



**19CH103–ENGINEERING CHEMISTRY Unit-5 INSTRUMENTAL METHODS OF ANALYSIS** 

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## Chromatography

The substances in a mixture are not chemically combined, so therefore they can be separated through some physical process.

chromatography, technique for separating the components, or solutes, of a mixture on the basis of the relative amounts of each solute distributed between a moving fluid stream, called the mobile phase, and a contiguous stationary phase.

The mobile phase may be either a liquid or a gas, while the stationary phase is either a solid or a liquid. Chromatography is the ability to separate molecules using partitioning characteristics of molecule to remain in a stationary phase versus a mobile phase. Once a molecule is separated from the mixture, it can be isolated and quantified.

## **Different Chromatographic Techniques**

**Chromatography Stationary Phase** 

- 1. Thin Layer Chromatography
- 2. Paper Chromatography
- 3. Column Chromatography

Mobile Phase

- 1. Liquid chromatography
- 2. Gas Chromatography

Classification according to the force of separation

- 1- Adsorption chromatography.
- 2- Partition chromatography.
- 3- Ion exchange chromatography.
- 4- Gel filtration chromatography.
- 5- Affinity chromatography