





AN AUTONOMOUS INSTITUTION

Approved by AICTE New Delhi & Affiliated to Anna University Chennai Accredited by NBA & Accredited by NAAC with "A+" Grade, Recognized by UGC

COIMBATORE

DEPARTMENT OF CIVIL ENGINEERING

19CEB201 – CONSTRUCTION MATERIALS

II YEAR / III SEMESTER

Unit 1 : Stones – Bricks – Building Blocks Topic 5 : Deterioration of Stone Work





The various natural agents such as rain, heat, etc. and chemicals deteriorate the stones with time.

RAIN:

- Rain water acts both physically and chemically on stones.
- The physical action is due to the erosive and transportation powers and the latter due to the decomposition, oxidation and hydration of the minerals present in the stones.

PHYSICAL ACTION:

Alternate wetting by rain and drying by sun causes internal stresses in the stones and consequent disintegration.

CHEMICAL ACTION:

In industrial areas the acidic rain water reacts with the constituents of stones leading to its deterioration.





DECOMPOSITION:

- The disintegration of alkaline silicate of alumina in stones is mainly because of the action of chemically active water.
- The hydrated silicate and the carbonate forms of the alkaline materials are very soluble in water and are removed in solution leaving behind a hydrated silicate of alumina (Kaolinite).
- > The decomposition of felspar is represented as

 $K_{2}Al_{2}O_{3}.6H_{2}O + CO_{2} + nH_{2}O = K_{2}CO_{3} + Al_{2}O_{3}.2SiO_{2}.2H_{2}O + 4SiO_{2}.nH_{2}O$





OXIDATION AND HYDRATION:

- Rock containing iron compounds in the forms of peroxide, sulphide and carbonate are oxidised and hydrated when acted upon by aciduated rain water.
- As an example the peroxide—FeO is converted into ferric oxide—Fe2O3 which combines with water to form FeO.nH2O.
- This chemical change is accompanied by an increase in volume and results in a physical change manifested by the liberation of the neighbouring minerals composing the rocks.
- As another example iron sulphide and siderite readily oxidize to limonite and liberates sulphur, which combines with water and oxygen to form sulphuric acid and finally to sulphates.





FROST:

In cold places frost pierces the pores of the stones where it freezes, expands and creates cracks.

WIND:

> Since wind carries dust particles, the abrasion caused by these deteriorates the stones.

TEMPERATURE CHANGES:

Expansion and contraction due to frequent temperature changes cause stone to deteriorate especially if a rock is composed of several minerals with different coefficients of linear expansion.





VEGETABLE GROWTH:

- Roots of trees and weeds that grow in the masonry joints keep the stones damp and also secrete organic and acidic matters which cause the stones to deteriorate.
- Dust particles of organic or nonorganic origin may also settle on the surface and penetrate into the pores of stones.
- When these come in contact with moisture or rain water, bacteriological process starts and the resultant micro-organism producing acids attack stones which cause decay.

MUTUAL DECAY:

- > When different types of stones are used together mutual decay takes place.
- For example, when sandstone is used under limestone, the chemicals brought down from limestone by rain water to the sandstone will deteriorate it.





CHEMICAL AGENTS:

Smokes, fumes, acids and acid fumes present in the atmosphere deteriorate the stones.
Stones containing CaCO₃, MgCO₃ are affected badly.

LICHENS:

- > These destroy limestone but act as protective coats for other stones.
- Molluses gradually weaken and ultimately destroy the stone by making a series of parallel vertical holes in limestones and sandstones.





Thank You!!

10/28/2023

Deterioration of Stone Work/ 19CEB201 -CE/ /Nithyapriya K. / AP / CIVIL/ SNSCT