

**SNS COLLEGE OF ALLIED HEALTH SCIENCE**

Affiliated to The Tamil Nadu Dr M.G.R Medical University, Chennai

**DEPARTMENT OF CARDIO PULMONARY PERFUSION CARE**

**TECHNOLOGY**

**COURSE NAME: BIOCHEMISTRY**

**UNIT : 1**

**TOPIC : PROTEINS – FUNCTIONS AND PROPERTIES**

**FACULTY NAME: MITHRA V**

# FUNCTIONS OF PROTEINS

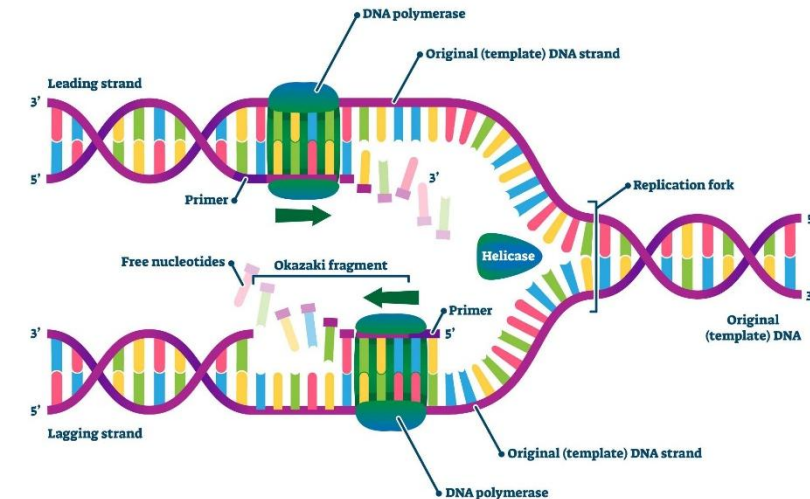
- **Hormonal Proteins**

- Secreted by **endocrine glands** as hormones.
- Regulate body functions (e.g., insulin controls blood sugar)

- **Enzymatic Proteins**

- Act as **catalysts** for chemical reactions.
- Speed up metabolism
- Help in **DNA replication**
- **Amylase:** breakdown starches into sugars.
- **Pepsin:** breaks down proteins into smaller polypeptides.
- **DNA polymerase:** Synthesizes DNA

## DNA POLYMERASE



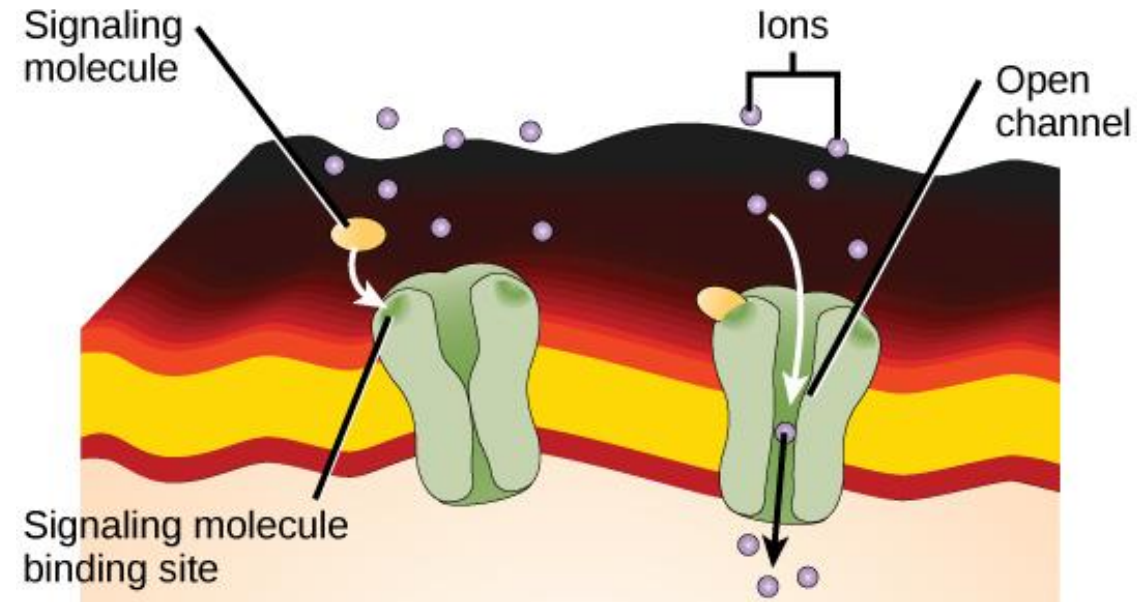
# FUNCTIONS OF PROTEINS

- **Structural Proteins**

- Fibrous proteins - provide **strength and shape**.
- Components of muscles, bones, skin, hair, cartilage.
- Excess protein can be converted to fat for storage.

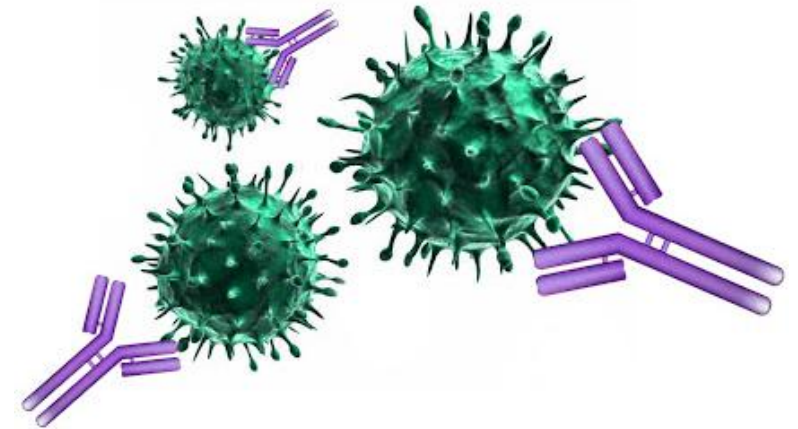
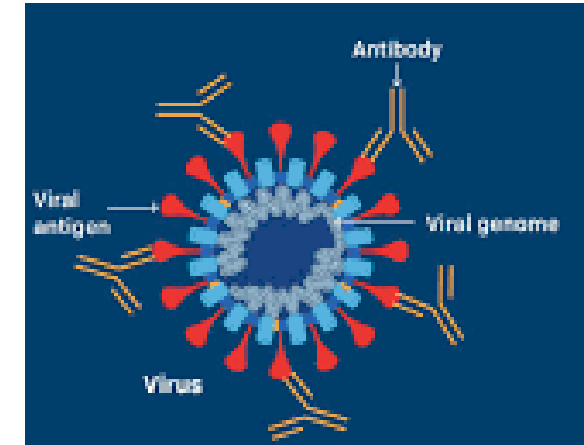
- **Receptor Proteins**

- Present on cell surfaces; control what substances **enter or leave the cell**.
- Enable cells to detect and respond to signals (**hormones, neurotransmitters**, etc.).



# FUNCTIONS OF PROTEINS

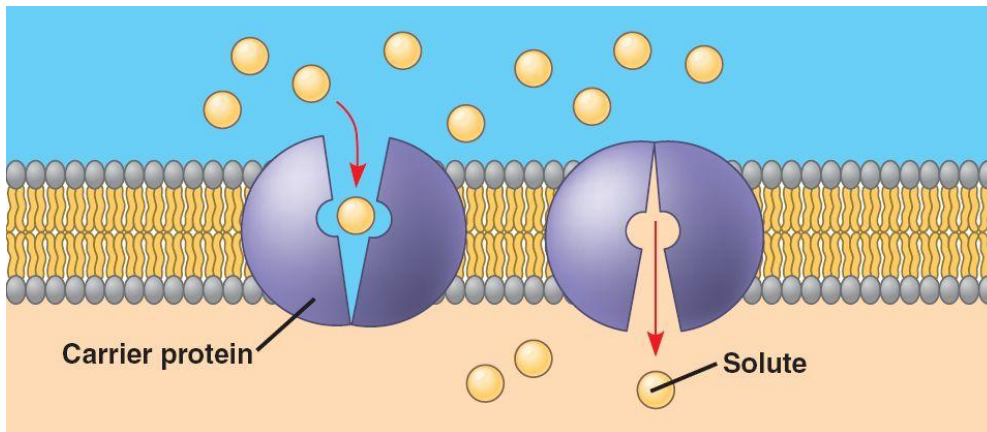
- **Defensive Proteins (Antibodies/Immunoglobulins)**
  - Produced by **white blood** cells to **fight pathogens**.
  - Identify, neutralize, **destroy antigens**.
  - Part of immune system for healing and protection.
- **Storage Proteins**
  - Store minerals and nutrients (e.g., **casein** in milk, **ovalbumin** in egg whites).
  - Provide amino acids or ions (like iron in ferritin, potassium, etc.) when needed.



# FUNCTIONS OF PROTEINS

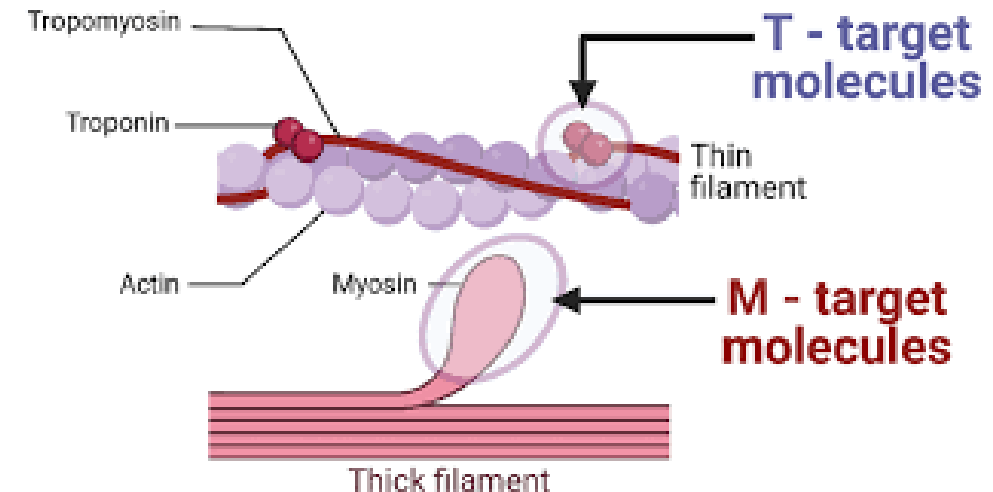
## • Transport Proteins

- Carry vital substances throughout the body (e.g., hemoglobin carries O<sub>2</sub>)
- Move molecules across cell membranes or in blood.



## • Contractile Proteins

- Responsible for movement (e.g., actin and myosin in muscles).
- Regulate force and speed of muscle and heart contractions.

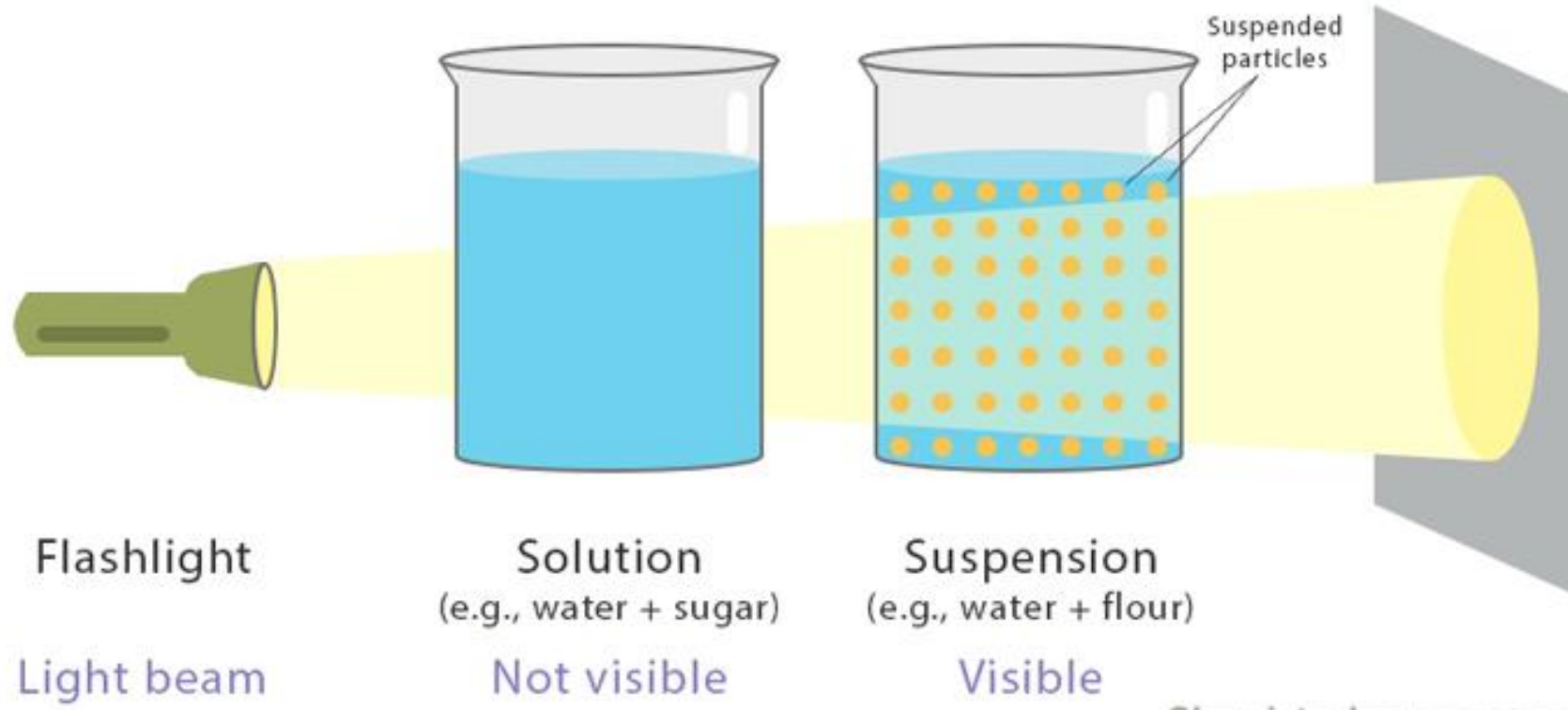


# PHYSICAL PROPERTIES

Property	Description	Example
<b>MOLECULAR WEIGHT</b>	Ranges from ~5 kDa (insulin) to million Da	Hemoglobin: ~64.5 kDa (tetramer)
<b>SHAPE/CONFORMATION</b>	Globular (spherical), fibrous (elongated, thread like)	Globular: Myoglobin Fibrous: Collagen and Keratin
<b>SOLUBILITY</b>	Depends on surface hydrophilicity/hydrophobicity	Albumin and Myosin
<b>COLLOIDAL NATURE</b>	Form colloids in solution → high viscosity, light scattering (Tyndall effect)	Egg white (albumin solution) becomes viscous when beaten
<b>OPTICAL ACTIVITY</b>	Rotate plane-polarized light	All natural proteins are laevorotatory (L-amino acids)

## Tyndall Effect

Scattering of light by particles in a medium





# CHEMICAL PROPERTIES

Property	Description	Example
<b>ZWITTERION</b>	<p>Proteins contain <b>both acidic</b> (<math>-\text{COOH}</math>) and <b>basic</b> (<math>-\text{NH}_2</math>, side chains)</p> <p>Hybrid molecule, <b>+vely &amp; -vely</b> charged <i>ionic groups</i>.</p>	<p>At physiological pH <math>\sim 7.4</math>, most proteins are negatively charged</p> <p>Glycine at pH = 6.0 is zwitterionic</p>
<b>ISOELECTRIC PH (PI)</b>	<p>Isoelectric point (pI) – The pH at which an amino acid or <b>protein has zero net charge</b>, so it doesn't move in an electric field to anode or cathode.</p>	<ul style="list-style-type: none"> <li>• Pepsin (highly acidic): pI <math>\approx 1.0</math></li> <li>• Lysozyme (basic): pI <math>\approx 11</math></li> <li>• Serum albumin: pI <math>\approx 4.7 \rightarrow</math></li> </ul>
<b>PEPTIDE BONDS</b>	<p><b><math>-\text{CO}-\text{NH}-</math> linkage</b>; partial double-bond character <math>\rightarrow</math> planar, rigid</p>	<p>Gives proteins their secondary structure backbone</p>

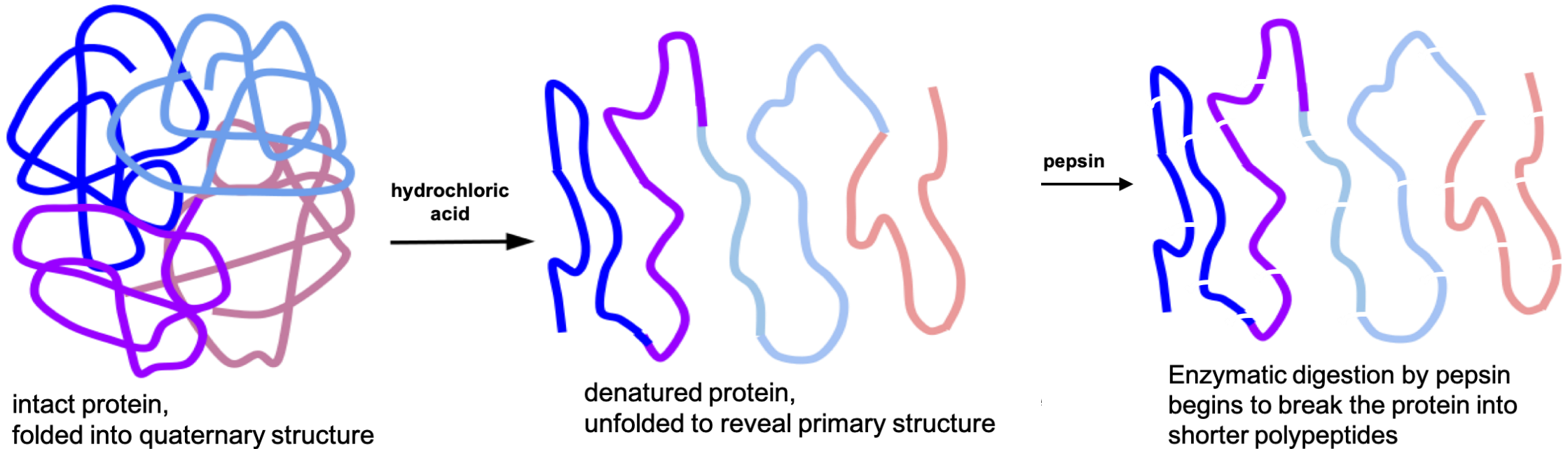


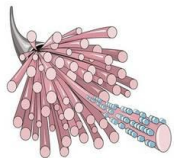
# BIOLOGICAL PROPERTIES

## Denaturation of Proteins

Agent/Type	Mechanism	Example
HEAT	Breaks H-bonds, hydrophobic interactions	Boiling egg → egg white albumin coagulates (solid)
STRONG ACID/BASE	Disrupts ionic bonds and H-bonds	Curdling of milk by lactic acid
ORGANIC SOLVENTS	Disrupts hydrophobic core	Alcohol disinfection → denatures microbial proteins
MECHANICAL AGITATION	Shearing forces break weak interactions	Whipping egg white → temporary foam

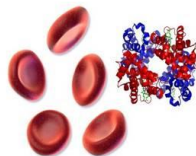
- Hydrochloric acid (HCL) in the stomach - pH of 1.5-3.5.
- **Acidity of the stomach causes food proteins to denature, unfolding their three-dimensional structure to reveal just the polypeptide chain.**



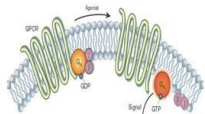


## Structural Proteins

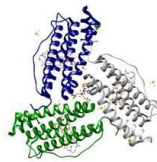
## Transport Proteins



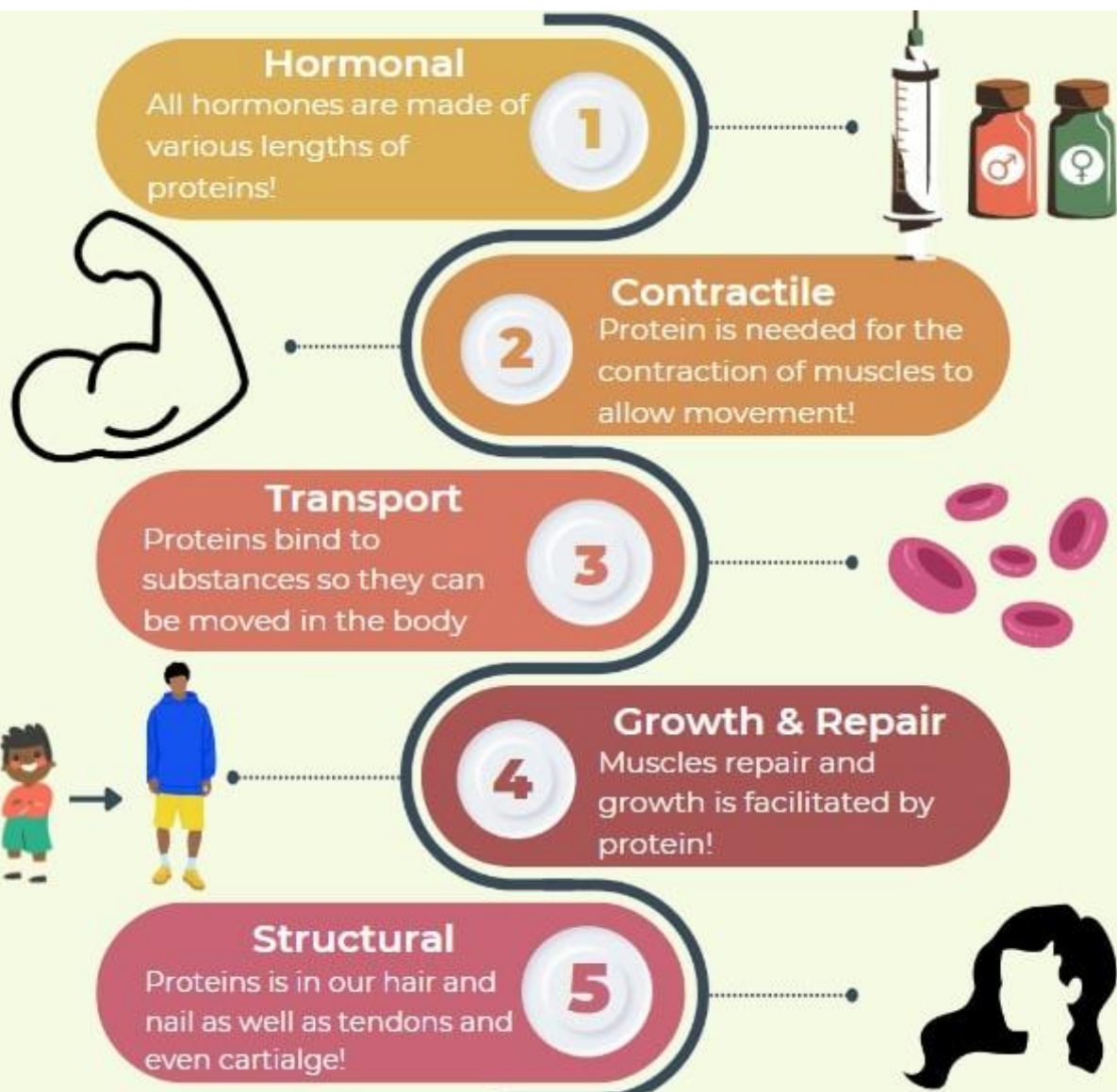
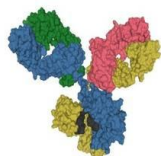
## Regulatory Proteins



## Storage Proteins



## Defensive Proteins



# REFERENCES

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# THANK YOU