SNS COLLEGE OF PHARMACY AND HEALTH SCIENCES



Coimbatore -641035

COURSE NAME: NOVEL DRUG DELIVERY SYSTEM (BP 706 T)

VII SEM / IV YEAR

TOPIC 1: CONTROLLED RELEASE DRUG DELIVERY SYSTEMS



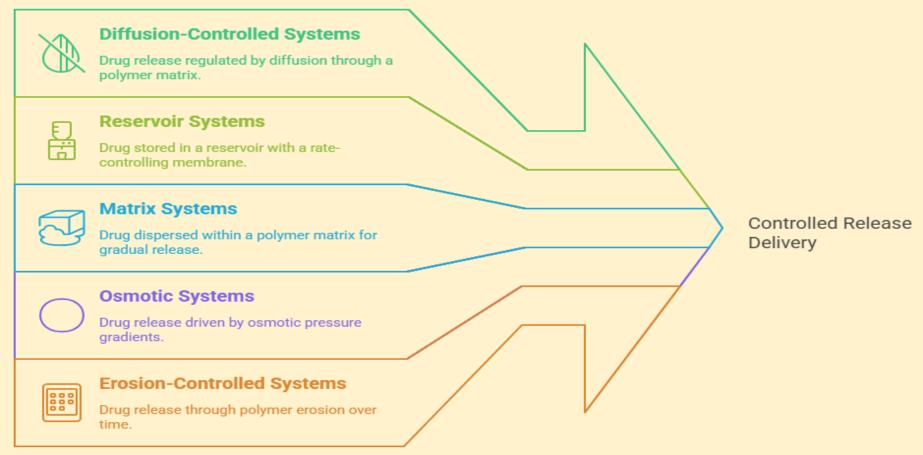
Design Thinking in CRDDS

- 1. Empathize Deeply understand the user or patient's challenges, needs, and experiences. This involves engaging with patients, caregivers, and healthcare providers to uncover pain points, preferences, and unmet needs
- 2. **Define**: Reframe the problem based on insights from the empathize phase and establish clear context. This involves synthesizing data to pinpoint the core issue, such as defining the need for a more convenient drug delivery method
- **3. Ideate:** Brainstorm and explore a wide range of ideas and potential solutions, including innovative biomaterials or delivery systems.
- 4. Prototype: Simulate and build delivery devices to enhance patient compliance

MINDMAP



Pathways to Controlled Release



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INTRODUCTION



Controlled drug delivery systems can include the maintenance of drug levels within a desired range, the need for fewer administrations, optimal use of the drug in question, and increased patient compliance.

The basic idea behind CDDS concept is to alter the pharmacokinetics & pharmacodynamics of bioactivities either by modifying the molecule structure or physiological parameters.

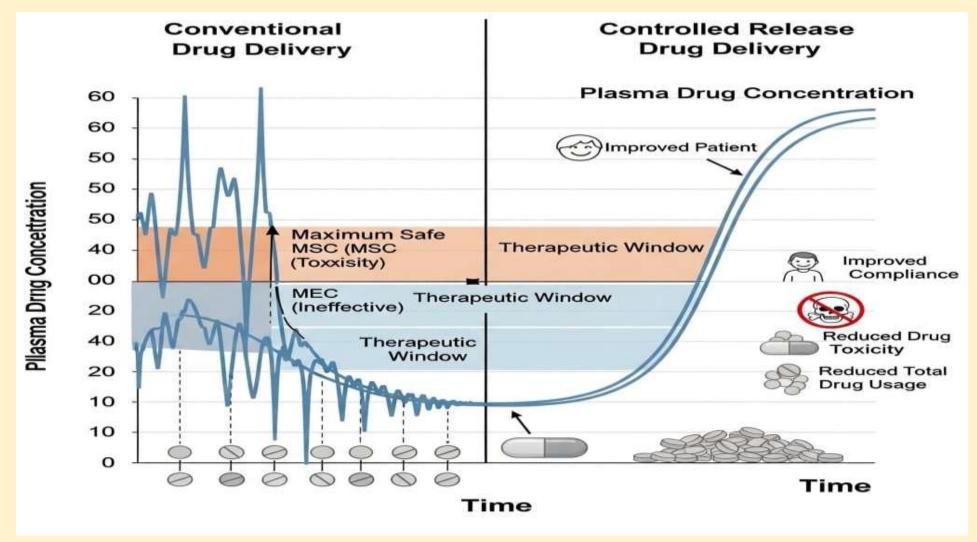




ADVANTAGES OF CRDDS



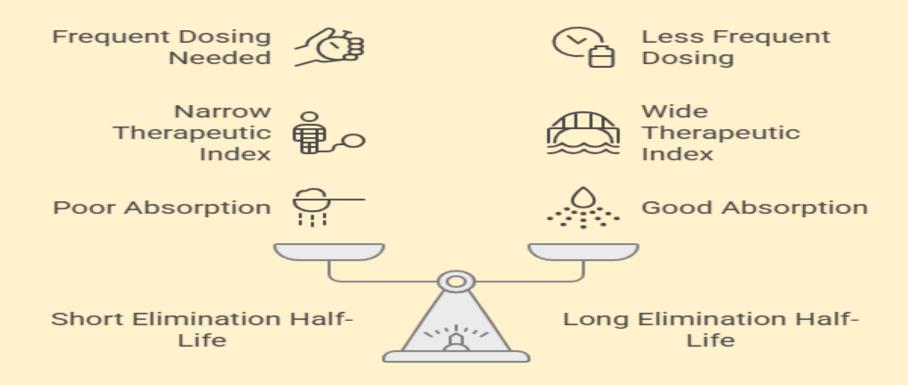
Empathize



DRUGS SUITABLE FOR CRDDS



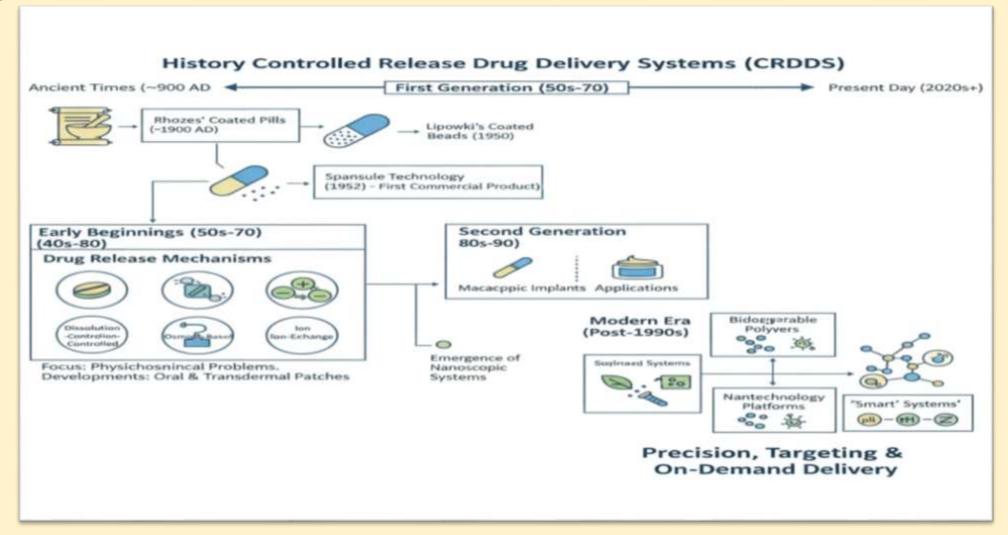






HISTORY OF CRDDS



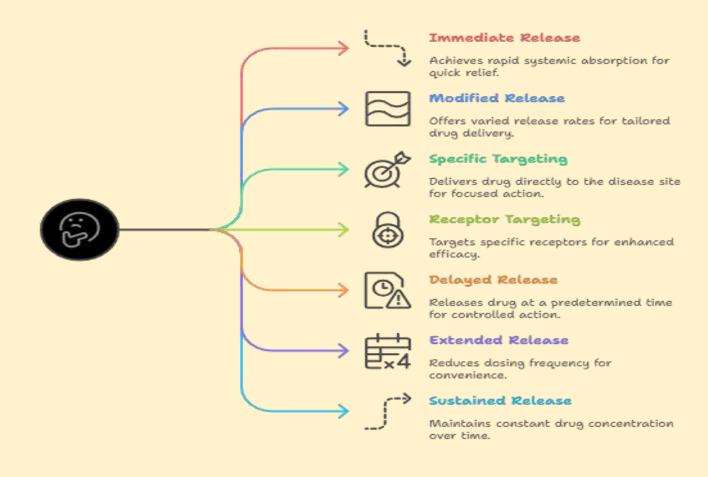




TERMINOLOGY USED IN CRDDS



CONTROLLED RELEASE DRUG DELIVERY SYSTEM





APPROACHES TO DESIGN CONTROLLED RELEASE FORMULATIONS



Methods of Controlled Release

Dissolution
Controlled
Release
Resins
Controlled
Release
Resins
Controlled
Release

Diffusion Controlled Release Diffusion and Dissolution Systems

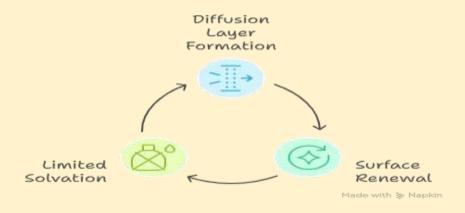
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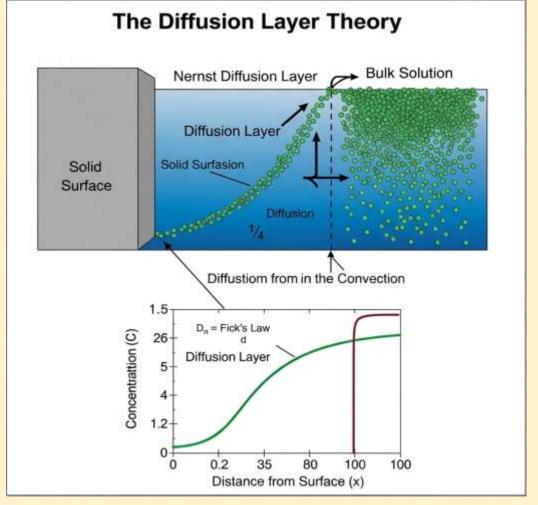




It is a rate determining step when liquid is diffusing from solid.

Theories of Dissolution



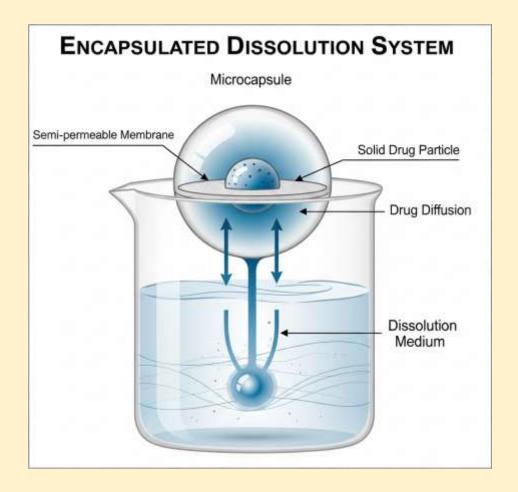


ENCAPSULATED DISSOLUTION SYSTEM





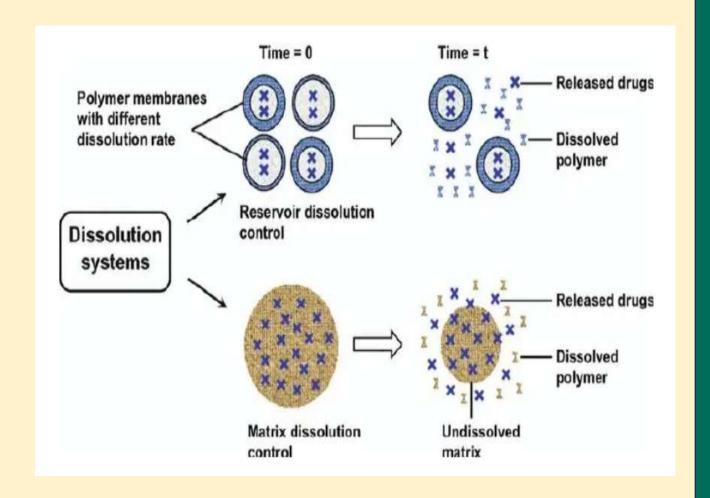
- Refers to a mechanism where the drug is enclosed within an encapsulating material, such as a capsule or coating, that controls its release through dissolution.
- The encapsulating material dissolves gradually in the presence of body fluids, allowing the drug to be released over time.





MATRIX DISSOLUTION-CONTROLLED SYSTEMS

- The drug is embedded with in a matrix that dissolves slowly, releasing the drug gradually.
- This approach is often used for water-soluble drugs where the matrix material controls the release rate.





DIFFUSION-CONTROLLED SYSTEMS

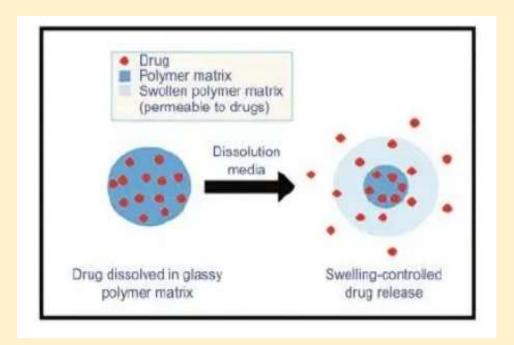
- This involves the movement of drug molecules from a region of higher concentration (within the delivery system) to a region of lower concentration (the surrounding environment, such as bodily fluids) through a rate-limiting barrier, such as a polymer matrix or membrane.
- The rate of release depends on factors like the drug's concentration gradient, the properties of the barrier (e.g., porosity, thickness), and the diffusion coefficient of the drug.

RESERVOIR SYSTEMS



In these systems, the drug is enclosed within a core surrounded by a rate-controlling polymeric membrane. Examples include transdermal patches

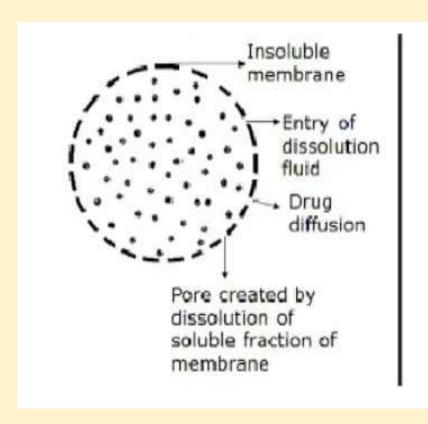
(e.g., Nitro-Dur for nitro-glycerine) and ocular inserts (e.g., Ocusert for pilocarpine).







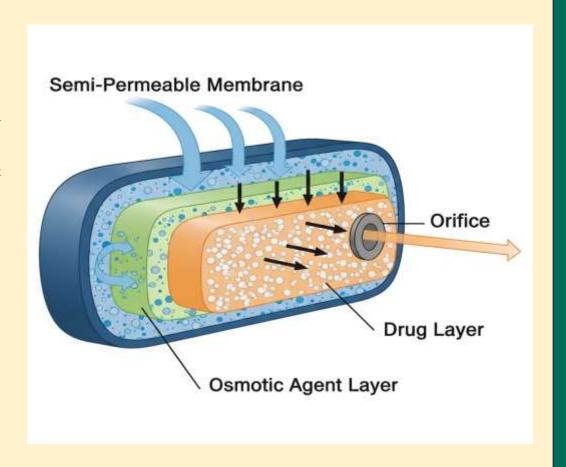
In matrix systems, the drug is dispersed uniformly throughout a polymeric matrix. The drug release occurs as it diffuse out of the matrix. (e.g., Theophylline in Theo-24)







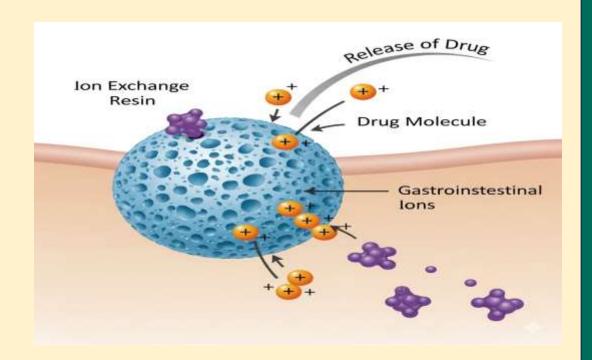
Utilize osmotic pressure to control drug release. The system typically consists of a core containing the drug and osmotic agents surrounded by a semi-permeable membrane with a laser-drilled orifice..



ION EXCHANGE SYSTEMS



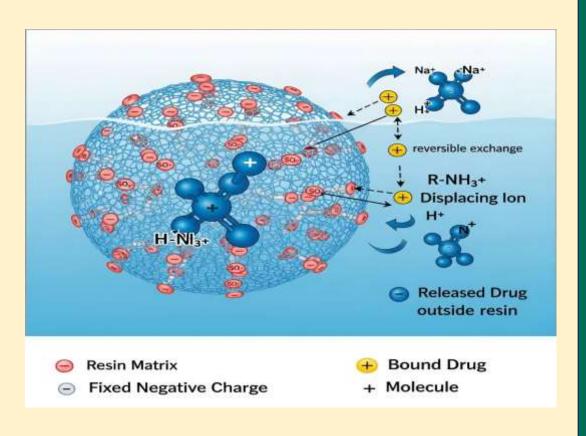
• The drug is bound to an ion exchange resin, and release occurs when ions in the gastrointestinal tract exchange with the drug ions, causing a controlled release





MECHANISM OF ION EXCHANGE RESINS

• The basic principle of ion exchange resins involves the reversible exchange of ions between the resin and the surrounding medium.



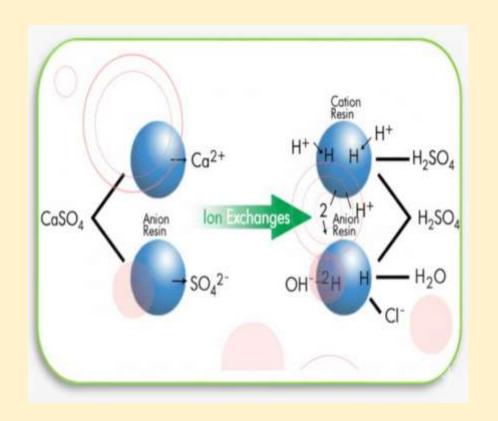


CATION EXCHANGE RESINS:

These resins contain acidic functional groups (e.g., sulfonic acid or carboxylic acid) and exchange their positively charged counter-ions (usually hydrogen ions) with cationic drugs

ANION EXCHANGE RESINS:

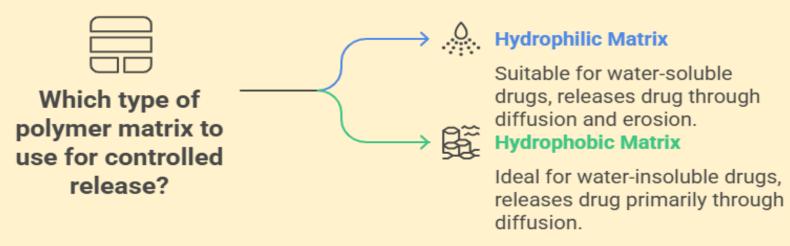
These resins contain basic functional groups(e.g., quaternary ammonium) and exchange their negatively charged counter-ions (usually chloride or hydroxide ions) with anionic drugs







- They are versatile and widely used in controlled release formulations. In these systems, the drug is embedded in a polymer matrix that controls the release rate.
- The matrix can be designed to release the drug by diffusion, erosion, or a combination of both.



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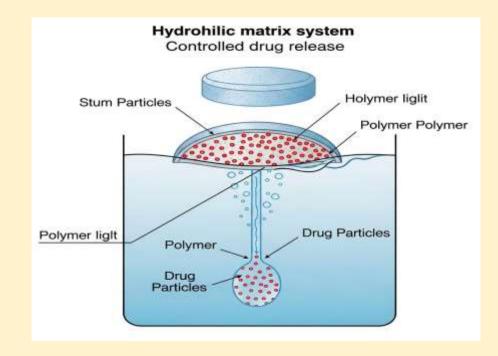
HYDROPHILIC MATRIX

• The drug is dispersed in a hydrophilic matrix

(e.g., hydroxypropyl methylcellulose). When exposed to water

or gastrointestinal fluids, the matrix swells, forming a gel

barrier that control drug release





HYDROPHOBIC MATRIX SYSTEMS

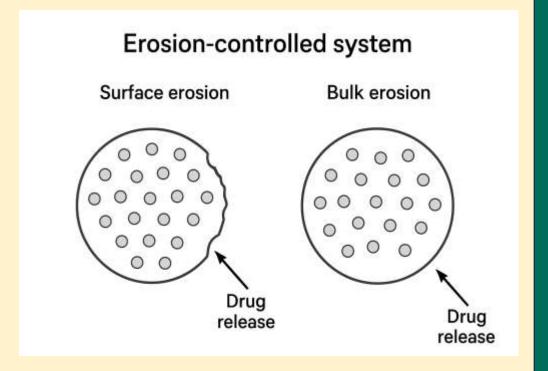
• The drug is dispersed in a hydrophobic matrix (e.g., ethyl cellulose) that remains intact in the gastrointestinal tract and the drug is released by diffusion through the matrix.

Example: Ferro-Sequels (ferrous fumarate) uses a hydrophobic matrix to control iron release for the treatment of anaemia.

EROSION-CONTROLLED SYSTEMS



• Erosion-controlled systems rely on the gradual degradation or erosion of the matrix material to control drug release.





pH-RESPONSIVE SYSTEMS

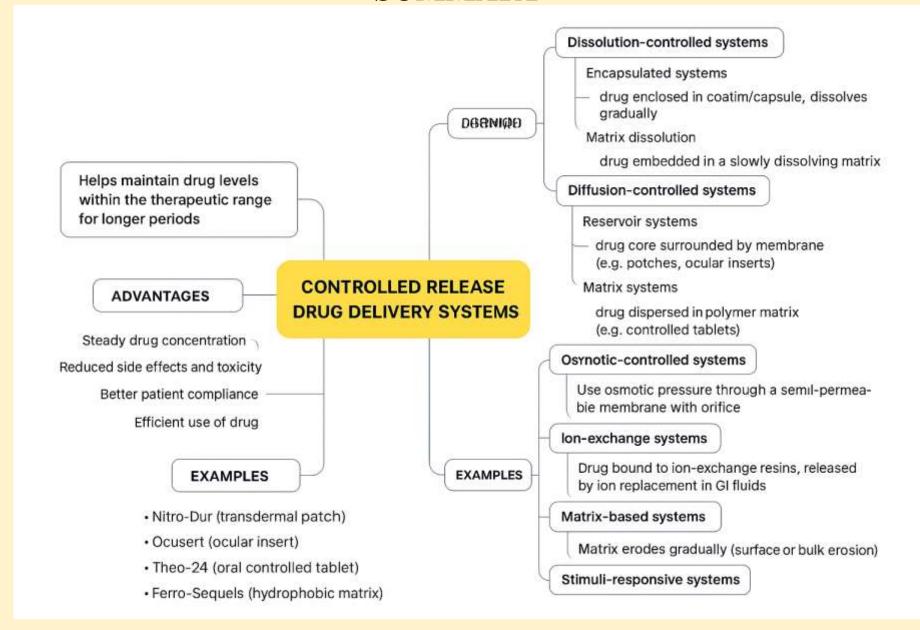
• These systems release drugs in response to pH changes in the environment. They are particularly useful for targeting specific regions in the gastrointestinal tract

THERMO-RESPONSIVE SYSTEMS

• These systems release drugs in response to temperature changes. For example, hydrogels that become more permeable at body temperature can be used for localized delivery

SUMMARY





25-09-2025

CLASS ASSESMENTS



1. What is the primary advantage of Controlled Release Drug Delivery Systems (CRDDS) over conventional drug delivery?

- a. They require more frequent administration.
- b. They maintain drug levels within a desired range for an extended period.
- c. They reduce patient compliance.
- d. They are only suitable for drugs with a narrow therapeutic index.
- 2. In an osmotic-controlled system, what is the key mechanism for drug release?
- a. Gradual dissolution of the polymer matrix.
- b. Diffusion of the drug through a polymeric membrane.
- c. The gradual degradation or erosion of the matrix material.
- d. Osmotic pressure pushes the drug out through a laser-drilled orifice.



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- 3. Which type of system relies on the reversible exchange of ions between the resin and the surrounding medium to release the drug?
- a. Diffusion-controlled system
- b. Ion-exchange system.
- c. Erosion-controlled system.
- d. Dissolution-controlled system.
- 4. According to the presentation, a drug dispersed in a hydrophilic matrix, such as hydroxypropyl methylcellulose, is released when:
- a. The matrix erodes completely.
- b. The matrix swells and forms a gel barrier.
- c. The drug diffuses through a hydrophobic membrane.
- d. The drug is pushed out by osmotic pressure.



5. What is the main difference between an anion exchange resin and a cation exchange resin?

- a. Anion resins exchange positively charged ions while cation resins exchange negatively charged ions.
- b. Anion resins release drugs through surface erosion while cation resins release drugs through bulk erosion.
- c. Anion resins contain basic functional groups and exchange their negatively charged counter-ions, while cation resins contain acidic functional groups and exchange their positively charged counter-ions.
- d. There is no difference; they are two terms for the same mechanism.

25-09-2025

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