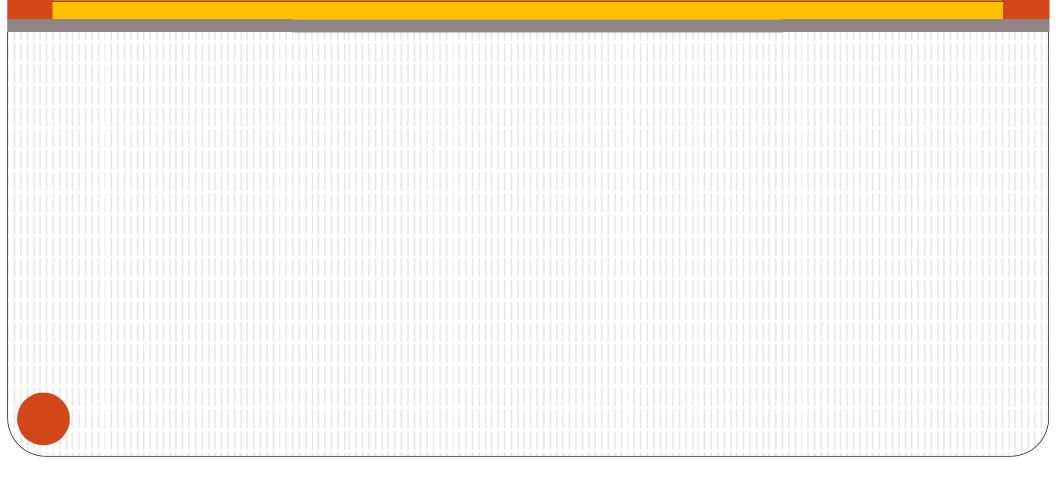




Unit-III Heat Treatment





Introduction







Introduction



- □ The properties of metals and alloys can be changed as desired by the heat treatment process
- Heat treatment is the controlled heating and cooling of metals for the purpose of altering their properties



Heat Treatment - Definition



It is an operation (or) combination of operations involving heating and cooling of metal/alloy in solid state to obtain desirable properties









Look at the chart and say the COLOR not the word

YELLOW BLUE ORANGE BLACK RED GREEN PURPLE YELLOW RED ORANGE GREEN BLACK BLUE RED PURPLE GREEN BLUE ORANGE

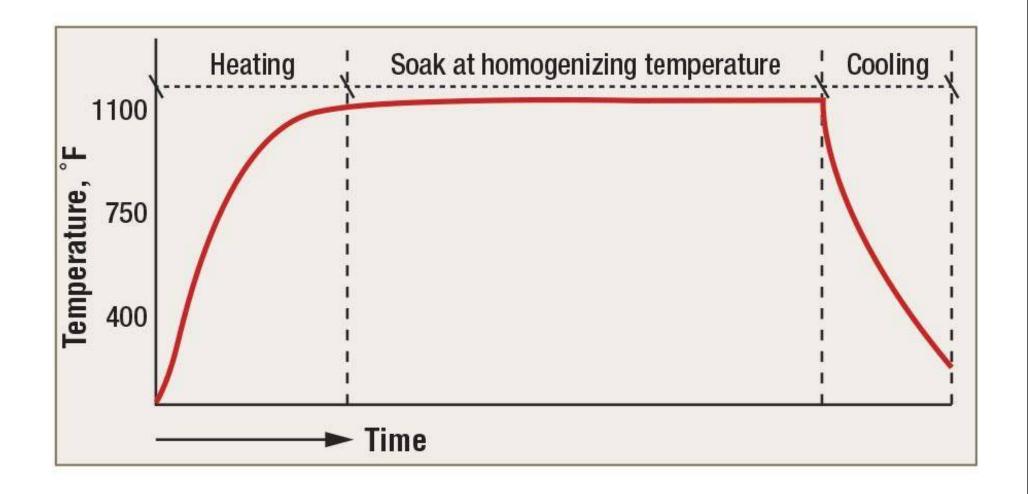
Left - Right Conflict

Your right brain tries to say the color but your left brain insists on reading the word

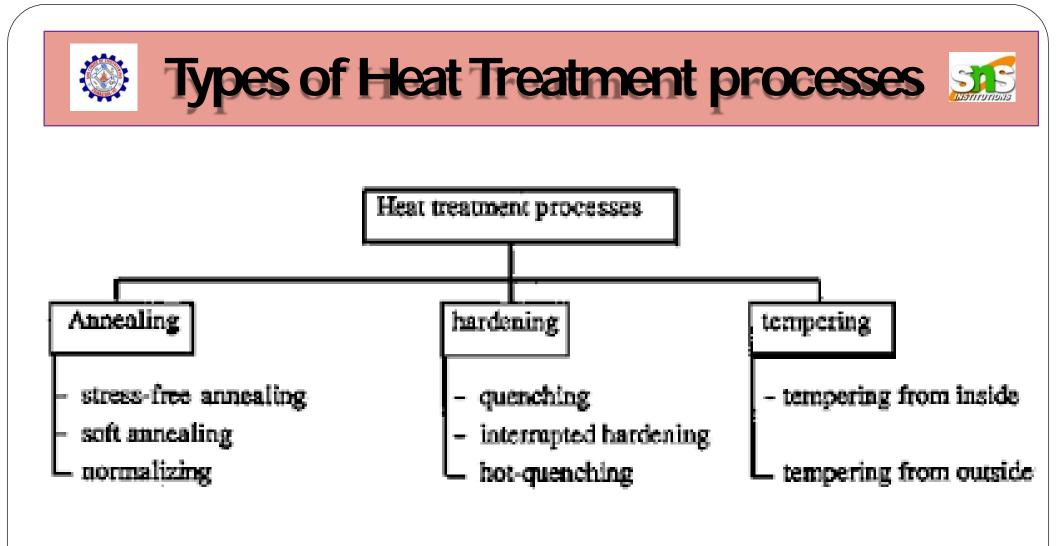
Mahendran S/Engg.Metallurgy/UNIT-II



Stages of Heat Treatment













- □ It is a heat treatment process in which a material is exposed to an elevated temperature for an extended time period and then slowly cooled
- □ It is one of the most widely used processes in the heat treatment of iron and steel



Types of Annealing

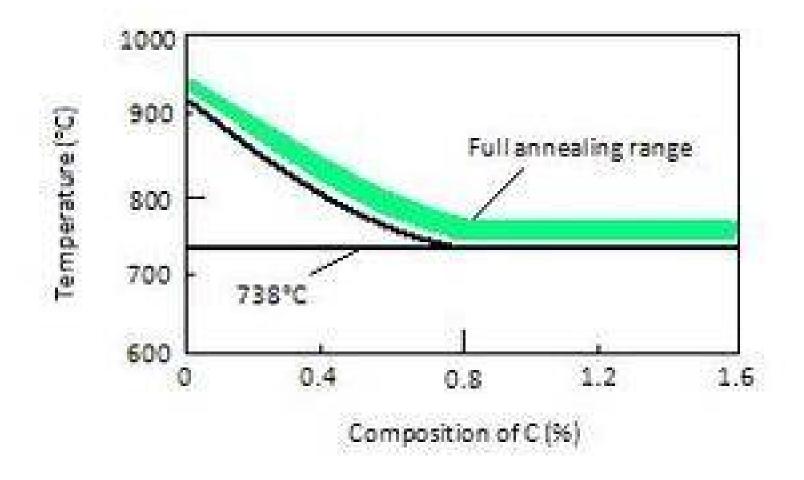


- 1) Full Annealing
- 2) Process Annealing
- 3) Stress relief Annealing
- 4) Recrystallization Annealing
- 5) Spheroidise Annealing















□ Actual definition of annealing describes only the full annealing

□ Full annealing consists of heating the steel to a temperature at or near the critical point, holding there for a time period and then allowing it to cool slowly in the furnace itself

□ Temp: 723°C to 910°C

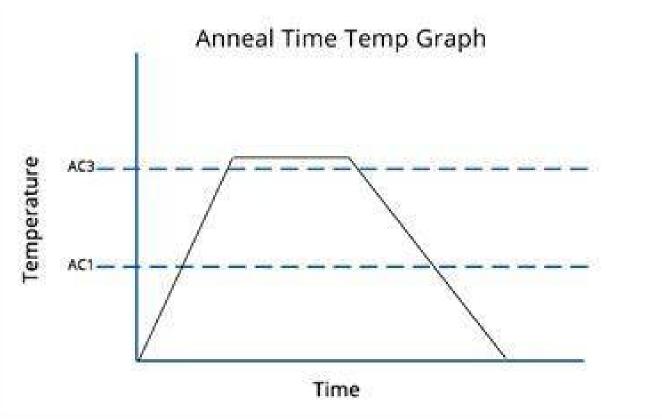
Objectives :

- □ To soften the metal
- □ To refine its crystalline structure
- □ To relieve the stress





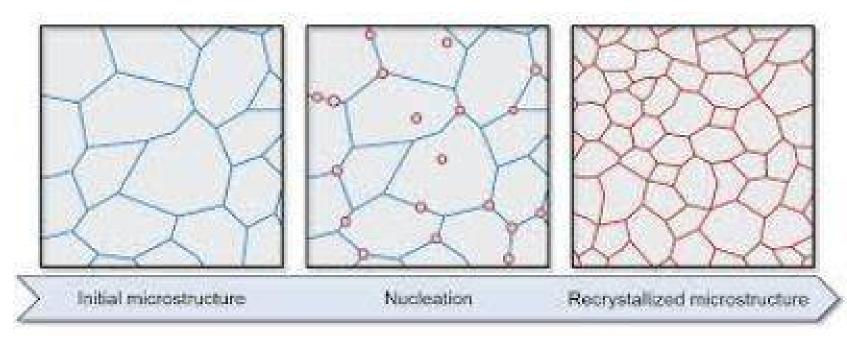
□ It's a heat treatment process that is employed to eliminate internal residual stresses induced by casting, quenching, machining, cold working, welding etc.







- Recrystallization: It is a process by which distorted grains of cold worked metal are replaced by new, strain free grains during heating above a specific minimum temperature.
- □ **Recrystallization temperature:** the temperature at which crystallization takes place







Medium and high carbon steels having a microstructure containing even coarse pearlite is too hard to conveniently machine or plastically deform.

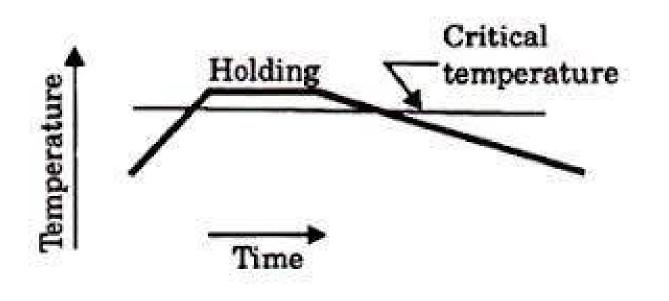


Fig. 5.20. Spheroidise annealing.



Normalising

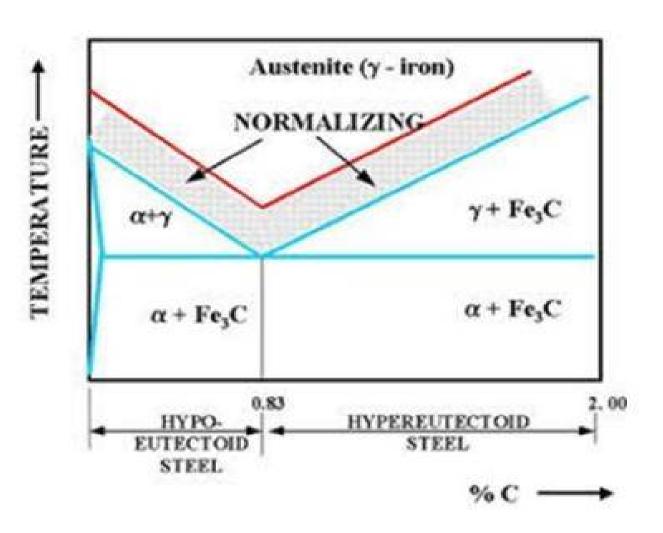


- □ Normalising is similar to full annealing, but cooling is established in air rather than in the furnace
- □ Full annealing is an expensive and time consuming process. when a reasonably soft and ductile material is required and cost savings are desired, then the normalizing process is used instead of full annealing
- □ Steel is heated to 50°C to 60°C above its upper critical temperature



Normalising



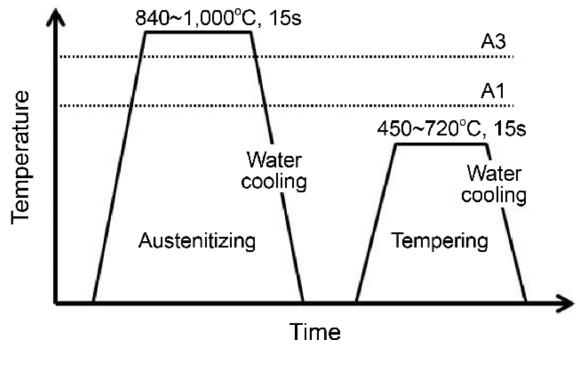




Quenching



- □ Its an accelerated cooling
- □ The cooling can be accomplished by contact with the a quenching medium which may be a gas , liquid or solid
- □ Most of the times liquid quenching media is widely used to achieve rapid cooling

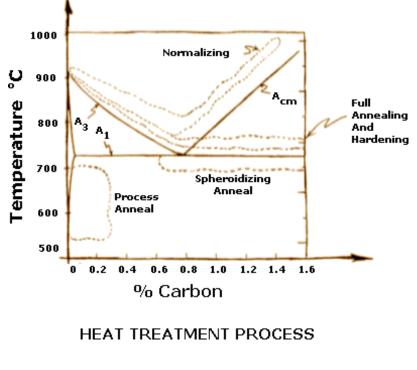


Hardening (By Quenching)



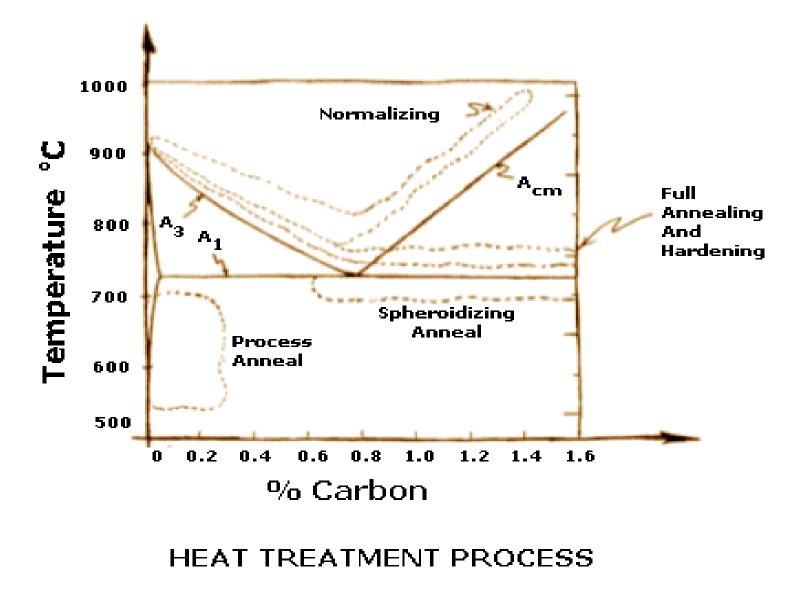
Operation:

- □ Heating: the steel to be heat treated is heated slowly in a furnace to 30°C to 50°C above the upper critical temperature
- Soaking: The heated steel is held at this temperature for considerable length of time





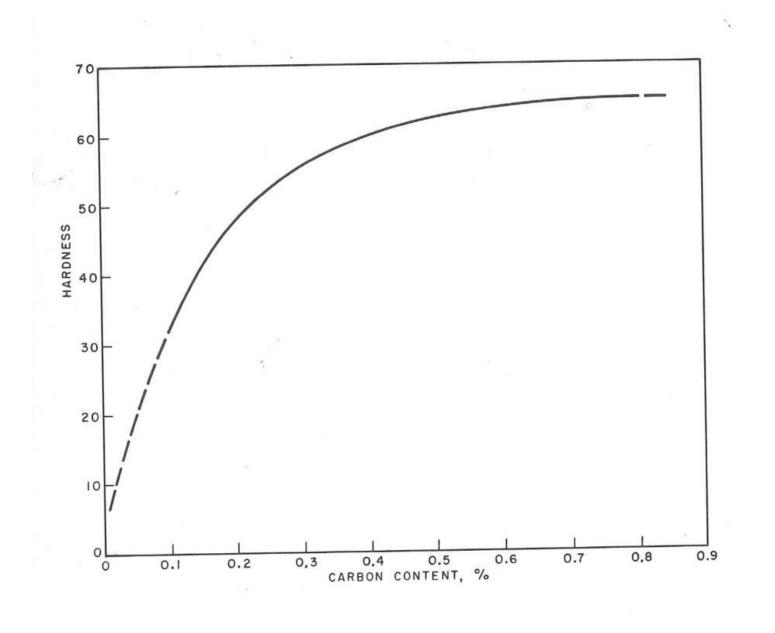
Hardening (By Quenching)





Factors affecting the Hardness



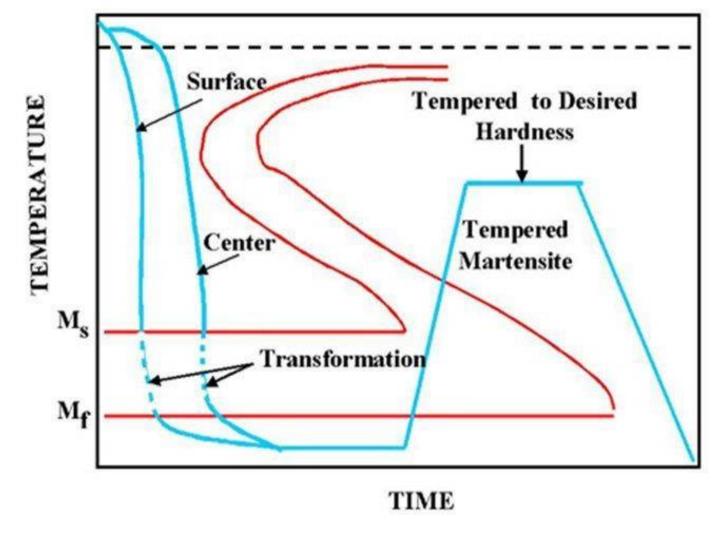




Tempering

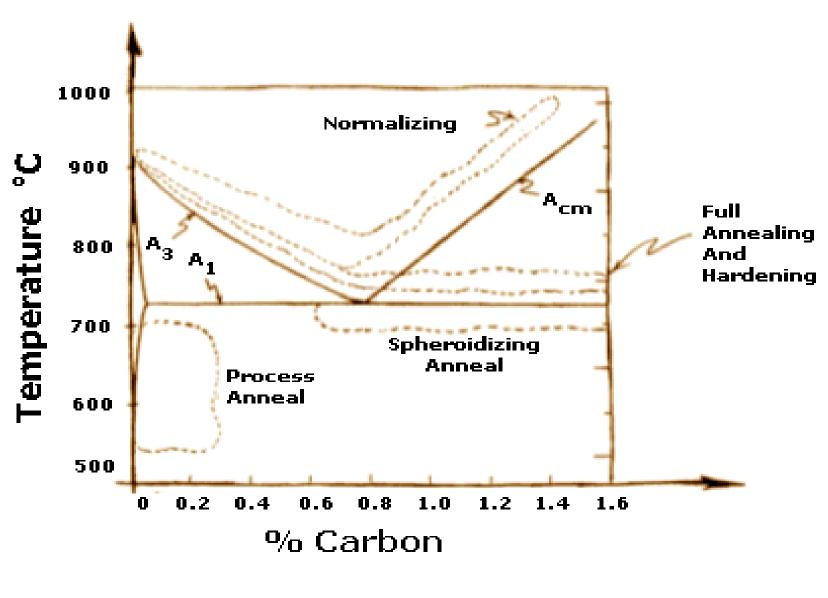


CONVENTIONAL QUENCHING AND TEMPERING









HEAT TREATMENT PROCESS



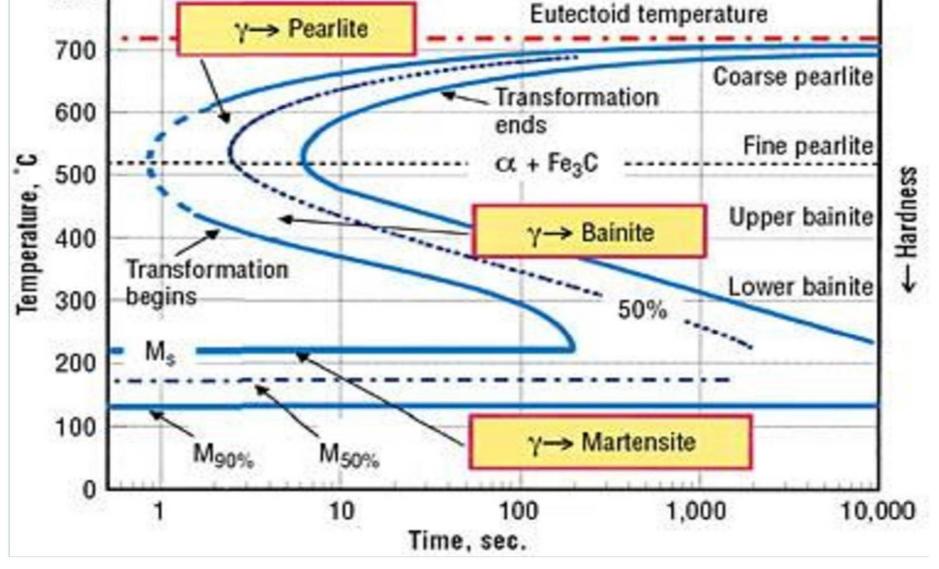
TTT Diagram



- Its also called as *Isothermal Transformation (IT) diagram*
- It's a tool used by heat treaters to predict quenching reactions in steels.
- Secause of the shape, it is also called S Curves or C Curves.
- Sometimes TTT diagram is also known as *Bain's Curve*.
- Temperature Vs Time diagram
- Temperature is constant during transformation
- Its is used to determine when transformations begin and end for an isothermal heat treatment of a previously <u>austenitized alloy</u>

Austenitizing heat treatment is heating a steel above the critical temperature, holding for a period of time long enough for transformation to occur.

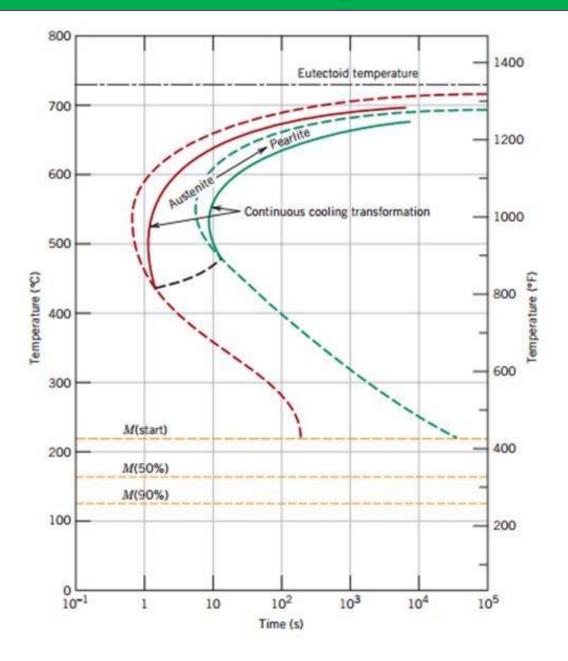
800 Eutectoid temp













Critical cooling rate



- The slowest rate of cooling of austenite that will result in 100% martensite transformation is known as critical cooling rate
- Critical rate is most important in hardening. In order to obtain 100% martensitic structure on hardening, the cooling must be much higher than the critical cooling rate

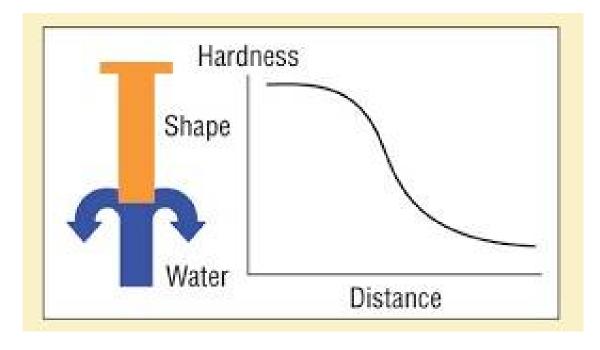


Hardenability



The property which determines the depth and distribution of hardness induced by quenching from the austenitic condition

Hardenability is a measure of ease of forming martensite





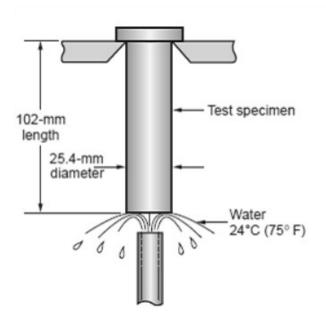
Specimen



* A bar of steel to be tested is machines to give a cylinder shape

- ✓ Long 102 mm (4 inch)
- ✓ Diameter with an upper lip -25.4 mm (1 inch)

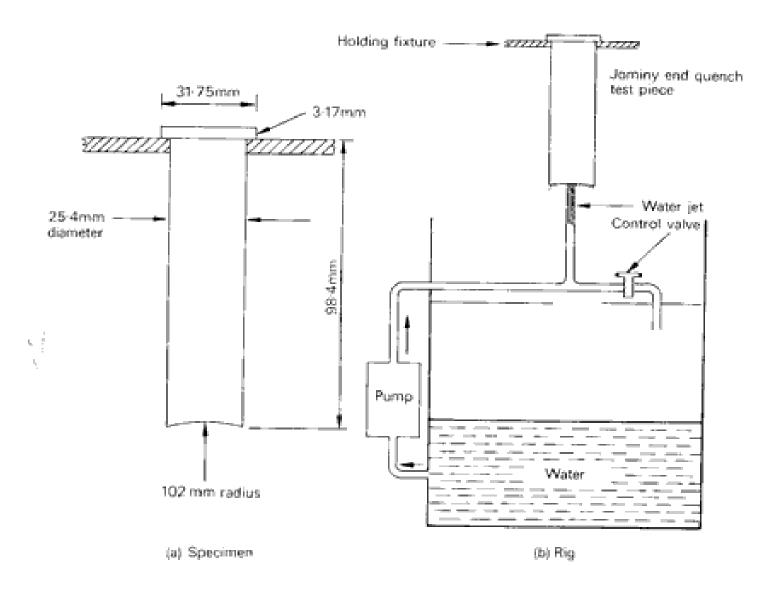








Jominy End Quench test



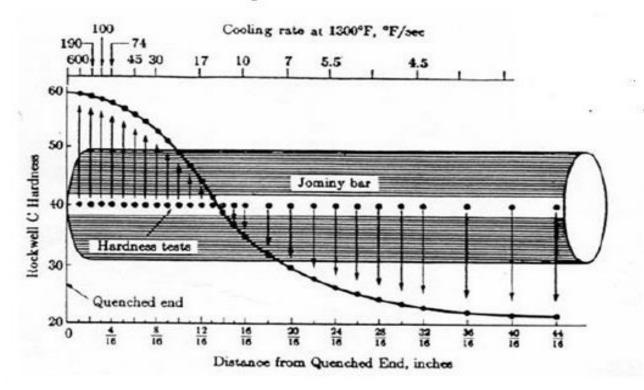
STE



Jominy End Quench test



Jominy Quench



Greatest hardness at the quenched end

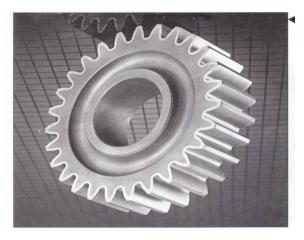


- It is desirable to that the surface of the components should have high hardness, while the inside (or) core should be soft. The treatment given to steels to achieve this are called surface heat treatments (or) Surface hardening
- Required soft and ductile for shock resistance
 Surface hardens for resisting wear and tear



Parts that are case hardened are

- Gears
- o Cams
- Bearing balls
- Universal joints













- Diffusion methods
 - ✓ Carburizing
 - ✓ Nitriding
 - ✓ Cyaniding
 - ✓ Carbonitriding
- Thermal Methods
 - ✓ Flame hardening
 - ✓ Induction Hardening

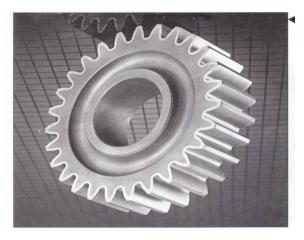


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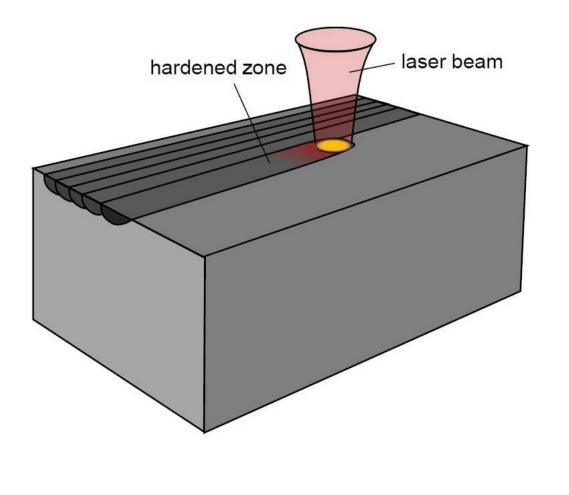




- Diffusion methods
 - ✓ Carburizing
 - ✓ Nitriding
 - ✓ Cyaniding
 - ✓ Carbonitriding
- Thermal Methods
 - ✓ Flame hardening
 - ✓ Induction Hardening

Diffusion Surface Hardening Treatments

Hardness of the surface is improved by diffusion interstitial elements like carbon, nitrogen or both into surface of steel components





Nitriding

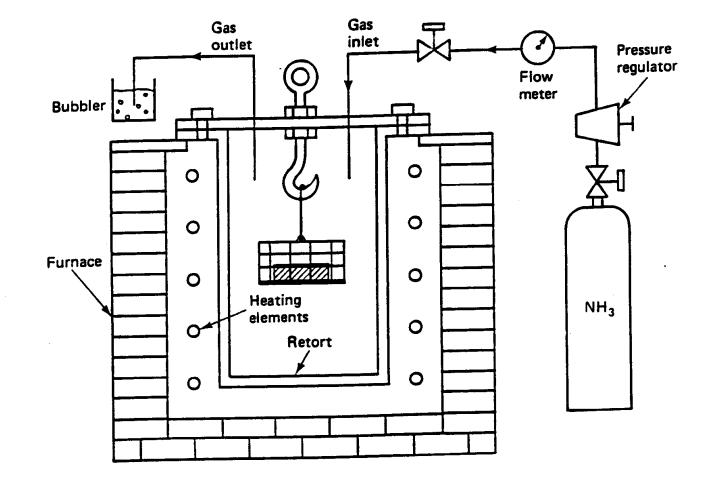


- Another process by which a case of hardened steel can be achieved
- In Nitriding, the steel piece is heated in a furnace between 500 600°C and at the same time is exposed to ammonia gas (NH3)
- The heat from the furnace causes the ammonia to decompose into hydrogen (H2) and nitrogen (N2)
- Nitrogen reacts with elements in the steel to form nitrides in the outer layer of the steel providing high hardness and wear resistance
- Nitriding times range between 1–100 hours depending on steel composition and depth of hardening desired



Nitriding





 $2NH_3 \rightarrow 2N + 3H_2$



Carbonitriding

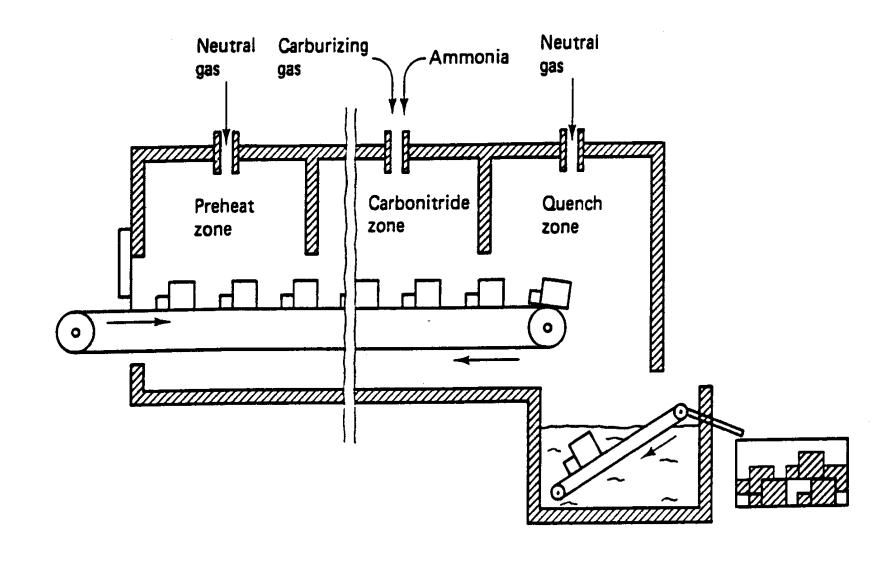


- This process involves both the diffusion of C and N into the steel surface
- Nitriding is performed in a gas atmosphere furnace using a carburizing gas such as propane or methane (source of C) mixed with several vol % of ammonia (NH3) (source of N)
- □ Carbonitriding is performed at temperatures above the UCT (700 800°C)
- Quenching is done in a gas which is not as severe as water quench (the result is less distortion on the material to be treated).



Carbonitriding





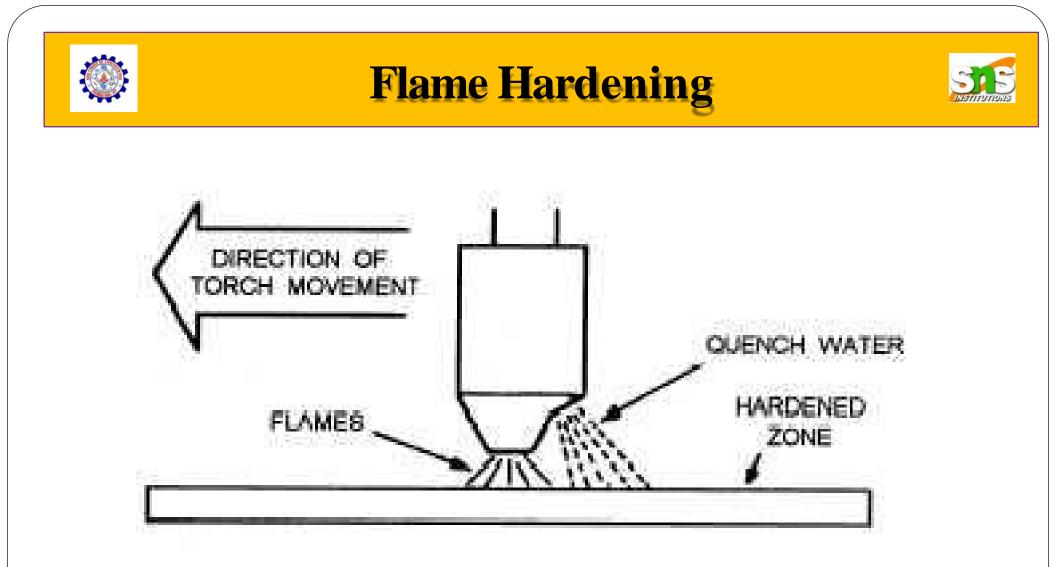




- □ Flame Hardening
- Induction hardening

Thermal Methods is applied because of one following reasons:

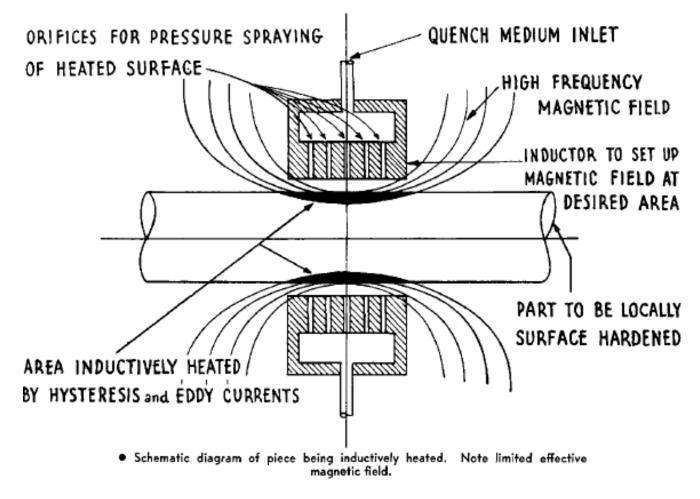
- 1. Only a small segment, section, or area of the part needs to be heat-treated. Typical examples are ends of valve stems and push rods, and the wearing surfaces of cams and levers
- 2. Better dimensional accuracy of a heat-treated part
- 3. Overall cost savings by using inexpensive steels to have the wear properties of alloyed steels.



Flame Hardening



Induction Hardening



Induction Hardening

