Biomolecules

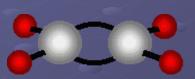
What is an organic compound?

What is so special about Carbon?

- Compounds containing C, H,O and often N, P, & S.
- Organic compounds make up all living things and are necessary for life.
- It can combine to form long chains which act as the backbone of large molecules.
- Macromolecules giant molecules.

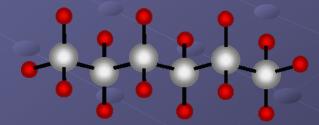
- How does carbon bond?
- Carbon needs to bond 4 times to fill it's outer shell.
- It can form single, double or triple covalent bonds.





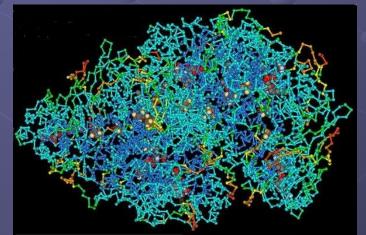


Carbon can form straight chains, rings or branched chains.



- Very large molecules.
- Carbon compounds can vary greatly in size. Some contain just one or two C atoms, others can have 10 or even 1000 C atoms.

 Macromolecules form when many smaller molecules bond together.



What is a polymer?

What is a monomer?

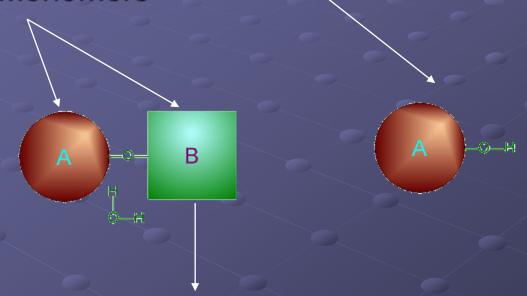
$$\begin{array}{cccc} CH_3 & & & CH_3 \\ Cl - Si - Cl & & & & - Si - I_n \\ CH_3 & & & CH_3 & & CH_3 \end{array}$$

- A molecule made up of many smaller molecules.
- Formed by a reaction called dehydration synthesis – which means water must be removed to bond them together.
- The building block of a polymer. Varies depending on the type of molecule being built

broken down?

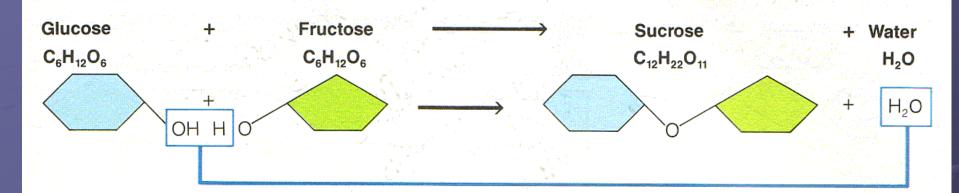
How are polymers
By a chemical reaction known as hydrolysis. Water is added back in and the monomers separate.

Monomers

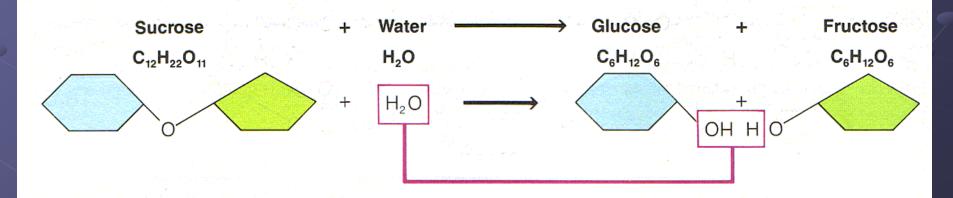


This is dehydration synthesis

DEHYDRATION SYNTHESIS



HYDROLYSIS

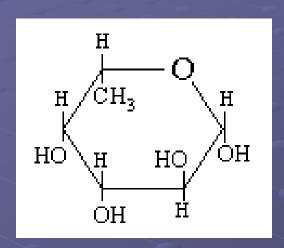


What is a carbohydrate?



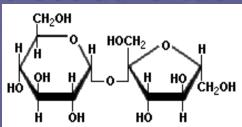
- Organic compound composed of C, H, & O in a 1:2:1 ratio
- \bullet C₆H₁₂O₆
- 3 types monosaccharides, disaccharides and polysaccharides.
- Function: main source of energy for all living things.
- Some structure (ex plant cell walls)

What is a monosaccharide?



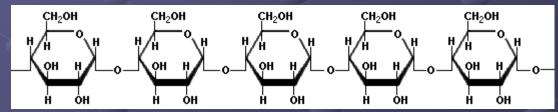
- Simple sugar only one sugar.
- Contains 3 7 carbon atoms in their skeleton.
- Can take ring form or straight chain form.
- ** monosaccharides are the building blocks for all larger carbs **

What is a dissaccharide?



What is a polysaccharide?

- Two monosaccharides combined minus water.
- Sucrose = glucose + fructose
- When many monosaccharides combine to form a large carbohydrate.
- Have no fixed size, but must be broken down into simple sugars to be used by the cell.
- Ex. Starch and cellulose

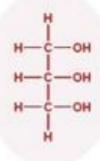


- Summary 3 Types of Carbohydrates
- 1. monosaccharides single sugar.
 - Ex. Glucose, fructose, galactose
 - Aka simple sugars
- 2. <u>Disaccharide</u>: 2 simple sugars
 - Ex. Sucrose (table sugar) maltose
- 3. polysaccharides: 3 or more sugars (complex carbs)
 - Ex. Cellulose used in cell walls
 - Starch stores energy in plants
 - Glycogen stored energy in animals

What are lipids?

- Organic compounds made up of C, H, & O, but not in any fixed ratio.
- The building blocks of lipids are fatty acids.
- Usually 3 fatty acids combine with one glycerol to form a triglyceride.
- properties of fats and oils are determined by the fatty acids that make them up.

Glycerol



A "free" Fatty Acid

Triglyceride

• What is a saturated fat?

- What is an unsaturated fat?
- All the carbon atoms are joined by single bonds (usually solid fats)
- The carbon chain contains double or triple bonds (usually oils)

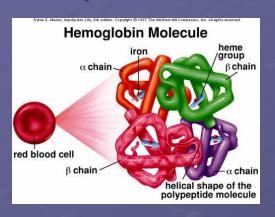
Saturated

Unsaturated

What is the function of lipids?

- Lipids are often called fats or oils, but are large macromolecules with 2 primary functions:
- 1. long term energy storage
- 2. building cell membranes.

What are proteins?



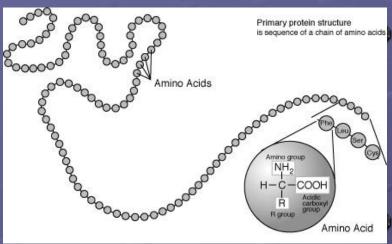
- Organic compounds that contain C, H, O & N.
- Every cell contains protein.
- Functions of protein:
 - Used in structural components.
 - Messengers and receptors on the cell membrane
 - Defend against disease
 - Act as facilitators for chemical reactions (ENZYMES)

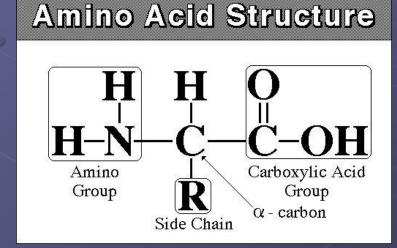
What are amino acids?

Amino acids the building blocks of proteins!!

They consist of a central carbon atom with a H, a – COOH, a NH₂ and a "R" group attached.

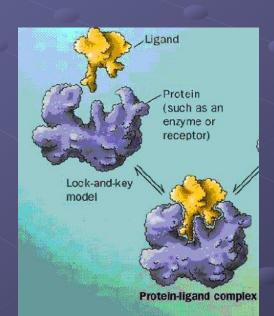
The "R" group is different for each of the 20 different amino acids.

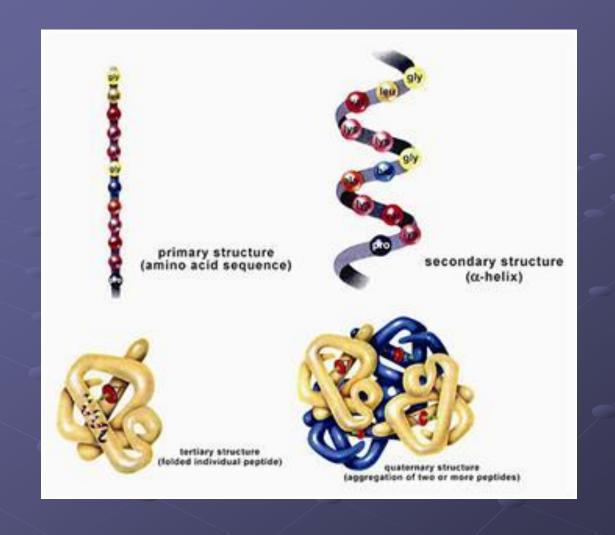




- What is a peptide bond?
- The bond that holds together amino acids into a large macromolecule called a polypeptide.
- Longer polypeptides are called proteins and can be made up of 50 – 300 amino acids.

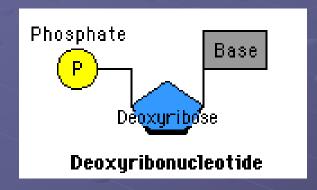
- The order of amino acids give a protein its shape. The shape determines the protein's function.
- Even one amino acid out of place will prevent a protein from doing its job.
- Proteins that speed up the rate of chemical reactions
- Without enzymes chemical reactions would occur too slowly for life to exist.





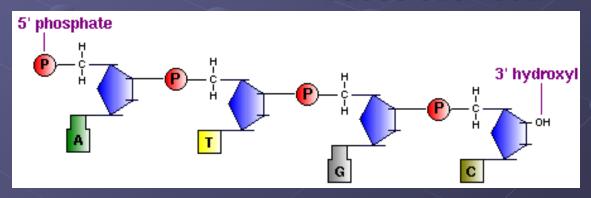
- What are nucleic acids?
- Organic molecule made up of C,H,O,N,& P
- Nucleic acids are passed from parent to offspring, you get one copy from each parent for a total of 2 complete sets.
- Nucleic acids dictate amino acid sequence in proteins which in turn control all life processes.
- DNA forms the genes or units of genetic material that determine your characteristics.

What is a nucleotide?



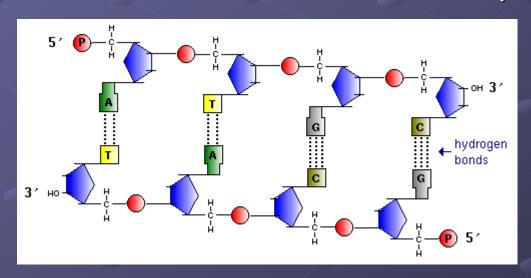
- Nucleotides are the building blocks of Nucleic acids.
- Each nucleotide is made up of 3 parts:
 - A 5 Carbon sugar (deoxyribose or ribose)
 - A phosphate group
 - A nitrogen base (a ring containing C, H, & N)

- What are the different types of nucleotides?
- Adenine, guanine, cytosine thymine, and uracil.
- Thymine is only in DNA, uracil is only in RNA.
- Adenine pairs with thymine (uracil)
- Guanine pairs with cytosine.
- Nucleotides link together between sugars and phosphates, nitrogen bases stick out.



What is DNA?

- Deoxyribonucleic acid
- Contains the sugar deoxyribose.
- The molecule of heredity.
- Double stranded, sugar and phosphates form the back bone, paired nitrogen bases hold the two strands together.
- The shape is called a double helix.



- What is RNA?
- Ribonucleic acid
- Contains the sugar ribose, uracil replaces thymine.
- Single stranded.
- 3 types each with a different function
 - Ribosomal
 - Transfer
 - messenger