

# SNS COLLEGE OF TECHNOLOGY



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#### 19FTB202- BIOCHEMISTRY FOR FOOD TECHNOLOGY

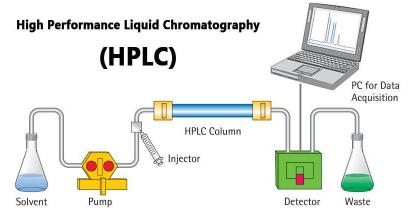
#### UNIT 5 - ADVANCED TECHNIQUES FOR BIOCHEMICAL INVESTIGATION

#### **TOPIC 5 HPLC & HTPLC**

**High-performance liquid chromatography or commonly known as HPLC, is an analytical technique used to separate, identify or quantify each component in a mixture.** The mixture is separated using the basic principle of column **chromatography** and then identified and quantified by spectroscopy.

In the 1960s, the column chromatography LC with its low-pressure suitable glass columns was further developed to the HPLC with its high-pressure adapted metal columns.

HPLC is thus basically a highly improved form of column liquid chromatography. Instead of a solvent being allowed to drip through a column under gravity, it is forced through under high pressures of up to 400 atmospheres.



# **HPLC Principle**

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- The purification takes place in a separation column between a stationary and a mobile phase.
- The stationary phase is a granular material with very small porous particles in a separation column.
- The mobile phase, on the other hand, is a solvent or solvent mixture which is forced at high pressure through the separation column.
- Via a valve with a connected sample loop, i.e. a small tube or a capillary made of stainless steel, the sample is injected into the mobile phase flow from the pump to the separation column using a syringe.
- Subsequently, the individual components of the sample migrate through the column at different rates because they are retained to a varying degree by interactions with the stationary phase.
- After leaving the column, the individual substances are detected by a suitable detector and passed on as a signal to the HPLC software on the computer.

• At the end of this operation/run, a chromatogram in the HPLC software on the computer is obtained.

• The chromatogram allows the identification and quantification of the different substances.

## **Instrumentation of HPLC**

## The Pump

- The development of HPLC led to the development of the pump system.
- The pump is positioned in the most upper stream of the liquid chromatography system and generates a flow of eluent from the solvent reservoir into the system.
- High-pressure generation is a "standard" requirement of pumps besides which, it should also to be able to provide a consistent pressure at any condition and a controllable and reproducible flow rate.
- Most pumps used in current LC systems generate the flow by back-and-forth motion of a motor-driven piston (reciprocating pumps). Because of this piston motion, it produces "pulses".

# Injector

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- An injector is placed next to the pump.
- The simplest method is to use a syringe, and the sample is introduced to the flow of eluent.
- The most widely used injection method is based on sampling loops.
- The use of the autosampler (auto-injector) system is also widely used that allows repeated injections in a set scheduled-timing.

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# Column

- The separation is performed inside the column.
- The recent columns are often prepared in a stainless steel housing, instead of glass columns.
- The packing material generally used is silica or polymer gels compared to calcium carbonate.

The eluent used for LC varies from acidic to basic solvents.

• Most column housing is made of stainless steel since stainless is tolerant towards a large variety of solvents.

#### Detector

 • Separation of analytes is performed inside the column, whereas a detector is used to observe the obtained separation.

- The composition of the eluent is consistent when no analyte is present. While the presence of analyte changes the composition of the eluent. What detector does is to measure these differences.
- This difference is monitored as a form of an electronic signal. There are different types of detectors available.

#### Recorder

- The change in eluent detected by a detector is in the form of an electronic signal, and thus it is still not visible to our eyes.
- In older days, the pen (paper)-chart recorder was popularly used. Nowadays, a computer-based data processor (integrator) is more common.
- There are various types of data processors; from a simple system consisting of the inbuilt printer and word processor while those with software that are specifically designed for an LC system which not only data acquisition but features like peakfitting, baseline correction, automatic concentration calculation, molecular weight determination, etc.

## Degasser

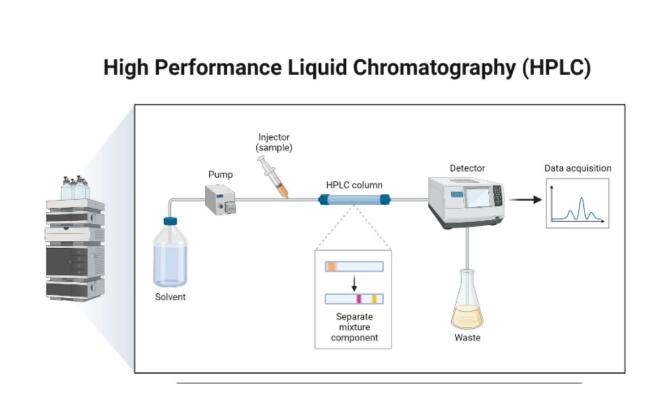
The eluent used for LC analysis may contain gases such as oxygen that are non-visible to our eyes.

- When gas is present in the eluent, this is detected as noise and causes an unstable baseline.
- Degasser uses special polymer membrane tubing to remove gases.
- The numerous very small pores on the surface of the polymer tube allow the air to go through while preventing any liquid to go through the pore.

## **Column Heater**

The LC separation is often largely influenced by the column temperature.

- In order to obtain repeatable results, it is important to keep consistent temperature conditions.
- Also for some analysis, such as sugar and organic acid, better resolutions can be obtained at elevated temperatures (50 to 80°C).
- Thus columns are generally kept inside the column oven (column heater).



## **Types of HPLC**

## 1. Normal phase:

Column packing is polar (e.g silica) and the mobile phase is non-polar. It is used for watersensitive compounds, geometric isomers, cis-trans isomers, and chiral compounds.

## 2. Reverse phase:

The column packing is non-polar (e.g C18), the mobile phase is water+ miscible solvent (e.g methanol). It can be used for polar, non-polar, ionizable, and ionic samples.

## 3. Ion exchange:

Column packing contains ionic groups and the mobile phase is buffer. It is used to separate anions and cations.

## 4. Size exclusion:

Molecules diffuse into pores of a porous medium and are separated according to their relative size to the pore size. Large molecules elute first and smaller molecules elute later.

## **Applications of HPLC**

The HPLC has developed into a universally applicable method so that it finds its use in almost all areas of chemistry, biochemistry, and pharmacy.

• Analysis of drugs

- Analysis of synthetic polymers
- Analysis of pollutants in environmental analytics
- Determination of drugs in biological matrices
- Isolation of valuable products
- Product purity and quality control of industrial products and fine chemicals

- Separation and purification of biopolymers such as enzymes or nucleic acids
- Water purification
- Pre-concentration of trace components
- Ligand-exchange chromatography
- Ion-exchange chromatography of proteins
- High-pH anion-exchange chromatography of carbohydrates and oligosaccharides

The **HPTLC** technique is an automated and sophisticated form of thin layer chromatography with superior and advanced separation efficiency and detection limits and is often an exceptional alternative to high-performance liquid chromatography (HPLC) and gas chromatography (GC). The high-performance thin-layer chromatography is also known as flat-bed chromatography or as planar chromatography.

#### High-Performance Thin-Layer Chromatography (HPTLC) Principle

The HPTLC works on the same principles as TLC such as the principle of separation is adsorption. The mobile phase or solvent flows through the capillary action. The analytes move according to their affinities towards the stationary phase (adsorbent). The higher affinity component travels slower towards the stationary phase. A low-affinity component travels rapidly toward the stationary phase. On a chromatographic plate, then, the components are separated.

#### **1.Sample Preparation:**

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This requires a highly concentrated solution since much less sample quantity needs to be applied. The plate's solvents must be non-polar of the volatile type. Polar solvents are commonly used to dissolve samples for reversed-phase chromatography.

#### 2.Selection of Chromatographic Layers:

The layer of HPTLC is available in the form of very fine particle size silica gel pre-coats which is widely used as adsorbent.

#### **3.Pre Washing:**

To water vapor or volatile impurities, the plates must be cleaned. It may be clean with a suitable solvent such as methanol.

## 4. Conditioning:

Plates are placed in an oven at 120 ° C for 15 to 20 minutes to perform conditioning.

# 5. Sample Application:

The size of the sample spot is not greater than 1 mm in diameter. There are various methods for spotting samples in HPTLC. One is a self-loading capillary in which small quantities of samples can apply on the **HPTLC** plate.

# 6. Pre-Conditioning:

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Saturation is necessary for highly polar mobile phases although there is no need for saturation for low polarity mobile phases.

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# 7. Mobile Phase of HPTLC:

Through trial and error, the mobile phase of the suitable solvents is to be selective.

# 8. Chromatographic Development:

The linear development method in high-performance thin-layer chromatography is the most common technique here the plate is positioned vertically in an appropriate container with a solvent or mobile phase. The mobile phase is generally fed by capillary action and both sides may produce chromatograms.

# 9. Detection of spot and Scanning:

The HPTLC instrument has attached to computer and data recording devices. The development of spots is viewed as peaks at wavelengths of selected UV regions. The height and the area of the peaks are determined by the instrument and recorded as a percentage.