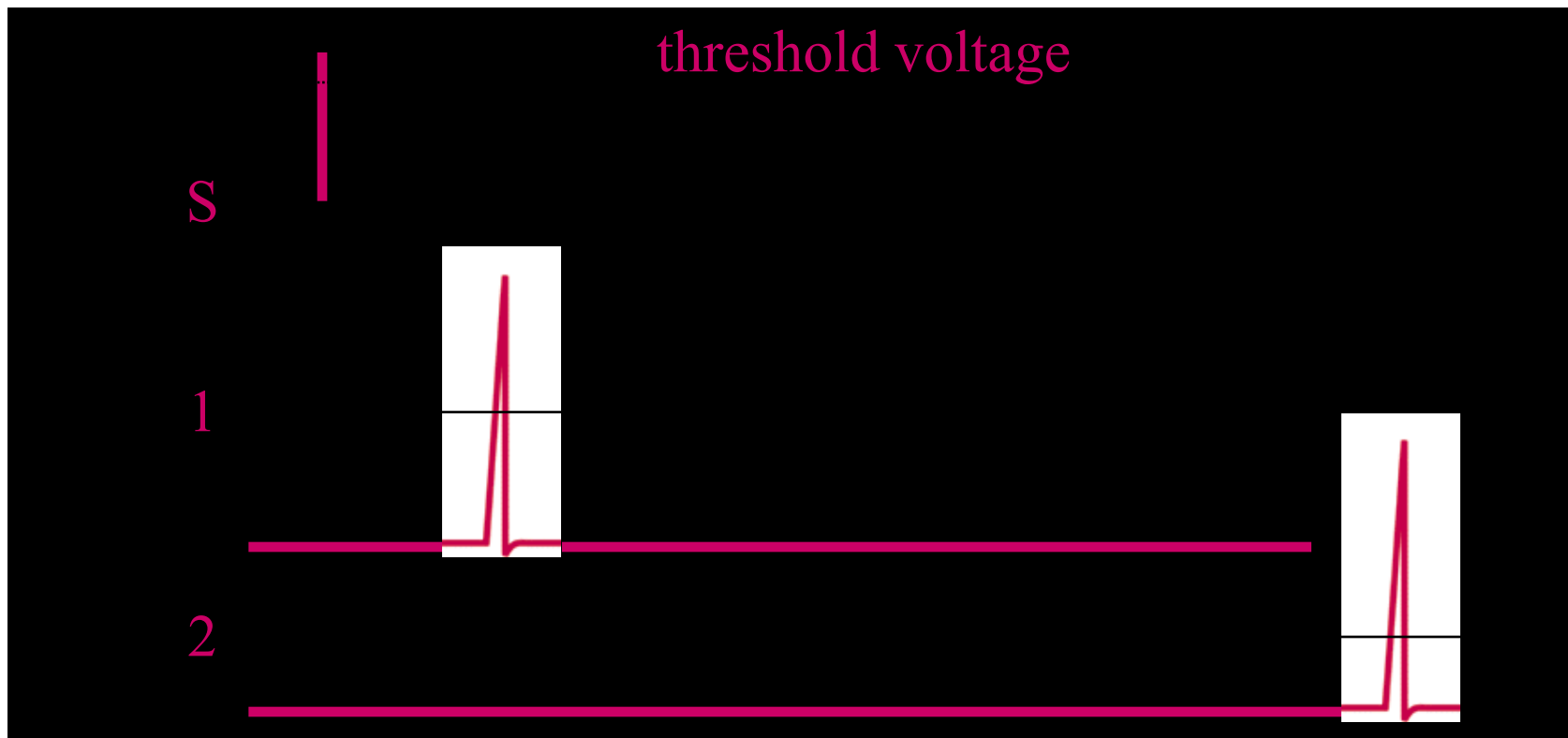
Graded Potential



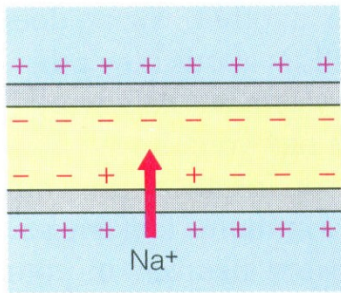
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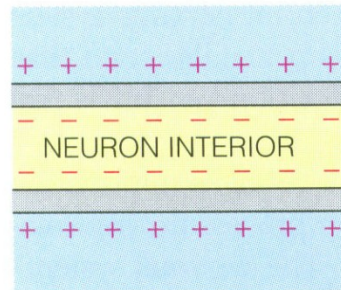
Action Potential



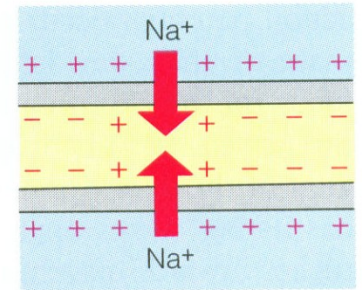
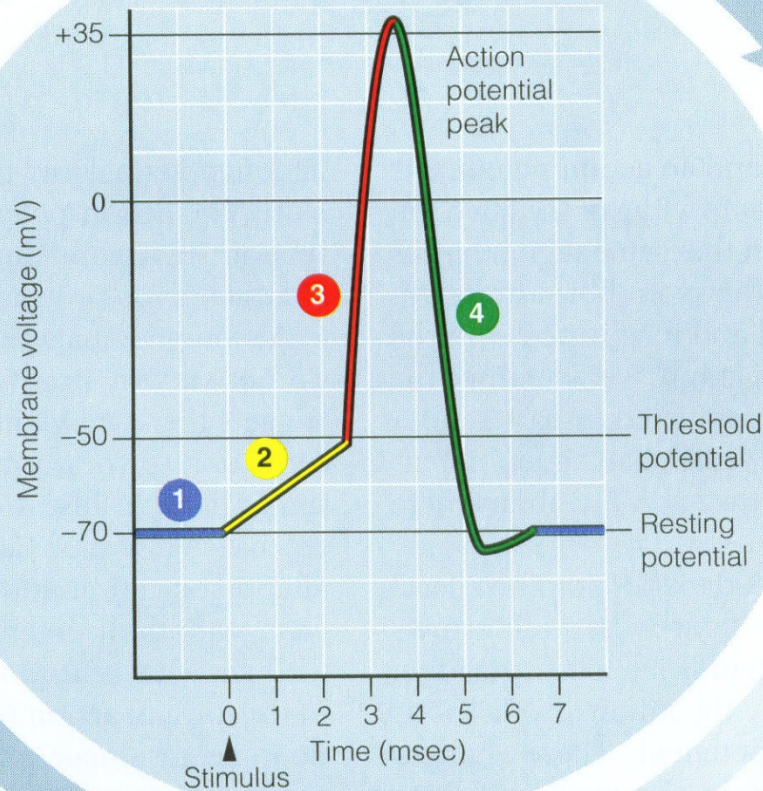
Time (ms)



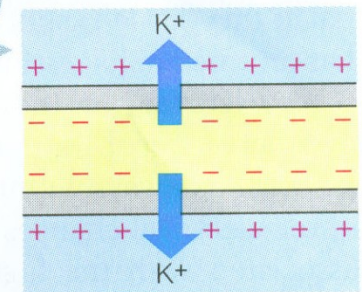
2 Na^+ gates start to open, and some Na^+ enters the neuron.



1 Resting state: no net ion flow across the membrane.



3 Many Na^+ gates open, and Na^+ rushes in. (K^+ gates are closed.)



4 Na^+ gates close and inactivate. K^+ gates open, and K^+ rushes out.

No “Resting” Potential for the Weary

Use the Goldman, Hodgkin, Katz equation to calculate the resting potential (V_m) for a cell having the following features:

Na^+ concentration is 150 mM on the outside and 17 mM on the inside

K^+ concentration is 5 mM on the outside and 143 mM on the inside

Cl^- concentration is 165 mM on the outside and 20 mM on the inside

The cell membrane is 25 times more permeable to K^+ and Cl^- than to Na^+