SNS COLLEGE OF ALLIED HEALTH SCIENCE

Affiliated to The Tamil Nadu Dr M.G.R Medical University, Chennai



DEPARTMENT OF PHYSICIAN ASSISTANT

COURSE NAME: BIOCHEMISTRY

UNIT: 1

UNII . I

TOPIC: PROTEINS – FUNCTIONS AND PROPERTIES

FACULTY NAME: MITHRA V



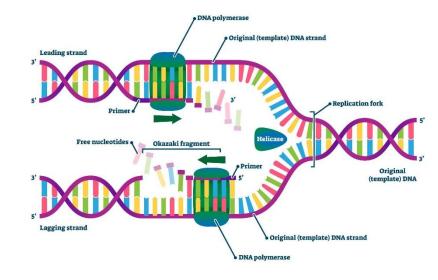
Hormonal Proteins

- Secreted by endocrine glands as hormones.
- Regulate body functions
 (e.g., insulin controls blood sugar)

Enzymatic Proteins

- Act as catalysts for chemical reactions.
- Speed up metabolism
- Help in DNA replication
- Amylase: breakdown starches into sugars.
- **Pepsin:** breaks down proteins into smaller polypeptides.
- **DNA polymerase**: Synthesizes DNA

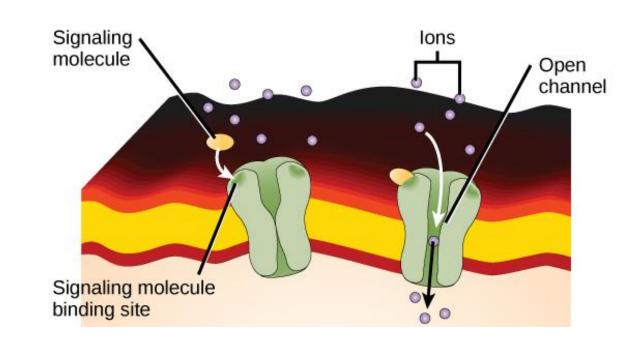
DNA POLYMERASE





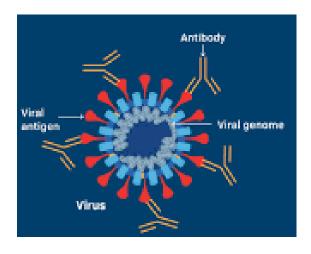
Structural Proteins

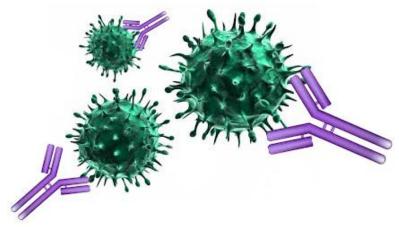
- Fibrous proteins provide strength and shape.
- · Components of muscles, bones, skin, hair, cartilage.
- Excess protein can be converted to fat for storage.
- Receptor Proteins
- Present on cell surfaces; control what substances enter or leave the cell.
- Enable cells to detect and respond to signals (hormones, neurotransmitters, etc.).





- Defensive Proteins (Antibodies/Immunoglobulins)
- Produced by white blood cells to fight pathogens.
- Identify, neutralize, destroy antigens.
- Part of immune system for healing and protection.
- Storage Proteins
- Store minerals and nutrients (e.g., casein in milk, ovalbumin in egg whites).
- Provide amino acids or ions (like iron in ferritin, potassium, etc.) when needed.

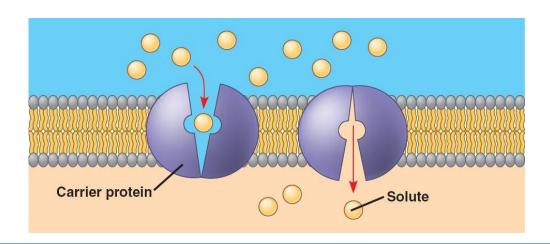






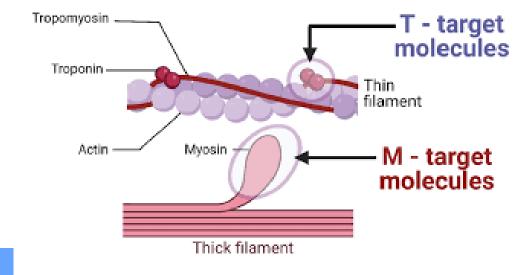
• Transport Proteins

- Carry vital substances throughout the body
 (e.g., hemoglobin carries O2)
- Move molecules across cell membranes or in blood.



Contractile Proteins

- Responsible for movement (e.g., actin and myosin in muscles).
- Regulate force and speed of muscle and heart contractions.



PHYSICAL PROPERTIES

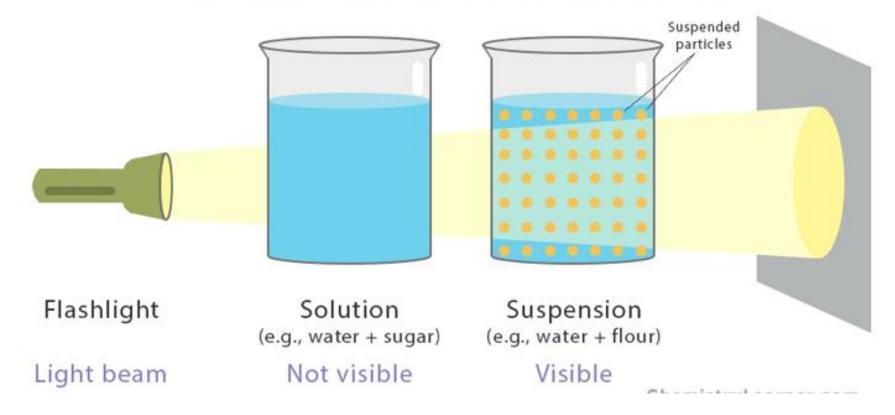


Property	Description	Example
MOLECULAR WEIGHT	Ranges from ~5 kDa (insulin) to million Da	Hemoglobin: ~64.5 kDa (tetramer)
SHAPE/CONFORMATION	Globular (spherical), fibrous (elongated, thread like)	Globular: Myoglobin Fibrous: Collagen and Keratin
SOLUBILITY	Depends on surface hydrophilicity/hydrophobicity	Albumin and Myosin
COLLOIDAL NATURE	Form colloids in solution → high viscosity, light scattering (Tyndall effect)	Egg white (albumin solution) becomes viscous when beaten
OPTICAL ACTIVITY	Rotate plane-polarized light	All natural proteins are laevorotatory (L-amino acids)





Scattering of light by particles in a medium



CHEMICAL PROPERTIES



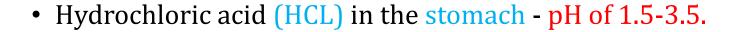
Property	Description	Example
ZWITTERION	Proteins contain both acidic (-COOH) and basic (-NH ₂ , side chains)	At physiological pH ~7.4, most proteins are negatively charged
	Hybrid molecule, +vely & -vely charged ionic groups.	Glycine at pH = 6.0 is zwitterionic
ISOELECTRIC PH (PI)	Isoelectric point (pI) – The pH at which an amino acid or protein has zero net charge, so it doesn't move in an electric field to anode or cathode.	 Pepsin (highly acidic): pI ≈ 1.0 Lysozyme (basic): pI ≈ 11 Serum albumin: pI ≈ 4.7 →
PEPTIDE BONDS	-CO-NH- linkage; partial double-bond character → planar, rigid	Gives proteins their secondary structure backbone

BIOLOGICAL PROPERTIES



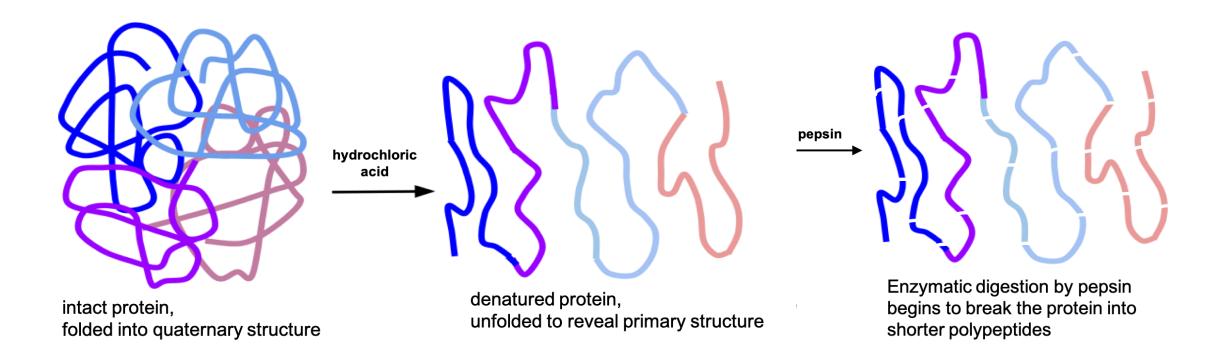
Denaturation of Proteins

Agent/Type	Mechanism	Example
НЕАТ	Breaks H-bonds, hydrophobic interactions	Boiling egg → egg white albumin coagulates (solid)
STRONG ACID/BASE	Disrupts ionic bonds and H-bonds	Curdling of milk by lactic acid
ORGANIC SOLVENTS	Disrupts hydrophobic core	Alcohol disinfection → denatures microbial proteins
MECHANICAL AGITATION	Shearing forces break weak interactions	Whipping egg white → temporary foam





 Acidity of the stomach causes food proteins to denature, unfolding their threedimensional structure to reveal just the polypeptide chain.

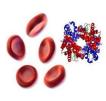


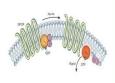


Structural Proteins



Transport Proteins





Regulatory Proteins

Storage Proteins





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Defensive Proteins





REFERENCES



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THANK YOU