

SEMI CONDUCTING MATERIALS!

Introduction: - sould out of a forther to A SemiConductor, is a Solid Which has the definite energy band. A bond gap of the Semi-Conductor is less than the band gap of an insulator.

* The resistivity of Semiconductor lies between Conducting and Insulating materials (10 2m to 0.52m)

* When the temperature is raised (or) Whom impunities are added, their Conductivity increases. (pa!).

* They have negative temperature co-efficient of resistance.

The Semi Conductors, both electrons and holes are charge carriers. Total conductivity (0) = 0e + 0h.

General properties of Semi Conducting Materials'-

- D they are formed by the Covalent bonds
- 2) They have an empty Conduction bond.
- 3) They have almost; filled Valence bound!
- 4) These materials have a Small energy gap.
- 5) They Posses Crystalline Structure.
 6) These materials have -ve temp Coefficient of resistance.
- 7) If impurities are added tit will increase the electrical Conductivity of Semiconductor.
- 8) If We increase the temperature of Semiconductor, its electrical Conductivity also increase,



Conduction in Somi Conductors:

In a pure Semiconductor, the number of free e is

Equal to the number of holes! Hence the botal current

Clensity (J). J = Je + Jh

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= qn Me E + 9 PMh E

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(maidwallers on J= 5 Endry ("0 = 2 (nme+PMh))

Where,

onere, 9 > charge of e (or) hole, n > Concentration of es,

P => Concentration of holes, E => applied Flectric field 5 ⇒ Conductivity of Semi Conductor, Me > mobility of es,

Mh -> mobility of holes.

Jov Since n = P = ni (intrinsic convier Concentration)

J=ni (Me+Mh) QE

Conductivity (0) = qni (µe+µh)

Hence, Conductivity of an intrinstic semiconductor depends On it's intrinstic Carrier Concentration and mobility of the electrons and holes.

1) Intrinsic Semi Conductor:

A Semi Corductor is an entremely pure from, without addition of impurities is known as an intrinsic semi-

