



## **SNS COLLEGE OF TECHNOLOGY**

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#### **DEPARTMENT OF FOOD TECHNOLOGY**

#### **19FTT202 – HEAT AND MASS TRANSFER**

#### II YEAR IV SEM

### **UNIT 1 - CONDUCTION**

**Topic 9 Conduction with Internal Heat Generation** 





- A medium through which heat is conducted may involve the conversion of mechanical, electrical, nuclear, or chemical energy into heat (or thermal energy).
- In heat conduction analysis, such conversion processes are characterized as heat generation.
- For example, the temperature of a resistance wire rises rapidly when electric current passes through it as a result of the electrical energy being converted to heat at high rate

Following are some of the cases where heat generation and heat conduction are encountered :

- (i) Fuel rods nuclear reactor;
- (ii) Electrical conductors;
- (iii) Chemical and combustion processes;
- (iv) Drying and setting of concrete.

It is of paramount importance that the heat generation rate be controlled otherwise the equipment may fail (*e.g.*, some nuclear accidents, electrical fuses blowing out). Thus, in the design of the thermal systems temperature distribution within the medium and the rate of heat dissipation to the surround-ings assumes ample importance / significance.





• A large amount of heat is generated in the fuel elements of nuclear reactors as a result of nuclear fission that serves as the *heat source* for the nuclear power plants

• The natural disintegration of radioactive elements in nuclear waste or other radioactive material also results in the generation of heat throughout the body

• The heat generated in the sun as a result of the fusion of hydrogen into helium makes the sun a large nuclear reactor that supplies heat to the earth





Refer to Fig. 2.91. Consider a plane wall of thickness L (small in comparison with other dimension) of uniform thermal conductivity k and in which heat sources are uniformly distributed in the whole volume. Let the wall surfaces are maintained at temperatures  $t_1$  and  $t_2$ .

Let us assume that heat flow is onedimensional, under steady state conditions, and there is a *uniform volumetric heat generation* within the wall.

Consider an element of thickness at a distance x from the left hand face of the wall.

Heat conducted in at distance x,

$$Q_{x} = -kA \frac{dt}{dx}$$

Heat generated in the element,

$$Q_g = A \cdot dx \cdot q_g$$

(where  $q_g$  = heat generated per unit volume per unit time in the element)

Heat conducted out at distance

$$(x + dx), \ Q_{(x + dx)} = Q_x + \frac{d}{dx}(Q_x) dx$$



Fig. 2.91. Plane wall uniform heat generation. Both the surfaces maintained at a common temperature.





...(2.88)

As  $Q_x$  represents an energy increase in the volume element, an energy balance on the element of thick dx is given by

or,

 $Q_g = \frac{d}{dx}(Q_x) dx$   $q_g \cdot A \cdot dx = \frac{d}{dx} \left[ -k A \frac{dt}{dx} \right] dx$   $= -k A \cdot \frac{d^2 t}{dx^2} \cdot dx$   $\frac{d^2 t}{dx^2} + \frac{q_g}{k} = 0$ 

 $=Q_x + \frac{d}{dx}(Q_x) dx$ 

or.



Eqn. (2.88) may also be obtained from eqn. (2.8) by assuming one-dimensional steady state conditions.

The first and second integration of Eqn. (2.88) gives respectively

 $Q_x + Q_g = Q_{(x+dx)}$ 

$$\frac{dt}{dx} = -\frac{q_g}{k}x + C_t \qquad ...(2.89)$$

$$t = -\frac{q_g}{2k} \cdot x^2 + C_1 x + C_2 \qquad \dots (2.90)$$

#### Case I. Both the surfaces have the same temperature :

Refer to Fig. 2.92.

At x = 0  $t = t_1 = t_w$ , and At x = L  $t = t_2 = t_w$ 

(where 
$$t_{m}$$
 = temperature of the wall surface

Using these boundary conditions in eqn. (2.90), we get

$$C_2 = t_w$$
 and  $C_1 = \frac{q_g}{2k} \cdot L$ 

Substituting these values of  $C_1$  and  $C_2$  in eqn. (2.90), we have

$$t = -\frac{q_s}{2k}x^2 + \frac{q_s}{2k} \cdot L \cdot x + t_w$$
  
$$t = \frac{q_s}{2k}(L - x)x + t_w \qquad \dots (2.91)$$

or,

In order to determine the location of the maximum temperature, differentiating the eqn. (2.91)w.r.t x and equating the derivative to zero, we have

$$\frac{dt}{dx} = \frac{q_g}{2k}(L - 2x) = 0$$

$$\frac{q_g}{2k} \neq 0, \text{ therefore,}$$

Since,

$$L - 2x = 0$$
 or  $x = \frac{L}{2}$ 

Thus the distribution of temperature given by eqn. (2.91) is the parabolic and symmetrical about

the midplane. The maximum temperature occurs at 
$$x = \frac{L}{2}$$
 and its value equals  
$$t_{max} = \left[\frac{q_g}{2k}(L-x)x\right]_{x=\frac{L}{2}} + t_w$$





or.

 $= \left[\frac{q_x}{2k}\left(L - \frac{L}{2}\right)\frac{L}{2}\right] + t_w$   $t_{max} = \frac{q_x}{2k} \cdot L^2 + t_w \qquad \dots (2.92)$ 

i.e.

Heat transfer then takes place towards both the surfaces, and for each surface it is given by

$$Q = -kA \left(\frac{dt}{dx}\right)_{x=0 \text{ or } x=L}$$
  
=  $-kA \left[\frac{q_s}{2k} (L-2x)\right]_{x=0 \text{ or } x=L}$   
 $Q = \frac{AL}{2} \cdot q_s$  ...(2.93)

1.0.0

When both the surfaces are considered,

$$Q = 2 \times \frac{AL}{2} q_s = A \cdot L \cdot q_s$$
 [2.93 (a)]

Also heat conducted to each wall surface is further dissipated to the surrounding atmosphere at temperature  $t_{a}$ .

Thus.

OF.

 $\frac{AL}{2} \cdot q_g = hA(t_w - t_o)$   $t_w = t_o + \frac{q_g}{2h} \cdot L \qquad \dots (2.94)$ 

Substituting this value of  $t_w$  in eqn. (2.91), we obtain

$$t = t_a + \frac{q_e}{2k} \cdot L + \frac{q_s}{2k} (L - x) x \qquad ...(2.95)$$

x = L/2 *i.e.*, at the midplane :

At



temperatures.









The eqn. (2.95) also works well in case of conduction in an insulated walf Fig. (2.92).

The following boundary conditions apply in the full hypothetical walt of thickness 2L:

The location x = L refers to the mid-plane of the hypothetical wall (or insulated face of given wall).

Eqns. (2.91) and (2.92) for temperature distribution and maximum temperature at the mid-plane (insulated end of the given wall) respectively can be written as

$$t = \frac{q_g}{2k} (2L - x) x + t_w \quad \dots (2.96)$$
$$t_{max} = \frac{q_g}{2k} L^2 + t_w \quad \dots (2.97)$$

[Substituting L = 2L in eqn. (2.91) and (2.92)]

Case II. Both the surfaces of the wall have different temperatures :

Refer to Fig. 2.93

The boundary conditions are :

At 
$$x = 0$$
  
At  $x = L$   
 $t = t_{r_1}$ 

Substituting these values in eqn. (2.90), we obtain the values of constant  $C_1$  and  $C_2$  as :

$$C_2 = t_{w1};$$
  $C_1 = \frac{t_{w2} - t_{w1}}{L} + \frac{q_s}{2k} \cdot L$ 

Inserting these values in eqn. (2.90), we get

$$\begin{split} t &= -\frac{q_x}{2k} x^2 + \frac{t_{w2} - t_{w1}}{L} x + \frac{q_x}{2k} L \cdot x + t_{w1} \\ &= \frac{q_x}{2k} L \cdot x - \frac{q_x}{2k} x^2 + \frac{x}{L} (t_{w2} - t_{w1}) + t_1 \\ t &= \left[ \frac{q_x}{2k} (L - x) + \frac{t_{w2} - t_{w1}}{L} \right] x + t_{w1} \end{split}$$

OT.

The temperature distribution, in dimensionless form can be obtained by making the following transformations :

$$t - t_{w2} = \frac{q_x}{2k} L^2 \left[ \frac{x}{L} - \left( \frac{x}{L} \right)^2 \right] + \frac{x}{L} (t_{w2} - t_{w3}) + (t_{w1} - t_{w2})$$
  
$$\frac{t - t_{w2}}{(t_w - t_w)} = \frac{q_y}{2k} \cdot \frac{L^2}{(t_w - t_w)} \left[ \frac{x}{L} - \left( \frac{x}{L} \right)^2 \right] - \frac{x}{L} + 1$$

OT.

or, 
$$\frac{I - I_{w2}}{I_{w1} - I_{w2}} = \frac{q_g}{2k} \frac{L^2}{(I_{w1} - I_{w2})} \cdot \frac{s}{L} \left[ 1 - \frac{s}{L} \right] + \left[ 1 - \frac{s}{L} \right]$$

Replacing the parameter  $\frac{q_z}{2k} \frac{L^2}{(t_{w1} - t_{w2})}$  (a constant) by a factor Z, we have

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Vertical tank.

...(2.98)





$$\frac{t - t_{w2}}{t_{w1} - t_{w2}} = Z \cdot \frac{x}{L} \left[ 1 - \frac{x}{L} \right] + \left[ 1 - \frac{x}{L} \right]$$
  
or,  
$$\frac{1 - t_{w2}}{t_{w1} - t_{w2}} = \left[ 1 - \frac{x}{L} \right] \left[ \frac{Z x}{L} + 1 \right]$$
  
In order to get maximum temperature and  
its location, differentiating Eqn. (2.99) w.r.t x  
and equating the derivative to zero, we have  
$$\frac{dt}{d(x/L)} = \left( 1 - \frac{x}{L} \right) Z + \left( \frac{Zx}{L} + 1 \right) (-1) = 0$$

$$d(x/L) = (1 - L)^{2L} + (L + 1)^{(-1)} = 0$$
  
or,  $Z = \frac{Zx}{L} - \frac{Zx}{L} - 1 = 0$   
or,  $\frac{2Zx}{L} = Z - 1$   
or,  $\frac{x}{L} = \frac{Z - 1}{2Z} \dots (2.100)$ 

Thus the maximum value of temperature occurs at  $\frac{x}{r} = \frac{Z-1}{2\pi}$  and its value is given by:

$$Fig. 2.94$$

$$Fig. 2.94$$

$$\frac{I_{max} - I_{w2}}{I_{w1} - I_{w2}} = \left[1 - \frac{Z - 1}{2Z}\right] \left[Z \times \left(\frac{Z - 1}{2Z}\right) + 1\right]$$

$$\frac{I_{max} - I_{w2}}{I_{w1} - I_{w2}} = \left(\frac{Z + 1}{2Z}\right) \left(\frac{Z + 1}{2}\right)$$

$$(Z + 1)^{2}$$

4Z





...(2.101)

 $\operatorname{cor}_{p}$ 

Fig. 2.94 shows the effect of factor Z on the temperature distribution in the plane wall. The following points emerge :

- As the value of Z increases the slope of the curve changes; obviously the direction of heat flow can be reversed by an adequately large value of q<sub>e</sub>.
- When Z = 0, the temperature distribution is *linear* (*i.e.*, no internal heat generation).
- When the value of Z is negative, q<sub>e</sub> represents absorption of heat within the wall/body.

Case III. Current carrying electrical conductor :

When electrical current passes through a conductor, heat is generated  $(Q_{g})$  in it and is given by

$$Q_g = I^2 R, \text{ where } R = \frac{p L}{A}$$

$$I = \text{Current flowing in the conductor,}$$

$$R = \text{Electrical resistance,}$$

$$\rho = \text{Specific resistance or resistivity,}$$

$$L = \text{Length of the conductor, and}$$

$$A = \text{Area of cross-section of the conductor,}$$

$$Q_g = q_g \times A \times L$$

$$Q_g = q_g \times A \times L$$

Also,

d,

where,

$$q_{g} \times A \times L = I^{2} \times \frac{\rho L}{A}$$
 or,  $q_{g} = I^{2} \times \frac{\rho L}{A} \times \frac{1}{AL} = \frac{I^{2} \rho}{A^{2}}$ 





# **1.The temperature drop in a plane wall with uniformly distributed heat generation can be decreased by reducing**

a) Wall thickness

b) Heat generation rate

c) Thermal conductivity

d) Surface area

Answer: a

Explanation: On decreasing wall thickness, generally temperature drop

decreases.





#### 2.Notable example of uniform generation of heat within the conducting medium

are

- (i) Energy of a nuclear reactor
- (ii) Liberation of energy due to some exothermic chemical reactions
- (iii) Resistance heating in electrical appliances

#### Which of the statements made above are correct?

a) i, ii and iii b) i and ii c) i and iii d) Only ii

#### Answer: a

Explanation: All the statements are correct with respect to plane wall heat conduction.







#### **Book references:**

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