

SNS COLLEGE OF ALLIED HEALTH SCIENCES

SNS Kalvi Nagar, Coimbatore - 35 Affiliated to Dr MGR Medical University, Chennai

DEPARTMENT : OPERATION THEATRE AND ANAESTHESIA TECHNOLOGY

COURSE NAME : PHARMACOLOGY

UNIT : INHALATIONAL GASES

TOPICS : AGENTS - ETHER, HALOTHANE, ISOFLURANE, SEVOFLURANE, DESFLURANE





INHALATIONAL GASES



- Inhalational gases, also known as inhalation anesthetics or volatile anesthetics, are substances that are administered through inhalation to induce and maintain general anesthesia during surgical or medical procedures.
- These gases are designed to cause a reversible loss of consciousness and sensation, allowing patients to undergo procedures without experiencing pain or awareness.



ETHER (Diethyl Ether)



Chemical Structure: Diethyl ether (C4H10O).

Mechanism of Action:

Ether induces anesthesia by depressing the central nervous system. It enhances the inhibitory neurotransmitter gamma-aminobutyric acid (GABA) and reduces neuronal excitability.





Administered via inhalation. Ether has a low blood-gas solubility, allowing for rapid induction and recovery from anesthesia.

Clinical Use:

Historically used for general anesthesia. However, its use has declined due to flammability, airway irritation, and the availability of safer alternatives.



HALOTHANE



Chemical Structure:

2-bromo-2-chloro-1,1,1-trifluoroethane.

Mechanism of Action:

Halothane enhances GABAergic neurotransmission and inhibits excitatory neurotransmission. It also has myocardial depressant effects.





Administered by inhalation. Halothane has a moderate bloodgas solubility, resulting in a slightly longer induction and recovery compared to newer agents.

Clinical Use:

Once widely used but has been largely replaced due to concerns about hepatotoxicity, especially in susceptible individuals.



ISOFLURANE



Chemical Structure:

1-chloro-2,2,2-trifluoroethyl difluoromethyl ether.

Mechanism of Action:

Isoflurane, like other inhalation anesthetics, enhances GABAergic neurotransmission and inhibits excitatory neurotransmission. It has less myocardial depressant effects compared to halothane.





Administered by inhalation. Isoflurane has a low blood-gas solubility, leading to rapid induction and recovery.

Clinical Use:

Commonly used for the maintenance of anesthesia during surgical procedures.



SEVOFLURANE



Chemical Structure:

1,1,1,3,3,3-hexafluoro-2-(fluoromethoxy)propane.

Mechanism of Action:

Similar to other inhalation anesthetics, it enhances GABAergic neurotransmission and inhibits excitatory neurotransmission. Sevoflurane has a low blood-gas solubility.





Administered by inhalation. Sevoflurane has a pleasant odor and is often used for induction and maintenance of anesthesia, particularly in pediatric patients.

Clinical Use:

Widely used in various surgical procedures.



DESFLURANE



Chemical Structure:

1,1,1,2,2,2-hexafluoro-2-(fluoromethoxy)ethane.

Mechanism of Action:

Similar to other inhalation anesthetics, it enhances GABAergic neurotransmission and inhibits excitatory neurotransmission. Desflurane has the lowest blood-gas solubility among commonly used inhalation agents.





Administered by inhalation. Characterized by rapid induction and emergence due to its low blood-gas solubility.

Clinical Use:

Mainly used for the maintenance of anesthesia, especially in fast-track surgery, where quick recovery is desired.



TECHNICIAN ROLE



Patient Monitoring:

- Continuous monitoring of the patient's respiratory rate, heart rate, blood pressure, and oxygen saturation.
- Anesthesia providers adjust the concentrations of agents based on individual patient needs and procedure requirements.



ASSESSMENT



- What is the Pharmacokinetics of Ether ?
- What all are the Clinical use of Desflurane ?