

SNS COLLEGE OF TECHNOLOGY



Vazhiamyampalayam, Coimbatore-35

(An Autonomous institution)

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DEPARTMENT OF CHEMISTRY

COURSE NAME: 19CHB102- ENGINEERING CHEMISTRY FOR ELECTRICAL SCIENCES

I YEAR / II SEMESTER

UNIT: 3. NANOCHEMISTRY

TOPIC: 2. SOL GEL METHOD







- Bottom up method
- Extended composition range
- Better homogeneity
- Less energy consumption
- Economical method



Sol

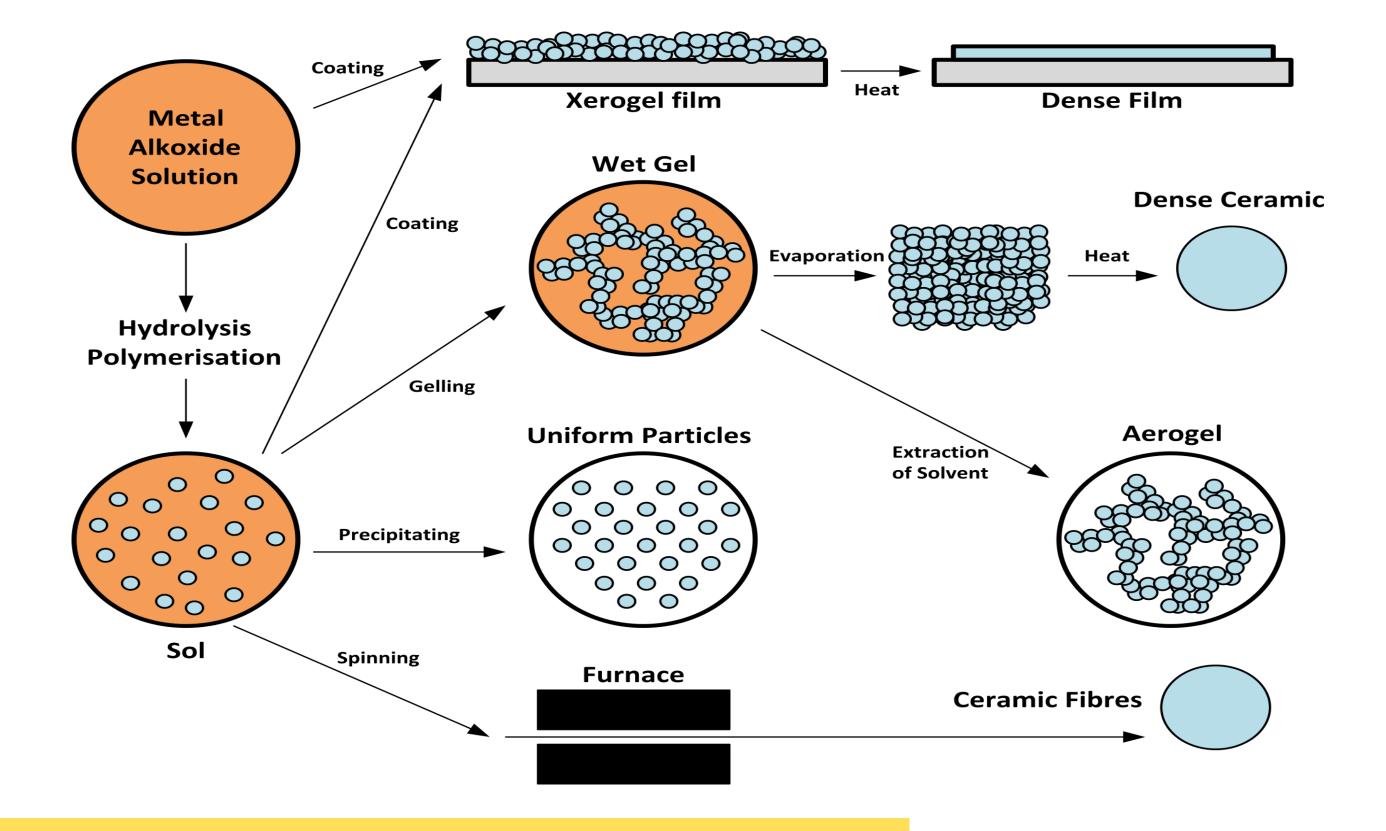


Gel



SCHEMATIC REPRESENTATION OF PROCESS



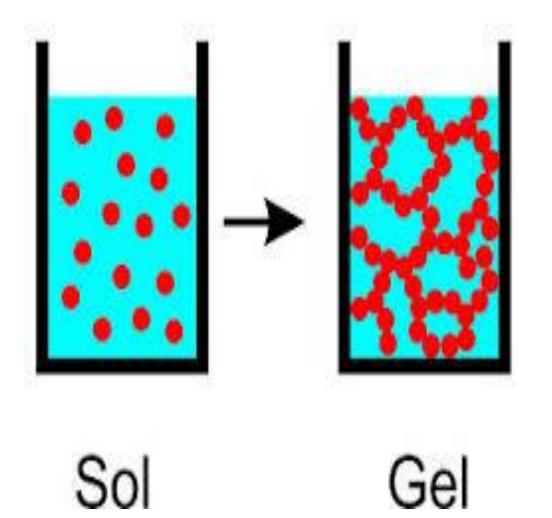




PROCESS



- Sol-gel is a chemical solution process used to make ceramic and glass materials in the form of thin films, fibers or powders.
- A sol is (a colloidal or molecular suspension) obtained from (starting materials).
- A gel is a semi-rigid mass that forms when the solvent from the sol begins to evaporate and the particles or ions left behind begin to join together in a continuous network

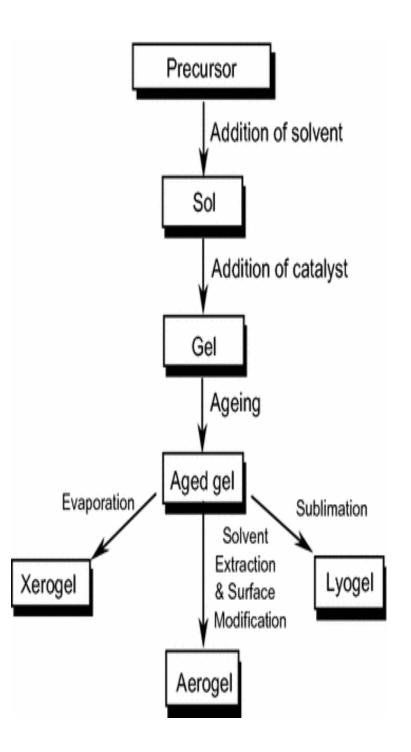








- The sol-gel process is a wet-chemical technique that uses either a chemical solution (sol short for solution) or colloidal particles (sol for nanoscale particle) to produce an integrated network (gel).
- Metal alkoxides and metal chlorides are typical precursors. They undergo hydrolysis and polycondensation reactions to form a colloid, a system composed of nanoparticles dispersed in a solvent. The sol evolves then towards the formation of an inorganic continuous network containing a liquid phase (gel)





PROCESS



- Formation of a metal oxide involves connecting the metal centers with oxo (M-O-M) or hydroxo (M-OH-M) bridges, therefore generating **metal-oxo or metal-hydroxo polymers** in solution.
- After a drying process, the liquid phase is removed from the gel. Then, a thermal treatment (calcination) may be performed in order to favor further poly condensation and enhance mechanical properties







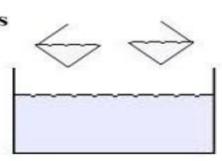








Mix reactives

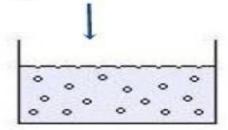


Hydrolysis

$$Si - OR + HOH \longrightarrow Si - OH + ROH$$

Hydrolysis and Condesation reactions take place





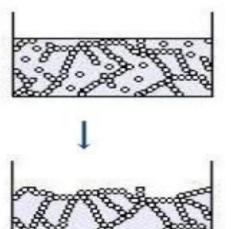
Condensation

$$Si - OH + HO - Si - O - Si + H2O$$

$$Si - OR + HO - Si \longrightarrow Si - O - Si + ROH$$

Gelification

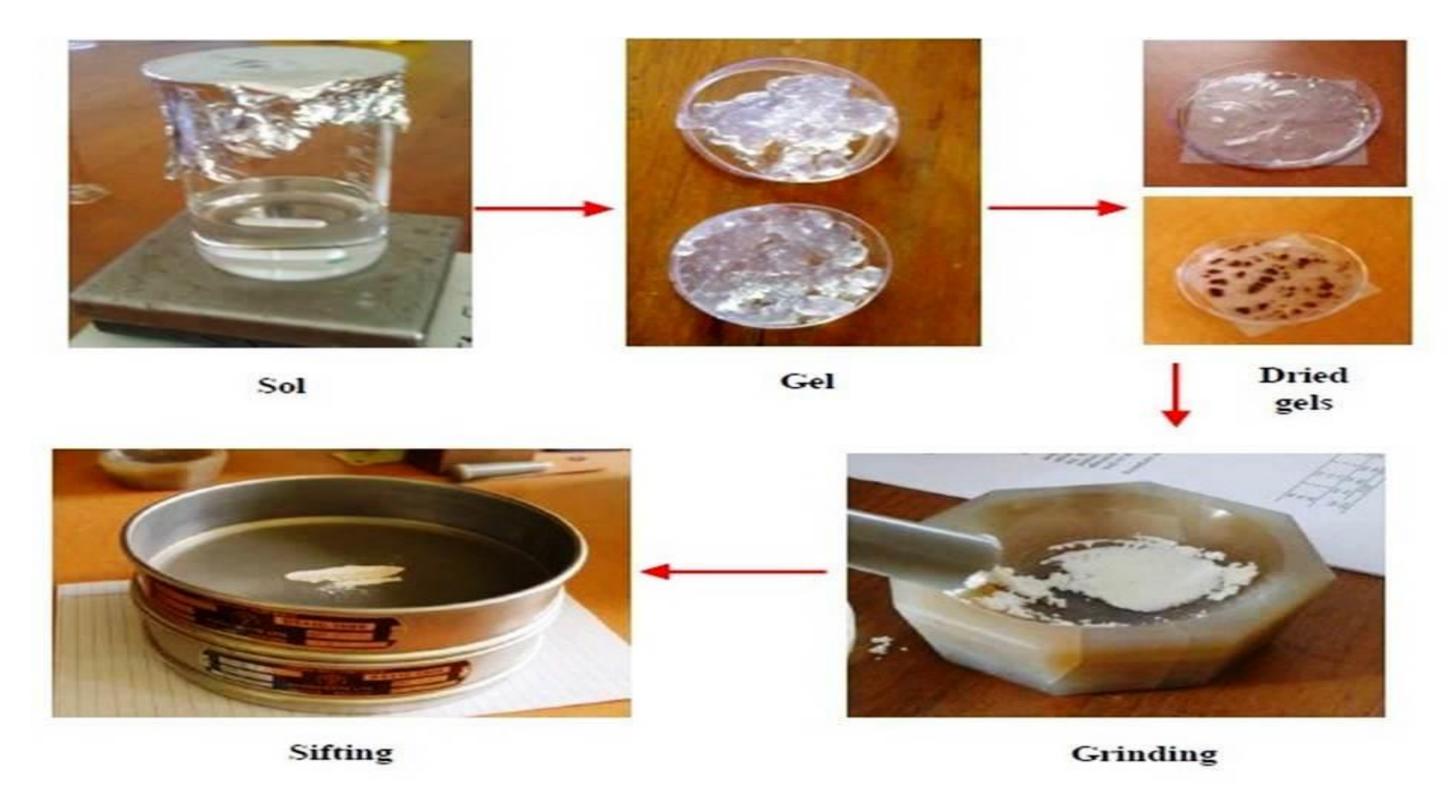






PICTORIAL REPRESENTATION OF PROCESS

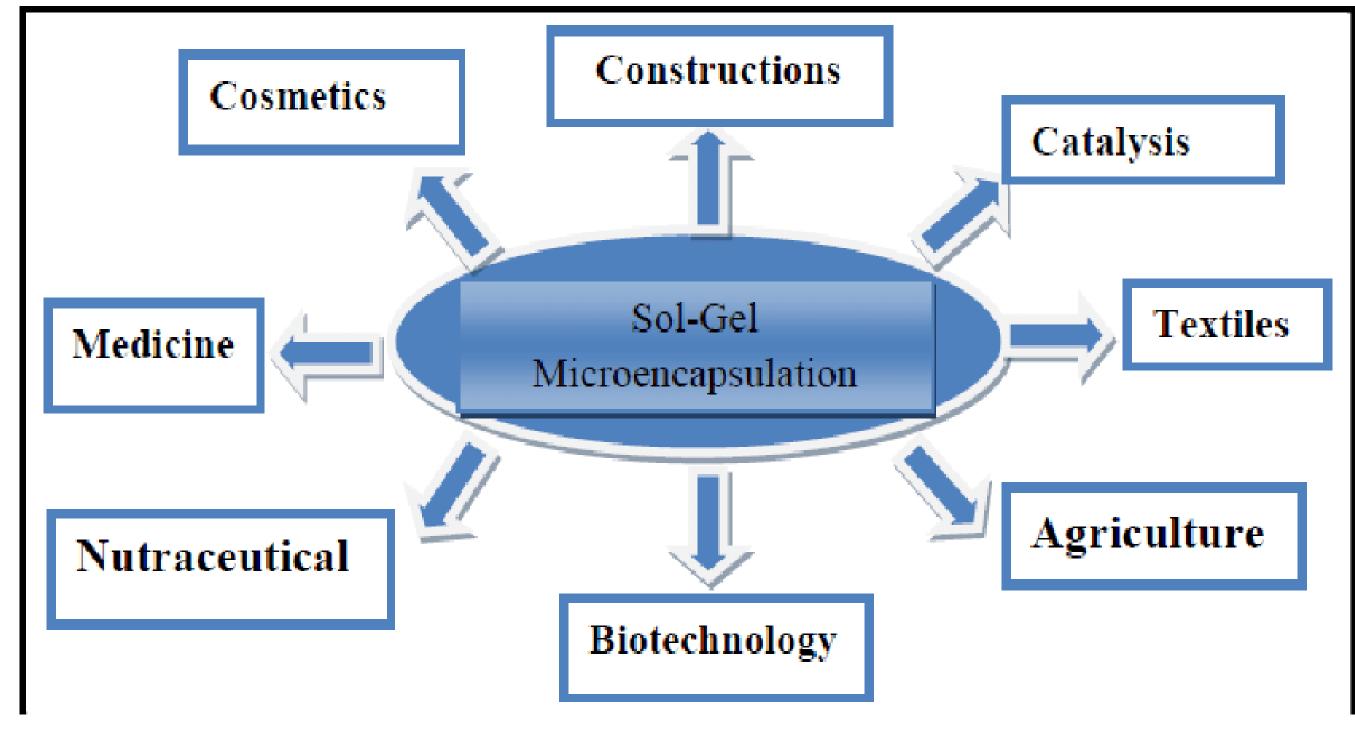






APPLICATIONS OF SOL GEL METHOD







ASSESSMENT



1. List out the various stages of gel formation

2. List out any two nanoparticles prepared by sol gel method





SUMMARY



REFERENCES



- 1. Dr.V. Veeraiyan, "Engineering Chemistry-II" VRB Pub. Co. Ltd, Chennai. 2016...
- 2. Wiley, "Engineering Chemistry", John Wiley & Sons. InC, USA.
- 3. P.C.Jain & Monicka Jain, "Engineering Chemistry", Dhanapat Rai Publising Company Pvt. Ltd. 2017.

